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Fundamental constants and tests of theory in Rydberg states of hydrogen-like ions

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Outline

- **Introduction**
- **Rydberg constant and the proton radius**
- **Rydberg ion states**
- **QED in Rydberg states**
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Introduction

Comparison of precision frequency measurements to quantum electrodynamics (QED) predictions for Rydberg states of hydrogen-like ions can yield information on values of fundamental constants and test theory.

For suitable transitions, the uncertainty in the theory of the energy levels is sufficiently small that such a comparison can yield an improved value of the Rydberg constant.

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Phys. Rev. Lett. **100**, 160404 (2008).

CODATA 2006 Least-squares adjustment of the fundamental constants - references

“CODATA recommended values of the fundamental physical constants: 2006”

P. J. Mohr, B. N. Taylor, and D. B. Newell

Reviews of Modern Physics, **80**, 633 (2008)

Journal of Physical and Chemical Reference Data **37**, 1187 (2008)

Values of the constants: physics.nist.gov/constants

Bibliography on constants: physics.nist.gov/constantsbib

Hydrogen and deuterium frequencies: physics.nist.gov/hdel

Fundamental Constants from the Least-Squares Adjustment

- Hydrogen and deuterium spectra related constants
 - Rydberg constant - R_∞ [6.6 x 10⁻¹²]
 - Proton rms charge radius - R_p [7.8 x 10⁻³]
 - Deuteron rms charge radius - R_d [1.3 x 10⁻³]
- Constants determined mainly by other data
 - fine-structure constant
 - electron-proton mass ratio
 - electron-deuteron mass ratio
- Relevant input data
 - 23 transition frequencies in H and D \geq [1.4 x 10⁻¹⁴]
 - R_p and R_d electron scattering data

Rydberg constant from the LSA

Experiment:

$$\nu_{\text{H}}(1\text{S}_{1/2} - 2\text{S}_{1/2}) = 2\,466\,061\,413\,187.074(34) \text{ kHz} \quad [1.4 \times 10^{-14}]$$

MPQ (2006)

Theory:

$$\nu_{\text{H}} = \frac{3}{4} R_{\infty} c \left[1 - \frac{m_e}{m_p} + \frac{11}{48} \alpha^2 - \frac{28}{9} \frac{\alpha^3}{\pi} \ln \alpha^{-2} - \frac{14}{9} \left(\frac{\alpha R_p}{\lambda_C} \right)^2 + \dots \right]$$

QED

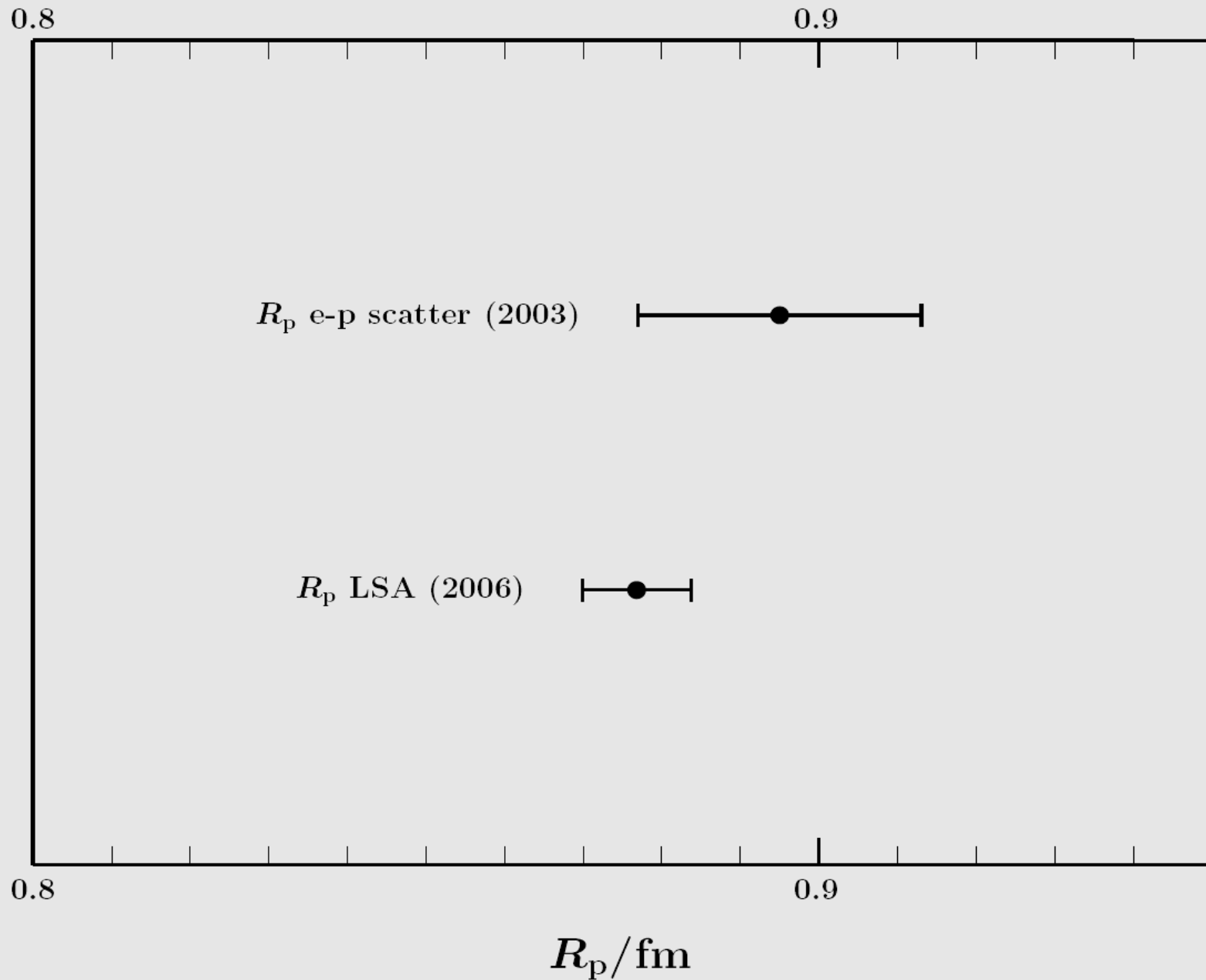
proton radius

Theory reviews:

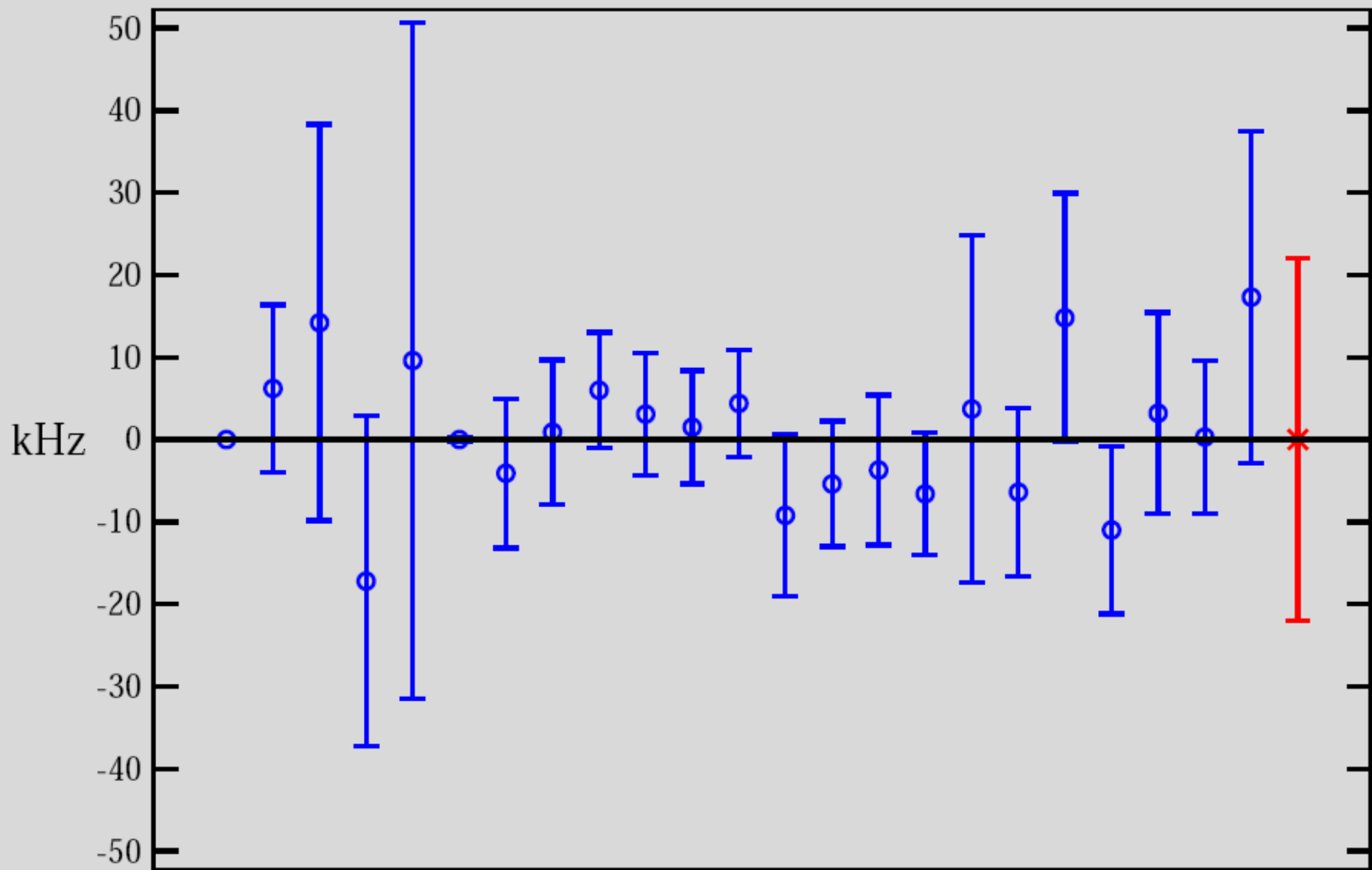
M. I. Eides, H. Grotch, V. Shelyuto, Phys. Reports **342**, 63 (2001).

J. R. Sapirstein and D. R. Yennie, in QED, ed. by T. Kinoshita (1990).

Proton charge radius from 2006 least squares adjustment



Transition frequencies in hydrogen and deuterium experiment - theory



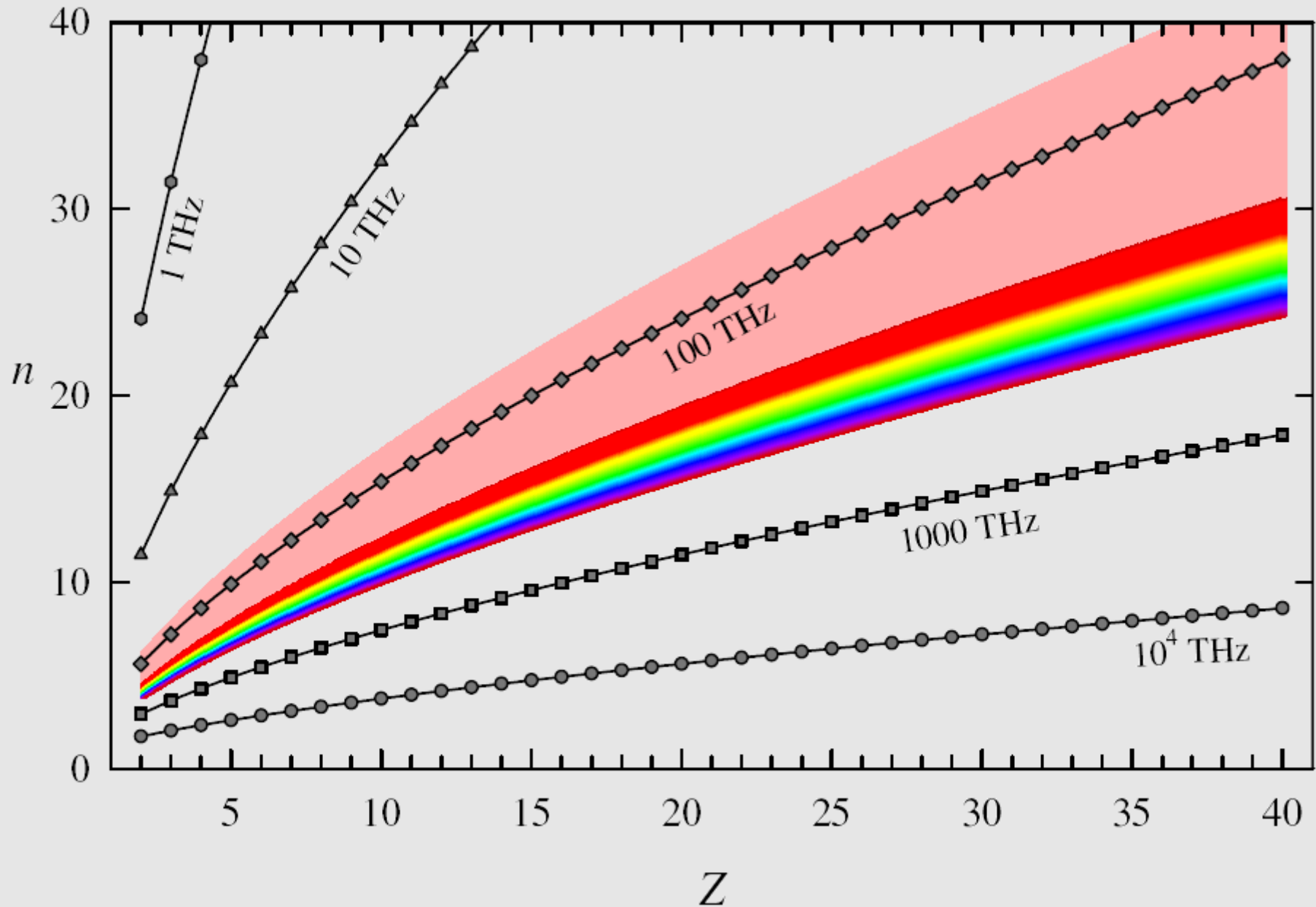
Rydberg states of hydrogen-like ions

Possible means to measure the Rydberg constant

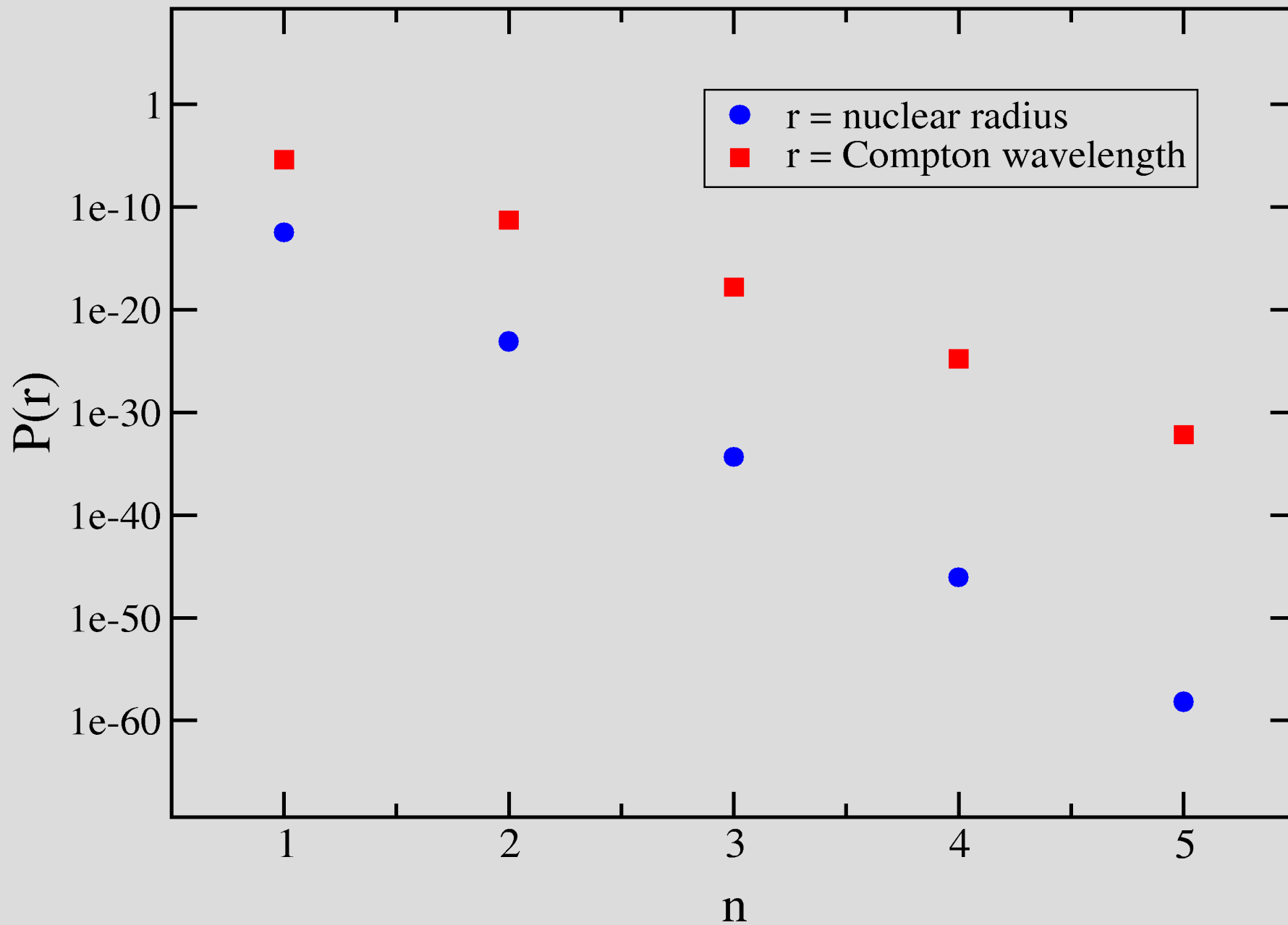
- Frequencies accessible with frequency comb lasers
- Small overlap with nucleus - no finite size correction
- Higher-order QED corrections are small

$$P(r) = \int_{|\mathbf{x}| \leq r} d\mathbf{x} |\psi(\mathbf{x})|^2 \approx \frac{1}{(2n+1)!} \left(\frac{2Zr}{na_0} \right)^{2n+1}$$

Transition frequencies in hydrogen-like ions



Electron probability to be inside radius r hydrogen-like neon; $l = n - 1$



Theory of Rydberg states for $l \geq 2$

$$E_n = E_{\text{DM}} + E_{\text{RR}} + E_{\text{QED}}$$

$$E_{\text{DM}} = 2hcR_\infty \left[\mu_r D - \frac{r_N \mu_r^3 \alpha^2}{2} D^2 + \frac{r_N^2 \mu_r^3 Z^4 \alpha^2}{2n^3 \kappa (2l + 1)} \right]$$

$$E_{\text{RR}} = 2hcR_\infty \frac{r_N Z^5 \alpha^3}{\pi n^3} \left\{ \mu_r^3 \left[-\frac{8}{3} \ln k_0(n, l) - \frac{7}{3l(l+1)(2l+1)} \right] \right. \\ \left. + \pi Z \alpha \left[3 - \frac{l(l+1)}{n^2} \right] \frac{2}{(4l^2 - 1)(2l + 3)} + \dots \right\}$$

$$\alpha^2 D = E_D - 1; \quad r_N = m_e/m_N; \quad \mu_r = 1/(1 + r_N)$$

One-photon contribution to hydrogen-like levels

$$E_{\text{SE}}^{(2)} = \frac{\alpha (Z\alpha)^4}{\pi n^3} F(Z\alpha) m_e c^2$$

$$F(Z\alpha) = A_{41} \ln(Z\alpha)^{-2} + A_{40} + A_{50} (Z\alpha) + A_{62} (Z\alpha)^2 \ln^2(Z\alpha)^{-2} \\ + A_{61} (Z\alpha)^2 \ln(Z\alpha)^{-2} + G_{\text{SE}}(Z\alpha) (Z\alpha)^2$$

$$F(Z\alpha) = A_{40} + A_{61} (Z\alpha)^2 \ln(Z\alpha)^{-2} + G_{\text{SE}}(Z\alpha) (Z\alpha)^2 \quad l \geq 2$$

$$A_{40} = -\frac{4}{3} \ln k_0(n, l) - \frac{A_1^{(2)}}{\kappa(2l+1)} \quad l \geq 2$$

$$G_{\text{SE}}(Z\alpha) = A_{60} + \dots$$

Two-photon contribution to hydrogen-like levels

$$E^{(4)} = \left(\frac{\alpha}{\pi}\right)^2 \frac{(Z\alpha)^4}{n^3} m_e c^2 F^{(4)}(Z\alpha)$$

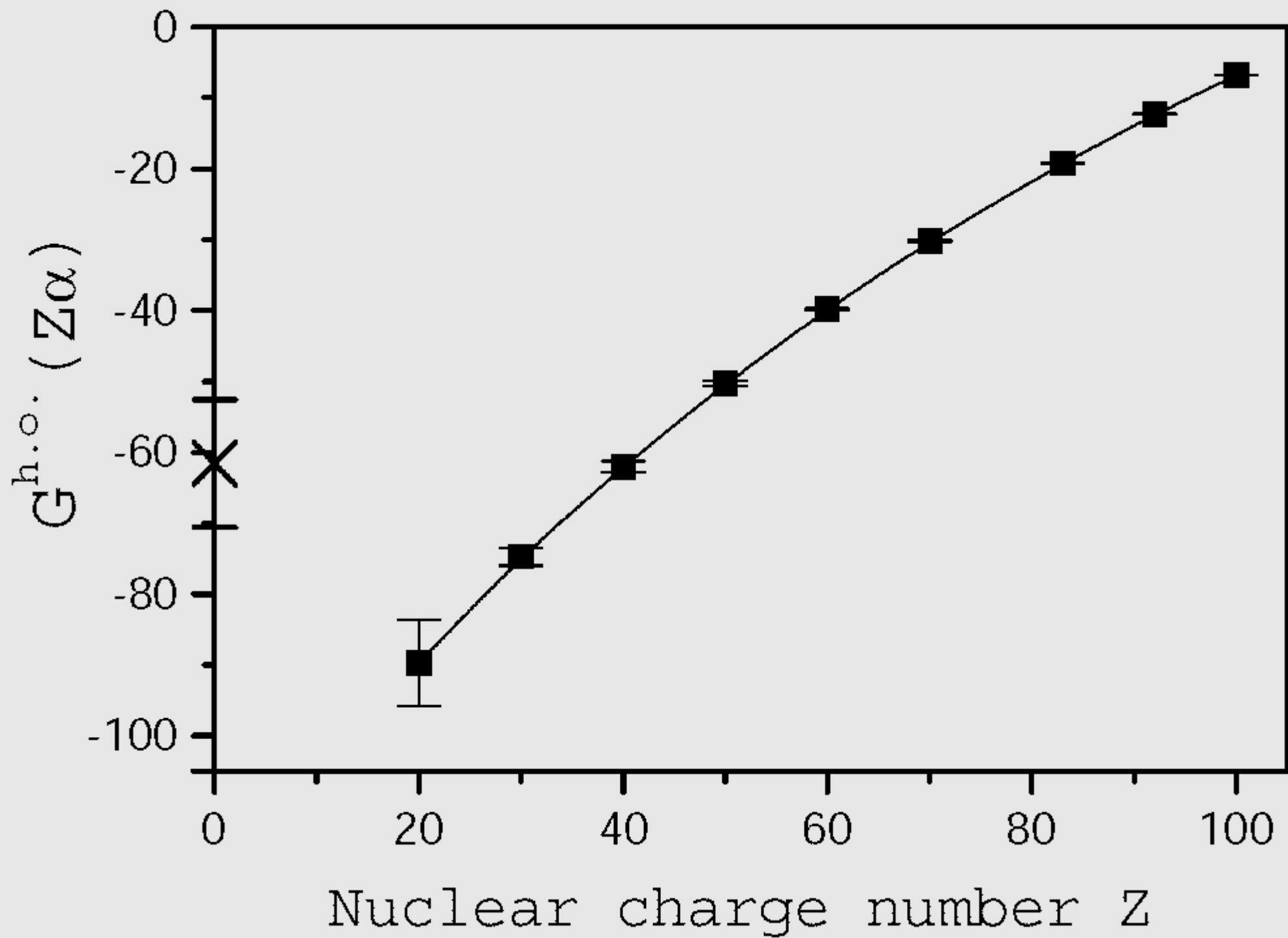
$$F^{(4)}(Z\alpha) = B_{40} + B_{50} (Z\alpha) + B_{63} (Z\alpha)^2 \ln^3(Z\alpha)^{-2} \\ + B_{62} (Z\alpha)^2 \ln^2(Z\alpha)^{-2} + B_{61} (Z\alpha)^2 \ln(Z\alpha)^{-2} + B_{60} (Z\alpha)^2 \\ + \dots$$

$$F^{(4)}(Z\alpha) = B_{40} + B_{60} (Z\alpha)^2 + \dots \quad l \geq 2$$

$$B_{40} = -\frac{A_1^{(4)}}{\kappa(2l+1)} \quad l \geq 2$$

electron anomalous magnetic moment

$$a_e(\text{QED}) = A_1^{(2)} \left(\frac{\alpha}{\pi}\right) + A_1^{(4)} \left(\frac{\alpha}{\pi}\right)^2 + A_1^{(6)} \left(\frac{\alpha}{\pi}\right)^3 + A_1^{(8)} \left(\frac{\alpha}{\pi}\right)^4 + \dots$$



Three- or more-photon contributions to hydrogen-like levels

$$E^{(6)} = \left(\frac{\alpha}{\pi}\right)^3 \frac{(Z\alpha)^4}{n^3} m_e c^2 [C_{40} + C_{60}(Z\alpha)^2 + \dots] \quad l \geq 2 \quad (?)$$

$$C_{40} = -\frac{A_1^{(6)}}{\kappa(2l+1)} \quad l \geq 2$$

Total QED contribution to hydrogen-like levels

$$E_{\text{QED}} = 2hcR_\infty \frac{Z^4 \alpha^2}{n^3} \left\{ -\mu_r^2 \frac{a_e}{\kappa(2l+1)} + \mu_r^3 \frac{\alpha}{\pi} \left[-\frac{4}{3} \ln k_0(n, l) + \frac{32}{3} \frac{3n^2 - l(l+1)}{n^2} \right. \right. \\ \left. \left. \times \frac{(2l-2)!}{(2l+3)!} (Z\alpha)^2 \ln \left[\frac{1}{\mu_r (Z\alpha)^2} \right] + (Z\alpha)^2 G(Z\alpha) \right] \right\}$$

$$G(Z\alpha) = A_{60} + A_{81}(Z\alpha)^2 \ln(Z\alpha)^{-2} + A_{80}(Z\alpha)^2 + \dots \\ + \frac{\alpha}{\pi} B_{60} + \dots + \left(\frac{\alpha}{\pi}\right)^2 C_{60} + \dots$$

Calculated values of A_{60}

- NRQED effective operators.
- Schrödinger Coulomb Green function on numerical grid.
- Sum over discrete-energy pseudo-state spectrum.
- B. Wundt poster on the calculation.

TABLE I. Calculated values of the constant A_{60} . The numbers in parentheses are standard uncertainties in the last figure.

n	l	$2j$	κ	A_{60}	$2j$	κ	A_{60}
13	11	21	11	$0.679\,575(5) \times 10^{-5}$	23	-12	$4.318\,998(5) \times 10^{-5}$
13	12	23	12	$0.469\,973(5) \times 10^{-5}$	25	-13	$2.729\,475(5) \times 10^{-5}$
14	12	23	12	$0.410\,825(5) \times 10^{-5}$	25	-13	$2.979\,937(5) \times 10^{-5}$
14	13	25	13	$0.296\,641(5) \times 10^{-5}$	27	-14	$1.945\,279(5) \times 10^{-5}$
15	13	25	13	$0.252\,108(5) \times 10^{-5}$	27	-14	$2.116\,050(5) \times 10^{-5}$
15	14	27	14	$0.189\,309(5) \times 10^{-5}$	29	-15	$1.420\,631(5) \times 10^{-5}$
16	14	27	14	$0.155\,786(5) \times 10^{-5}$	29	-15	$1.540\,181(5) \times 10^{-5}$
16	15	29	15	$0.121\,749(5) \times 10^{-5}$	31	-16	$1.059\,674(5) \times 10^{-5}$

TABLE II. Transition frequencies between the highest- j states with $n = 14$ and $n = 15$ in hydrogenlike helium and hydrogenlike neon.

Term	${}^4\text{He}^+$ ν (THz)	${}^{20}\text{Ne}^{9+}$ ν (THz)
E_{DM}	8.652 370 766 008(58)	216.335 625 5746(14)
E_{RR}	0.000 000 000 000	0.000 000 000 1
E_{QED}	-0.000 000 001 894	-0.000 001 184 1
Total	8.652 370 764 114(58)	216.335 624 3907(14)

TABLE III. Sources and estimated relative standard uncertainties in the theoretical value of the transition frequency between the highest- j states with $n = 14$ and $n = 15$ in hydrogenlike helium and hydrogenlike neon.

Source	He^+	Ne^{9+}
Rydberg constant	6.6×10^{-12}	6.6×10^{-12}
Fine-structure constant	7.0×10^{-16}	1.7×10^{-14}
Electron-nucleus mass ratio	5.8×10^{-14}	1.2×10^{-14}
a_e	5.1×10^{-20}	1.3×10^{-18}
Theory: E_{RR} higher order	6.2×10^{-17}	2.4×10^{-14}
Theory: $E_{\text{QED}}A_{81}$	1.7×10^{-18}	1.6×10^{-14}
Theory: $E_{\text{QED}}B_{60}$	8.6×10^{-18}	5.4×10^{-15}

Conclusion

- The current accuracy of the Rydberg constant is limited by uncertainties in the theory caused mainly by the proton charge radius.
- In Rydberg states of hydrogen-like atoms the overlap with the nucleus is extremely small.
- Higher-order binding effects in the QED corrections are suppressed in Rydberg states.
- Suitable combinations of n and Z provide Rydberg states with infrared or visible frequencies suitable for interrogation with a frequency comb.
- Measurement of the Rydberg constant in Rydberg states of hydrogen-like ions may be possible.