

## Mobility and toxicity of heavy metals in bottom sediments of Rybnik reservoir

A. Baran<sup>1</sup> and M. Tarnawski<sup>2</sup>

<sup>1</sup>Department of Agricultural and Environmental Chemistry, University of Agriculture in Krakow, al. A. Mickiewicza 21, 31-120 Krakow, POLAND, Agnieszka.Baran@ur.krakow.pl

<sup>2</sup>Department of Water Engineering<sup>2</sup>, University of Agriculture in Krakow, al. A. Mickiewicza 24/28, 30-059 Kraków, POLAND, rmtarnaw@cyf-kr.edu.pl

**Abstract.** Heavy metals are one of the most important factors among many others in biosphere pollution. Mobility and toxicity of metals associated with bottom sediments are generally affected by metal speciation and sediments compositions. The aims of this study were to investigate the distribution of metal speciation in sediment collected from Rybnik dam reservoir and to assess their toxicity for aquatic ecosystem. The speciation analysis of metal was performed using the three-step method of sequential fractionation by means of the modified BCR technique. Toxicity assessment of the bottom sediment samples was performed using tests: Phytotoxkit<sup>TM</sup>, Ostarcodotoxkit F<sup>TM</sup>, Microtox<sup>®</sup>. According to the above results, the mobilization potential of heavy metals in sediments in a decreasing order (fraction I+II): Zn > Cd > Ni > Pb > Cu > Cr. According to the Risk Assessment Code (RAC), the sediments having at low risk (Pb, Cu, Cr), medium risk (Cd, Ni) and high risk (Zn). The analysis of all sediment samples collected from the suited reservoir showed that Classes III (acute hazard  $50\% \leq PE < 100\%$ ) was represented by 80% samples.

**Key words:** metal speciation, mobility, sediment toxicity, battery biotests

### Introduction

Bottom sediments accumulated in water reservoirs constitute a very important part of ecosystems, play an important role in their functioning and element cycling between individual components of soil and groundwater system. Bottom sediments play the role of a natural filter and are indicators of the degree of environmental degradation. Heavy metals constitute a significant part of these pollutants, which at some concentrations may be toxic for aquatic ecosystem (Baran et al. 2011). Mobility and toxicity of metals associated with bottom sediments are generally affected by metal speciation and sediments compositions (Lin et al. 2003, Madeyski et al. 2009). Metals in exchangeable, carbonate-bound, and Fe/Mn oxide-bound speciation are considered to be more mobile and bioavailable. The organic matter-bound and residual metals are stable and non-bioavailable. An assessment of the environmental risks requires the measurement not only the total contents of heavy metal in sediment, but also for the amounts in each binding form. Mankiewicz-Boczek et al. (2008) described a need to apply a battery biotests for integral and ecologically meaningful evaluation of hazard of water and sediments.

The aims of this study were: (1) to investigate the distribution of metal speciation in sediment collected

from dam reservoir in the conditions of anthropomixion, (2) to assess their toxicity for aquatic ecosystem. To obtain information may provide a better understanding of environmental risks of heavy metal in sediment.

### Materials and Methods

In this study, surface sediment (0-10 cm) samples were collected from four stations located in Rybnik reservoir (Southern Poland). In order to get a representative samples for each stations, several samples were collected and mixed together. Sediments were sampled using an Ekman dredge. Once in the laboratory, the sediment samples were air dried, homogenized in a mortar, sieved to pass through a 2-mm stainless steel sieve and stored in polyethylene containers. The speciation analysis of metal was performed using the three-step method of sequential fractionation by means of the modified BCR technique (Mossop and Dawson 2003): fraction I – exchangeable and acid soluble fraction, extractable with CH<sub>3</sub>COOH at 0.11 mol · dm<sup>-3</sup> concentration and pH=2; fraction II – forms associated with free Fe and Mn oxides, extractable with NH<sub>2</sub>OHHCl at 0.5 mol · dm<sup>-3</sup> concentration and pH=1,5; fraction III – forms bonded to organic matter, extractable with hot 30% H<sub>2</sub>O<sub>2</sub> and then the mineralization products re-extracted with CH<sub>3</sub>COONH<sub>4</sub>

at  $0,5 \text{ mol} \cdot \text{dm}^{-3}$  concentration and  $\text{pH}=2$ ; fraction IV – residual forms, the difference between the total metal content and sum of the above three fractions. Total contents of heavy metals in the sediments were assessed after hot mineralization in a mixture of  $\text{HNO}_3$  and  $\text{HClO}_3$  acids (3:2). Metal concentrations in the obtained solutions were assessed using ICP-OES method on Optima 7300 DV PerkinElmer. Toxicity assessment of the bottom sediment samples was performed using direct – contact tests consisting of 5 species. The battery was composed of test species representative of different trophic levels of the food chain: producers (*Sorghum saccharatum*, *Lepidium sativum*, *Sinapis alba* – Phytotoxkit™), consumer (*Heterocypris incongruens* – Ostarcodotoxkit F™) and decomposer (*Vibrio fischeri* – Microtox®). The toxicity data has been classified according to the hazard classification (Persoone et al. 2003).

## Results and Discussion

### Chemical speciation

Based on the results shown in Figure 1, it was found that Zn mainly existed as exchangeable and acid soluble fractions (fraction I – 47%) and free Fe and Mn oxides fraction (fraction II – 27%). Cu was predominantly found organic matter-bound (fraction III – 72%). The participation of copper bound with the other fractions constituted, as a whole, less than 10% (fraction I + II) 20% (fraction IV). In the examined sediments Cd was bound first of all, with organic matter (fraction III - 43%) and free Fe and Mn oxides and (fraction III - 32%). The participation of cadmium bound with the other fractions constituted was about 22% with exchangeable and acid soluble fractions (fraction I) and 3% residual forms (fraction IV). The distribution of Pb was similar to Cu, being dominated by organic matter-bound (fraction III – 62%). Participation of Pb fraction bound with free Fe and Mn oxides, residual and exchangeable and acid soluble fractions was as follows: 25%, 12%, 1% in sediments from the reservoir at Rybnik. Based on the results shown in Figure 1, it was found that Ni mainly existed as organic matter (fraction III – 31%) and residual fractions (fraction IV – 30%). The

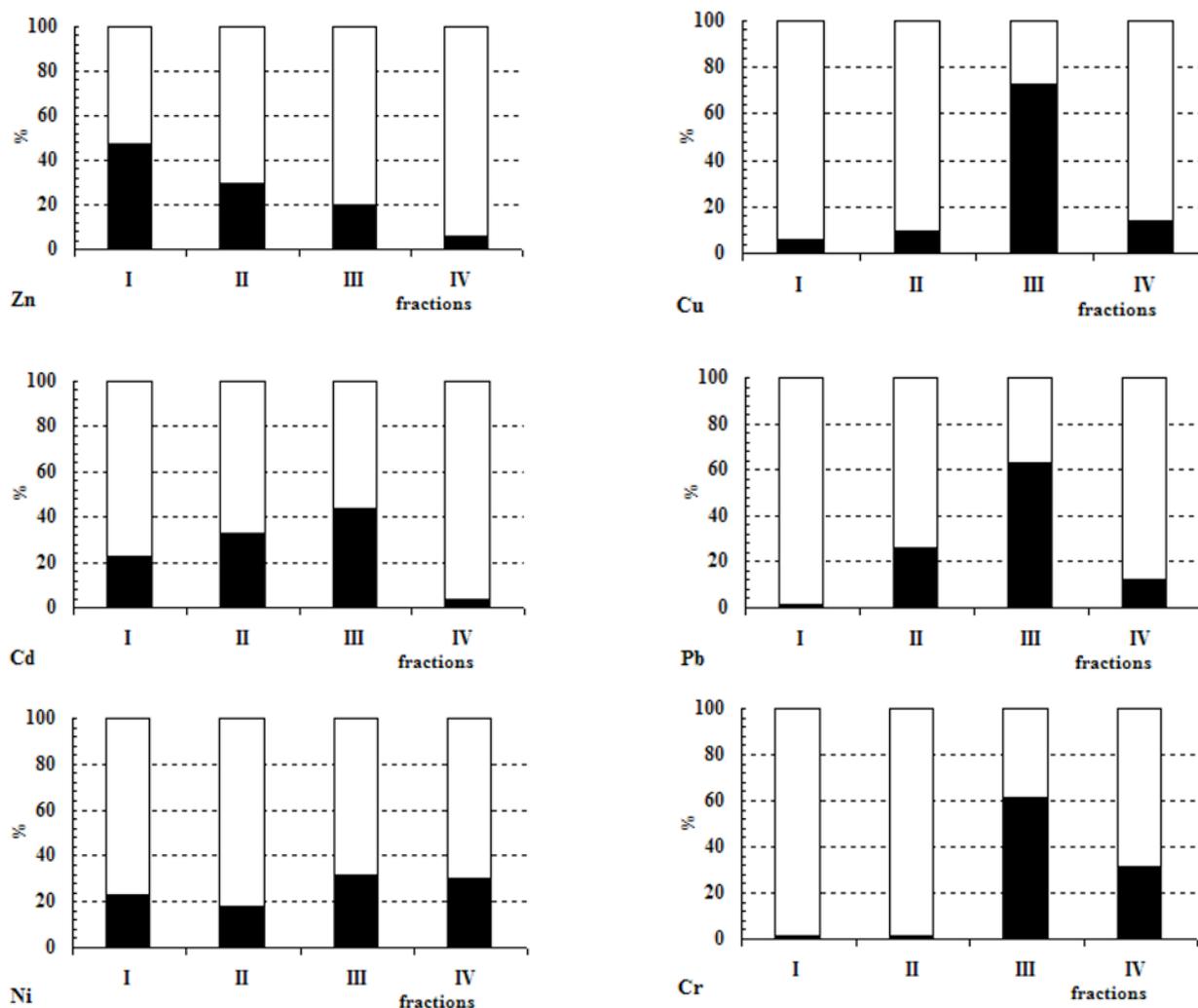


Fig. 1. Example of metal distribution in particular fractions.

behavior of Cr was similarly to Cu. A considerable percentage (about 60%) of Cr was mainly found in the residual fraction.

*Ecotoxicity test*

For the sediment samples, the highest number of toxic response was observed in the chronic Phytotoxkit™ with producers *Lepidium sativum* (24% responses), *Sinapis alba* (22% responses), respectively (Fig. 2). In the other proposed microbiotest with decomposer – bacteria *Vibrio fischeri* (Microtox®) and producer *Sorghum saccharatum* (Phytotoxkit™), the number of toxic response was respectively 20 and 15%. The lowest toxicity was determined in the chronic Ostarcodotoxkit F™ with consumer *Heterocypris incongruens* (Fig. 2).

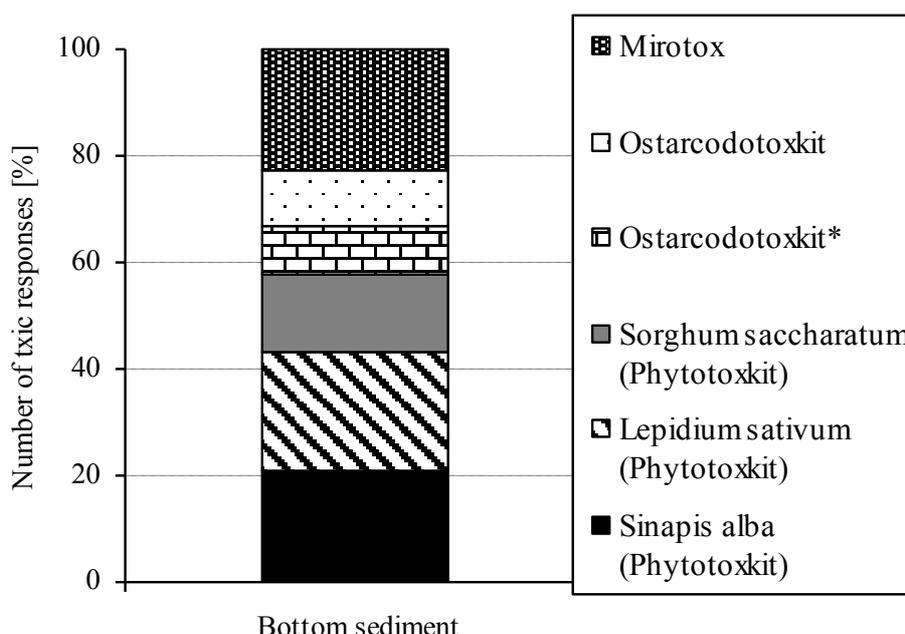
**Conclusion**

The sequential extraction for determination of metal speciation was good alternative for total metal analysis the environmental risks of heavy metals in sediments. According to the above results, the mobilization potential of heavy metals in sediments in a decreasing order (fraction I+II): Zn > Cd > Ni > Pb > Cu > Cr. Trace metals in the sediments were associated in the exchangeable fractions from 1 to 23%. Thus according to the Risk Assessment Code (RAC), these sediments were at low risk (Pb, Cu, Cr), and medium risk (Cd, Ni). According to RAC the concentration of Zn (47%) was a high risk. The analysis of all sediment samples collected from the suited reservoir showed that Classes III (acute

hazard 50%≤PE<100%) was represented by 80% samples. In Classes IV (high acute hazard PE=100%, in at least one test) was observed only in 20% analyzed samples.

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**Fig. 2.** The number of toxic response described for each microbiotest as the percentage from total number of tests (Ostarcodotoxkit\* - growth inhibition, Ostracodotoxkit – morality).

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