

**ROCHESTER INSTITUTE OF TECHNOLOGY  
MICROELECTRONIC ENGINEERING**

***Microelectromechanical Systems (MEMs)  
Materials for MEMs***

***Dr. Lynn Fuller***

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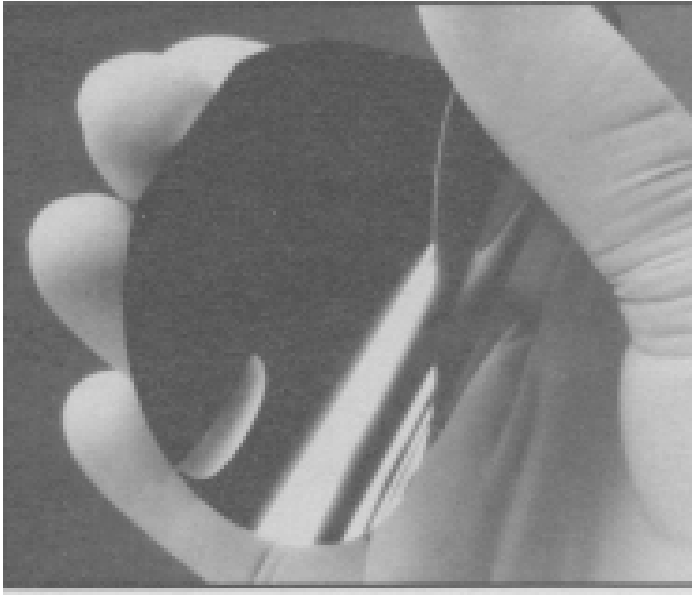
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***OUTLINE***

**Special Single Crystal Silicon Wafers**  
**Quartz Wafers**  
**Mechanical Properties**  
**Thermal Properties**  
**Electrical Properties**  
**Optical Properties**  
**Glass**  
**Ceramics**  
**Organic Materials**  
**Solder**  
**Wax**  
**Epoxy**

## *SPECIAL SINGLE CRYSTAL SILICON WAFERS*



Thickness = 10  $\mu\text{m}$   
Diameter 75 mm

A wide variety of thin wafers are available from Virginia Semiconductor, 1501 Powhatan St, Fredericksburg, VA 22401, Tel(540)373-2900, Fax(540)371-0371

Pure Sil, Inc., One Silicon Way, Bradford, PA 16701 Tel(814)362-3586, Fax (814)362-2135 supplies double sided polished, thinned wafers. A 4" wafer thickness=380 $\mu\text{m}$ , double sided polished costs \$15 each in small qty.

Double sided polished wafers can be made at RIT by using CMP equipment. Polishing takes about 1 hour per wafer.

## QUARTZ WAFERS

Hoya Electronics Corporation  
Electronic Materials Division  
2 Corporate Ave  
Suite 341  
Shelton, CT 06484

Product Code: QZ 4w55325-15C-STD

Standard Wafer Substrate, 4 inch quartz wafer with flat, 0.525 mm thick minimum Quantity 20 @ \$39 each (Quote: August 29, 1996) Fax P.O. to (203)929-2529, Tel (203)929-1677

Quartz wafers can be made from quartz mask blanks 60 mils thick (1.524mm). Most glass shops have standard circle cutters and can easily cut and sand edges making nice substrates.

**MECHANICAL PROPERTIES**

	Density gm/cm <sup>3</sup>	Youngs Modulus 10 <sup>12</sup> dyne/cm <sup>2</sup>	Yield Strength 10 <sup>10</sup> dyne/cm <sup>2</sup>	Ultimate Strength 10 <sup>10</sup> dyne/cm <sup>2</sup>	Knoop Hardness Kgm/mm <sup>2</sup>	Poisson's Ratio
Single Crystal Silicon	2.33	1.9	12	15	850	0.28
Poly Silicon	2.33	1.5	12	18	850	0.28
Silicon Dioxide	2.19	0.73	8.4	16	570	0.3
Silicon Nitride	3.44	3.85	14	28	3486	0.3
Aluminum	2.7	0.68	17		150	0.334
Nickel	8.9	2.07	59	310	112	0.31
Chrome	7.19	2.54		83	170	0.3
Copper	8.96	1.20	33	209		0.308
Gold	19.3	0.78		103		0.44
Tungsten	19.3	4.1	4	98	350	0.28
Titanium	4.5	1.05	140	220	100	0.34
Tantalum	16.6	1.86		35	124	0.35

**10 dyne/cm<sup>2</sup> = 1 newton/m<sup>2</sup>**

Metals Handbook

***THERMAL PROPERTIES***

	MP °C	Coefficient of Thermal Expansion ppm/°C	Thermal Conductivity w/cmK	Specific Heat cal/gm°C
Diamond	3825	1.0	20	0.169
Single Crystal Silicon	1412	2.33	1.5	0.167
Poly Silicon	1412	2.33	1.5	0.167
Silicon Dioxide	1700	0.55	0.014	
Silicon Nitride	1900	0.8	0.185	
Aluminum	660	22	2.36	0.215
Nickel	1453	13.5	0.90	0.107
Chrome	1890	5.1	0.90	0.03
Copper	1357	16.1	3.98	0.092
Gold	1062	14.2	3.19	0.031
Tungsten	3370	4.5	1.78	0.031
Titanium	1660	8.9	0.17	0.043
Tantalum	2996	6.5	0.54	0.033
Air			0.00026	0.24
Water	0		0.0061	1.00

$$1 \text{ watt} = 0.239 \text{ cal/sec}$$

***COEFFICIENTS OF THERMAL EXPANSION***

	Thermal Expansion
Silicone Elastomers	275-300 ppm/°C
Unfilled Epoxies	100-200
Filled Epoxies	50-125
Epoxy, glass laminates	100-200
Epoxy, glass laminate, xy axis	12-16
Aluminum	20-25
Copper	15-20
Alumina Ceramic	6.3
Type 400 Steels	6.3-5.6
Glass Fabric	5.1
Borosilicate Glass	5.0
Silicon	2.4
Inconel	2.4
Nickel-iron alloy (30 Ni - 61 Fe)	1.22
Quartz	0.3

***THERMAL EXPANSION***

1. How much will a 500 um long bar of aluminum expand if it is heated 200 C above ambient?

$$\begin{aligned} \text{DL/L} &= 22 \text{ ppm/C} \\ &= 22 \times 200 = 4400 \text{ ppm} = 4400\text{E-6} \end{aligned}$$

$$\text{DL} = 4400\text{E-6} \times 500 \text{ um} = 2.2 \text{ um}$$



***THERMAL CONDUCTIVITY***

1. What is the thermal resistance to ambient for a square heater of width = 100 um built on silicon dioxide layer 1 um in thickness?

$$\begin{aligned} Q_{th} &= 1/G_{th} \quad t/LW = Rho_{th} \quad t/LW \\ &= (1/0.014) \quad 1E-4/(100E-4)^2 \\ &= 71.4 \text{ K/w} \end{aligned}$$

***ELECTRICAL PROPERTIES***

	DC Resistivity ohm-cm	Permittivity Relative
Single Crystal Silicon	0.0001-1E15	11.7
Poly Silicon	0.0001-1E15	11.7
Silicon Dioxide	~1E15	3.9
Silicon Nitride	~1E15	7.5
Aluminum	2.73E-6	-
Nickel	7.16E-6	-
Chrome	13.5E-6	-
Copper	1.73E-6	-
Gold	2.25E-6	-
Tungsten	5.65E-6	
Titanium	55.4E-6	
Tantalum	0.9E-6	

**RESISTANCE**

1. How hot will a chrome resistance heater get if  $L=1000\mu\text{m}$  and  $W=10\ \mu\text{m}$  get using 10 volts power supply and 1000Å film thickness?

$$\begin{aligned} R &= \text{Rho} \cdot L / \text{Area} \\ &= 13.5\text{E-6} \cdot 1000\mu\text{m} / (0.1\mu\text{m} \times 10\mu\text{m}) \\ &= 135 \text{ ohms} \end{aligned}$$

$$P = V^2/R = 100/135 = 0.74 \text{ w}$$

$$\begin{aligned} \Delta T &= P/Q = 0.74\text{w} \times 71.4 \text{ K/w} = 52.8 \text{ K} \\ &\text{or } 52.8 \text{ }^\circ\text{C above ambient} \end{aligned}$$

*OPTICAL PROPERTIES*

Index of Refraction

	n
Single Crystal Silicon	3.44
Poly Silicon	3.44
Silicon Dioxide	1.47
Silicon Nitride	1.98
Glass	~1.5

*TYPES OF SILICON DIOXIDE*

Thermal -Dry

Thermal - Wet

LPCVD, LTO if <400 C

LPCVD, HTO if >900 C

TEOS (no high temp anneal, short high temp anneal, long...)

PECVD TEOS (no high temp anneal, short high temp anneal, long...)

BPSG - 5%Boron/5%Phosphorous Silicon dioxide

Boron Doped Glass

Phosphorous Doped Glass

*GLASS*

Glass is used to seal lids on packages, form insulation for metal pins into packages.

	Working Temp °C	Thermal Expansion ppm/°C	Thermal Conductivity W/cm K
Borosilicate	1115	46	0.02
Lead Glass	560	84	
Silica Glass	1580	5.5	0.016
Soda Lime Glass	1005	92	

***SPECIALTY GLASS***

TEMPAX: ~matches thermal coefficient of expansion of silicon, can be used for anodic bonding (alkali rich glass)

PYREX 7740 can be used for anodic bonding

Hoya SD2

PMMA (Plexiglass)

*CERAMIC*

	Thermal Expansion ppm/°C	Thermal Conductivity W/m K	Youngs Modulus
AlN	2.7	150	
Alumina Al <sub>2</sub> O <sub>3</sub>	6.3	40	5.3
Beryllia BeO	7	300	
BN	3.8	60	
SiC (silicon carbide)	3.3	350	7.0
Diamond	1.0	2000	10.35

PZT (lead zirconium titanate) piezoelectric ceramic



## *POLYIMIDE*

Polyimide has a melting point of 450 C, can be spin coated and imaged with lithographic processes making it useful for many applications.

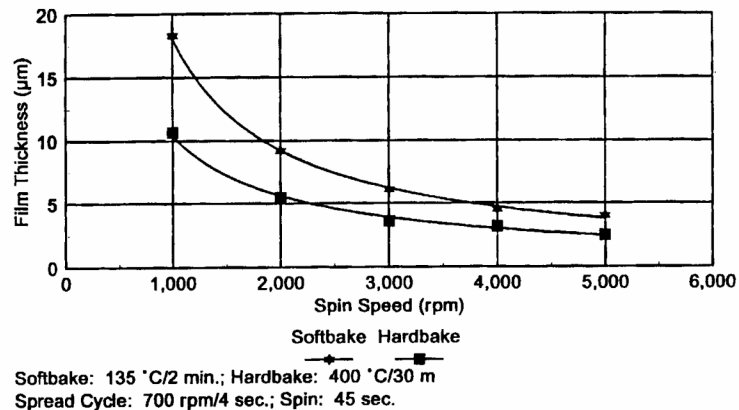
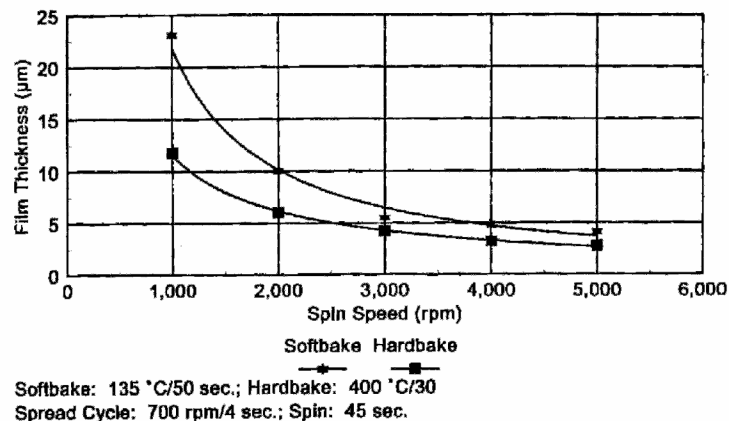
Using DuPont Corporations PI-2555 we can get film thickness between 2.5  $\mu\text{m}$  @ 5000 rpm and 5.0  $\mu\text{m}$  @ 1500 rpm. It is cured by placing on 120 °C hot plate for 30 min. and then on a 350 °C hot plate for 30 min. Multilayer coatings can give thickness greater than 10  $\mu\text{m}$ . (a 500 gm bottle costs ~\$250) Du Pont Co., Electronic Materials Division, Barley Mill Plaza, Reynolds Mill Building, Wilmington, DE 19898 (800)441-7543

OCG Microelectronic Materials, Belgium, makes a polyimide “Probimide 114A and 114A” which we have used. OCG Microelectronic Materials, 200 Massasoit Ave, East Providence, RI 02914 (Contact Henry Harding, Portsmouth, NH) 603-430-9849 or HAHarding@aol.com

These film are easily imaged using an aluminum barrier layer and conventional photoresist (such as Shipley System-8) followed by Oxygen Reactive Ion Etch.

*OCG Probimide 114A and 115A*

Process:  
 Spin 1000 rpm, 45 seconds  
 Softbake 135 C, 2 min  
 Hardbake 400 C, 30 min

*Probimide® 114A Spin Curve**Probimide® 115A Spin Curve*

**SOLDER**

	Solidus	Liquidus
70Sn/30Pb	183 °C	193 °C
63Sn/37Pb	183	183
60Sn/40Pb	183	190
50Sn/50Pb	183	216
40Sn/60Pb	183	238
10Sn/75Pb	268	302
42Sn/58Bi	138	138
70In/30Pb	160	174
70Sn/18Pb/12In	162	162
90Pb/5In/2.5Ag	300	310
97.5Pb/2.5Ag	303	303

## *WAX FOR WORKING WITH WAFERS & PIECES*

Mr. Roger Young (716)235-7698 or (800)724-7447, Fax (716)235-7254,  
J.H. Young Co.  
8 Symington Place  
Rochester, NY 14611-2409

Makes adhesives, pitches and wax for stacking, blocking of etch, bonding for polishing, etc. Product 7036 dissolves in Acetone and melts at about 100 °C. The price per stick is \$10. They make ten or more products of various hardness, melting temperatures, etc., that dissolves in a variety of solvents.

Example: For stacking wafer pieces on a full wafer for processing. Place a full wafer on aluminum foil on a hot plate set at 150 °C, rub the wax stick on the surface to deposit some wax, use a microscope slide to scrape off excess wax leaving a smooth thin layer. Place wafer pieces on the wax, press down, remove and cool, rinse off excess wax in acetone. Remove pieces by heating and cleaning with acetone, alcohol, DI water.

**EPOXY MATERIALS**

**Master Bond Inc**  
**154 Hobart St.**  
**Hackensack, NJ 07601**

(201) 343-8983  
 Offers over 100 different epoxy products, adhesives, sealants and coating.

Product Name	Mix Ratio	Viscosity RT, cps	Set-up time, RT	Cure Schedule	Applications
EP21TPND	100/100	thixotropic	30min	<u>48hrs@RT+2hrs@200F</u>	Polysulfide modified, Fuel and oil resistant sealant
EP30LTE	100/10	17,000	30min	<u>24hrs@RT+3hrs@200F</u>	Exceptionally low coefficient of expansion, low shrinkage
EP30	100/10	2000	25min	<u>24hrs@RT+2hrs@200F</u>	Clear system for optical and fiber optic bonding
EP77M-F	100/100	paste	8min	<u>1hr@150F+8hr@300F</u>	Electrically conductive silver filled epoxy
EP121AO	100/80	50,000	15hrs	<u>3hrs@200F+9hrs@200F</u>	Thermally conductive potting and encapsulation
SuperGel#7	100/100	500	3hrs	<u>60hrs@RT+3hrs@200C</u>	Soft resilient, transparent epoxy gel
SteelMaster 43HT	100/20	Thixotropic	25 min	<u>24hr@RT+2hr@200C</u>	Machinable, stainless steel filled

*EPOXY TECHNOLOGY, INC.*



14 Fortune Drive  
Billerica, MA 01821-3972  
(508) 667-3805

\$100/oz.

**REFERENCES**

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silicone materials
12. <http://web.mit.edu/3.091/www/pt/>

WebElements periodic table of the elements - the p

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Favo: ...

Jump start: select an element from the periodic table.

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Period																		
1	1 H																	2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba	* 71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra	** 103 Lr	104 Th	105 Pa	106 U	107 Np	108 Pu	109 Am	110 Cm	111 Bk	112 Cf	113 Es	114 Fm	115 Md	116 No	117 Uus	118 Uuo
*Lanthanoids			* 57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb		
**Actinoids			** 89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No		

Notes

WebElements aims to be a high quality source of information on the WWW relating to the periodic table. Coverage is such that professional scientists and students at school will all find something useful. You will find thousands of graphics showing elements structures and periodic properties here. Currently, most information

Internet



## ***HOMEWORK – MEMS MATERIALS***

1. Calculate how much the distance between two points on a silicon wafer 6 inches apart will increase for a  $2^{\circ}\text{C}$  increase in wafer temperature.
2. A square  $100\ \mu\text{m}$  by  $100\ \mu\text{m}$  polysilicon resistor is built on a  $1\ \mu\text{m}$  thick silicon dioxide layer. How much power (in watts) must be dissipated in the resistor to heat the resistor  $175\ ^{\circ}\text{C}$  above ambient?
3. Compare problem 2 to a similar heater built of polysilicon suspended in air,  $1\ \mu\text{m}$  above the silicon substrate.
4. Which material is the hardest, strongest, lightest, best thermal conductor, worst thermal conductor, largest thermal expansion, smallest thermal expansion. Give parameter values and units for the material.