



Synthesis, Characterization and Antimicrobial Evaluation of New Chalcone Derivatives From 3-benzyloxy-4-methoxybenzaldehyde

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ABSTRACT

A series of chalcone derivatives (2a–i) were prepared via the reaction of 3-benzyloxy-4-methoxybenzaldehyde with the appropriately acetophenone derivatives. The structures of all the new chalcone derivatives (2a–i) synthesized in this study were established on the basis of ¹H NMR and ¹³C NMR spectral data, and elemental analyses. The antibacterial activity of the synthesized compounds (2a–i) was carried out by well diffusion and MIC method.

Key words: chalcone, X-ray analysis, Claisen–Schmidt Condensation, antibacterial activity.

INTRODUCTION

Chalcones are products of condensation of simple or substituted aromatic with simple or substituted acetophenones in presence of alkali. Chalcone constitute an important group of natural products and some of them possess a wide range of biological activities such as anticancer¹ antitubercular², antiviral³, also they are used as anti-malarial⁴, anti protozoal⁵, anti-inflammatory⁶, immunomodulatory⁷⁻⁸, nitric oxide inhibition⁹, tyrosinase inhibition¹⁰, cytotoxic¹¹, antimicrobial¹², Geiger and Conn¹³ during their chemical studies on the structure of clavacin found that a structural feature which was responsible for antibacterial activity was α , β unsaturated keto functional group.

These molecules are also used as starting materials in the synthesis of UV absorption filters in polymers, photorefractive polymers, photosensitizers in Bcolor films, sweeteners in food technology, and in holographic recording technology. A natural medicine genus *Angelica* is known to contain large number of naturally occurring chalcones¹⁴. Chalcone derivatives are recognized for NLO properties and have good crystallization ability¹⁵. Structure of few related chalcones viz., (2E)-3-(biphenyl-4-yl)-1-(4-methoxyphenyl)prop-2-en-1-one (Fischer et al., 2007)¹⁶, (E)-3-(2,6-Dichlorophenyl)-1-(4-methoxyphenyl)prop-2-en-1-one (Benmekhbi et al., 2009)¹⁷.

Pharmacological properties of chalcones are due to the presence of both α,β unsaturation¹⁸ and an aromatic ring. Constant interest in chalcones has resulted in syntheses of new derivatives using both classical¹⁹⁻²⁰ and combinatorial techniques²¹.

In this study, a series of new chalcone-like compound (2a-i) were synthesized by the reaction of 3-benzyloxy-4-methoxybenzaldehyde with the appropriately acetophenon derivatives. The structures of all the chalcone derivatives (2a-i) synthesized in this study were established on the basis of ¹H NMR and ¹³C NMR spectral data, and elemental analyses. The structure of compound 2i was further confirmed by X-ray analysis of single crystal.

RESULTS AND DISCUSSION

Melting points of the compound were measured using an Electrothermal 9100 apparatus. IR spectrums (KBr or liquid) were taken by a Jasco FT-IR-430 IR spectrophotometer. ¹H and ¹³C NMR spectra were recorded using a Bruker Avance III instrument using tetramethylsilane (TMS, δ 0.00) for ¹H NMR and DMSO for ¹³C NMR spectroscopy as internal reference standards; *J* values were given in hertz. The multiplicities of the signals in the ¹H NMR spectra are abbreviated by s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet),

Reagent

3-benzyloxy-4-methoxybenzaldehyde and appropriately acetophenon derivatives were commercial products with the highest reagent grade.

Chemistry

To a mixture of 3-(benzyloxy)-4-methoxybenzaldehyde (2g, 0.008 mol) and appropriately acetophenon derivatives (0.008 mol) in ethanol 20 ml in the presence of a catalytic amount of sodium hydroxide solution (5 ml) was added slowly with stirring (6 h), neutralized with HCl solution (10%) the contents of the flask were poured into ice cold water (500 ml) and left to stand for 5 h, the organic layer was dried over anhydrous Na₂SO₄, (Scheme 1) The resulting crude solid was filtered and purified by recrystallization in ethanol. The structure of the compound 9(2i) was further confirmed by X-ray analysis of single crystal. Crystal

suitable for x-ray analysis was grown by slow evaporation of a mixture acetone/ ethanol solution at room temperature. The crystal used for data Collection was of the dimension) 0.51 x 0.31x0.15mm(

X-ray analysis

The compound, 2i (C₂₃H₁₉ClO₃), exists in an E conformation with respect to the C=C bond. The central benzene ring forms a dihedral angle of 88.96 (2)° with the chlorobenzene ring and a dihedral angle of 22.53 (2)° with the terminal benzene ring. No significant intermolecular interactions are observed (fig1, fig2).

Refinement

All H atoms were localized in Fourier maps but introduced in calculated positions and treated as riding on their parent C atoms with C—H = 0.95Å to 0.99Å and Uiso(H) = 1.2 or 1.5Ueq(C).

Crystallographic data and details of the data collection and structure solution and refinements are listed in **Table 2**

Data

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-phenylprop-2-en-1-one (2a)

Viscousoilbp201-210°C ¹H NMR(300MHZ, DMSO) α =7,90(d, *J*=15,5, 1H, CH=), 7,45-7,84(m, 5H, ArH, 2H, CH=CH), 7,56(d, *J*=15,5, 1H, =CH), 7,19(s, 5H, ArH), 6,62-6,74(m, 3H, Ar), 5,20(s, 2H, CH₂-O), 3,81(s, 3H, CH₃-O) ¹³C NMR(300MHZ, DMSO) δ =189,71, 149,7, 149, 145,2, 141, 137,2, 134,6, 129,9, 129,2 (2C), 127,7, 127,2 (2C), 121,2, 119,7, 115,2, 111, 72,2, 56,2. IR (liquid): 3135,3032, 3009,2811, 1683, 1656, 1525,1457, 1367, 1056, 806,775, 733 Anal. calcd. for C₂₃H₂₀O₃C, 80.21; H, 5.85; O, 13.94. Found C, 80.05; H, 5.90; O, 14.01.

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-(4-chlorophenyl)prop-2-en-1-one (2b)

Yellowish crystals, mp 145–150°C. ¹H NMR(300MHZ, DMSO) α =8,20(d, *J*=8,5, 2H), 7,96(d, *J*=8,5, 2H), 7,84(d, *J*=15,5, 1H), 7,74(d, *J*=15,5, 1H), 7,73(d, *J*=1,2, 1H), 7,68(d, *J*=8,5, 2H), 7,6(d, *J*=8,4, 1H), 7,52(d, *J*=8,4, 2H), 7,35-7,45(m, 2H), 7,08(d, *J*=8,4, 1H), 5,20(s, 2H), 3,81(s, 3H) ¹³C NMR(300MHZ, DMSO) δ =189,71, 149,7, 149,

Table 1: The new chalcone derivatives

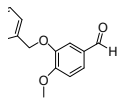
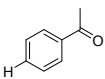
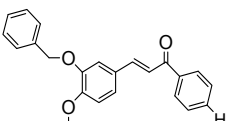
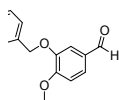
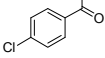
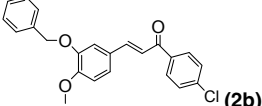
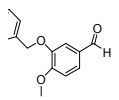
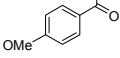
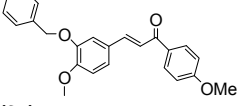
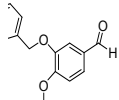
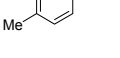
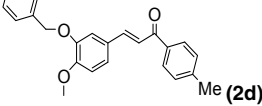
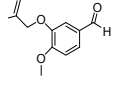

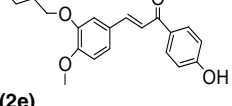
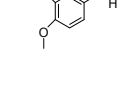

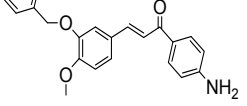
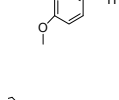

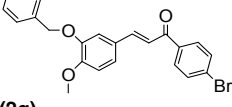
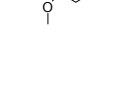

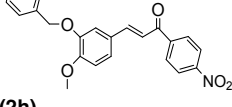
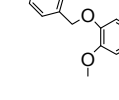
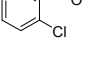
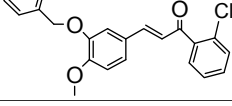
Entry aromatic	acetophenonderivatives	products	Isolated yield (%)
		 (2a)	83(%)
		 (2b)	87(%)
		 (2c)	90(%)
		 (2d)	85(%)
		 (2e)	80(%)
		 (2f)	82(%)
		 (2g)	80(%)
		 (2h)	
			80(%)

Table 2: Crystallographic Data Collection and Structure Refinement Parameters of 2i

Crystal data C ₂₃ H ₁₉ ClO ₃ Mr= 378.83 Triclinic, P1 Hall symbol: -P 1 a = 7.9049 (4) Å b = 10.9621 (5) Å c = 11.8899 (5) Å α = 102.956 (2)° β = 105.739 (2)° γ = 96.418 (2)° V = 949.93 (8) Å ³	Z = 2 F(000) = 396 Dx = 1.324 Mg m ⁻³ Mo Kα radiation, λ = 0.71073 Å Cell parameters from 2479 reflections θ = 2.7–27.4° μ = 0.22 mm ⁻¹ T = 150 K Prism, colourless 0.51 × 0.32 × 0.15 mm
Data collection APEXII, Bruker-AXS diffractometer Graphite monochromator CCD rotation images, thin slices scans Absorption correction: multi-scan [Sheldrick, G.M. (2002). SADABS Bruker AXS Inc., Madison, Wisconsin, USA] Tmin = 0.884, Tmax = 0.967	8538 measured reflections 4333 independent reflections 3336 reflections with I > 2σ(I) Rint = 0.027 θmax = 27.5°, θmin = 3.0° h = -10→10 k = -12→14 l = -15→12
Refinement Refinement on F ₂ Least-squares matrix: full R[F ₂ > 2σ(F ₂)] = 0.043 wR(F ₂) = 0.107 S = 1.04 4333 reflections 245 parameters 0 restraints Primary atom site location: structure-invariant directmethods	Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained w = 1/[σ ² (F _o) 2] + (0.0425P) ² + 0.2095P] where P = (F _o 2 + 2F _c 2)/3 (Δσ) _{max} = 0.002 Δρ _{max} = 0.25 e Å ⁻³ Δρ _{min} = -0.26 e Å ⁻³

145,2,141,2, 140,1, 136, 131,3(2C), 129,4(2C), 129(2C), 128,5, 127,7, 127,2(2C), 121,4, 119,7, 115,2, 111,6, 71,2, 56,2. IR (KBr): 3125, 3112, 3019, 2819, 1673, 1662, 1535, 1551, 1467, 1368, 1150, 816, 765, 723 Anal. calcd. for C₂₃H₁₉ClO₃, 72.92; H, 5.06; Cl, 9.36; O, 12.67. Found C, 72.90; H, 5.08; Cl, 9.26; O, 12.77

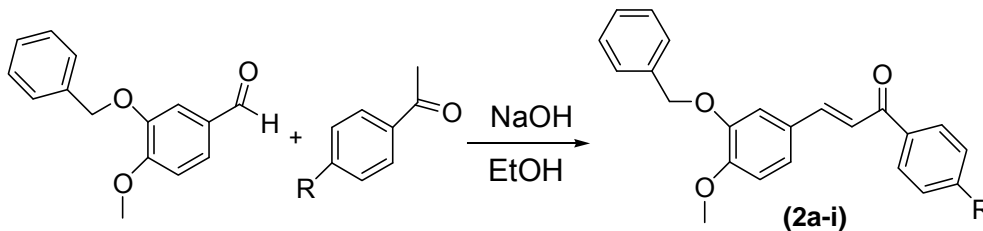
(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-(4-methoxyphenyl)prop-2-en-1-one (2c)

Yellowish crystals, mp 135–140°C. ¹H NMR(300MHZ, DMSO) α=8,18(d, J=8,5, 2H), 7,94(d, J=8,5, 2H), 7,80(d, J=15,5, 1H), 7,77(d, J=15,5, 1H), 7,75(d, J=1,2, 1H), 7,62(d, J=8,5, 2H), 7,59(d, J=8,4, 1H), 7,52(d, J=8,4, 2H), 7,30-7,41(m, 2H), 7,01(d,

J=8,4, 1H), 5,28(s, 2H), 3,77 (s, 6H) ¹³C NMR(300MHZ, DMSO) α=189,71, 149,5, 148, 144,2,142,2, 141,1, 135, 130,3(2C), 128,4(2C), 128(2C), 127,7, 127,3, 126(2C), 120,4, 119, 114,2, 112,6, 70,2, 56,2, 54,2. IR (KBr): 3112, 3100, 3049, 2779, 1713, 1622, 1515, 1460, 1328, 1159, 1032, 806, 735, 703 Anal. calcd. For C₂₄H₂₂O₄, 376.99; H, 5.92; O, 17.09 Found C, 76.89; H, 5.90; O, 17.21.

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-p-tolyprop-2-en-1-one (2d)

Yellowish crystals, mp 143–150°C. ¹H NMR(300MHZ, DMSO) α=8,12(d, J=8,5, 2H), 7,90(d, J=8,5, 2H), 7,80(d, J=15,5, 1H), 7,76(d, J=15,5, 1H), 7,73(d, J=1,2, 1H), 7,65(d, J=8,5, 2H), 7,61(d, J=8,4,



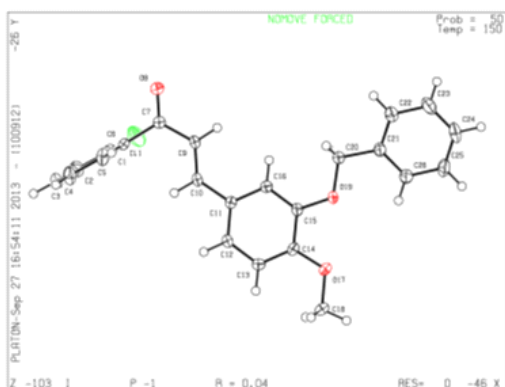
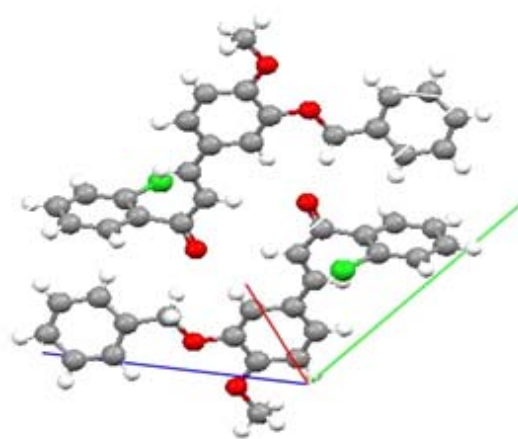
Scheme 1

Table 3: Antibacterial test of the synthesized compounds by disc diffusion method against tested strains

Compound No	Inhibition zone (diameter) mm of synthesized compound				
	<i>Escherichia Coli</i>	<i>Pseudomonas Aeruginosa</i>	<i>Klebsiella Pneumoniae</i>	<i>Proteus Mirabilis</i>	<i>Staphylococcus aureus</i>
2a	12	14	14	12	12
2b	14	20	16	-	10
2c	20	18	20	14	18
2d	18	16	22	10	12
2e	20	16	14	12	18
2f	12	10	14	18	20
2g	10	12	-	12	10
2h	18	-	16	-	16
2i	20	14	18	18	16

Table 4: Minimal inhibitory concentration (MIC) in $\mu\text{g}\cdot\text{ml}^{-1}$ of synthesized compounds against tested strains

Compound No	Minimal inhibitory concentration (MIC) in $\mu\text{g}\cdot\text{ml}^{-1}$				
	<i>Escherichia Coli</i>	<i>Pseudomonas Aeruginosa</i>	<i>Klebsiella Pneumoniae</i>	<i>Proteus Mirabilis</i>	<i>Staphylococcus aureus</i>
2a	17	14	14	17	12
2b	14	17	16	-	10
2c	19	18	19	14	16
2d	17	16	21	15	15
2e	17	16	14	12	15
2f	12	10	14	18	17
2g	11	10	-	11	11
2h	18	-	16	-	16
2i	20	18	19	18	17

**Fig. 1.** The ORTEP diagrams and the molecular structure of the compound 2i with atom labels**Fig. 2:** Packing of the molecules when viewed down along crystallographic 'c' direction

1H), 7,502(*d*, *J*=8,4, 2H), 7,33-7,41(*m*, 2H), 7,12(*d*, *J*=8,4, 1H), 5,13(*s*, 2H), 3,81(*s*,3H), 2,35(*s*,3H), ¹³C NMR(300MHZ, DMSO) α =189,61, 149,61, 148, 145,22,142,2, 140,15, 136, 131,3(2C), 129,11(2C), 129(2C), 128,51, 127,77, 127,21(2C), 121,11, 119,17, 115,12, 110,6, 71,2, 56,2, 24,55. IR (KBr): 3120, 3102, 3009, 2719, 1603, 1592, 1555, 1487, 1338, 1110,1054, 819, 735, 703 Anal. calcd. For C₂₄H₂₂O₃C, 80.42; H, 6.19; O, 13.39. Found C, 80.32; H, 6.24; O, 13.44

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-(4-hydroxyphenyl)prop-2-en-1-one (2e)

White solid 146–150°C ¹H NMR(300MHZ, DMSO) α =7,96(*d*, *J*=8,5, 2H), 7,90(*d*, *J*=8,5, 2H), 7,81(*d*, *J*=15,5, 1H), 7,74(*d*, *J*=15,5, 1H), 7,72(*d*, *J*=1,2, 1H), 7,68(*d*, *J*=8,5, 2H), 7,60(*d*, *J*=8,4, 1H), 7,52(*d*, *J*=8,4, 2H), 7,31-7,40(*m*, 2H), 7,18(*d*, *J*=8,4, 1H), 5,10(*s*, 2H), 5,02(*s*,1H), 3,35(*s*,3H), ¹³C NMR(300MHZ, DMSO) α =189,61, 149,61, 148, 145,22, 142,2, 140,15, 136, 131,3(2C), 129,11(2C), 129(2C), 128,57, 127,54, 127,10(2C), 122,11, 118,14, 115,12, 111,6, 71,2, 56,2. IR (KBr): 3572,31053130, 3033, 2111, 1653, 1612, 1515, 1411, 1342, 1122, 832, 722, 711 Anal. calcd. For C₂₃H₂₀O₄C, 76.65; H, 5.59; O, 17.76. Found C, 76.35; H, 5.79; O, 17.86.

(E)-1-(4-aminophenyl)-3-(3-(benzyloxy)-4-methoxyphenyl)prop-2-en-1-one (2f)

White solid, mp 152–158°C ¹H NMR(300MHZ, DMSO) α =8,01(*d*, *J*=8,5, 2H), 7,92(*d*, *J*=8,5, 2H), 7,72(*d*, *J*=15,5, 1H), 7,70(*d*, *J*=1,2, 1H), 7,64(*d*, *J*=8,5, 2H), 7,60(*d*, *J*=8,4, 1H), 7,50(*d*, *J*=8,4, 2H), 7,33-7,45(*m*, 2H), 7,20(*d*, *J*=8,4, 1H), 5,11(*s*, 2H), 4,02(*s*,2H), 3,35(*s*,3H), ¹³C NMR(300MHZ, DMSO) α =187,61, 147,61, 146,11, 144,22, 142,21, 138,15, 136,12, 131,30(2C), 129,16(2C), 128,60(2C), 128,17, 127,54(2C), 121,29, 122,11, 119,14, 115,12, 110,6, 71,2, 56,2. IR (KBr): 31013052, 3019, 2519, 1673,1661, 1642, 1525, 1437, 1343, 1152, 810, 715, 702 Anal. calcd. For C₂₃H₂₁NO₃C, 76.86; H, 5.89; N, 3.90; O, 13.35. Found C, 76.76; H, 5.99; N, 3.75; O, 13.50.

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-(4-bromophenyl)prop-2-en-1-one (2g)

Yellowish crystals, mp 155–157°C ¹H NMR(300MHZ, DMSO) α =8,16(*d*, *J*=8,5, 2H), 7,92(*d*, *J*=8,5, 2H), 7,84(*d*, *J*=15,5, 1H), 7,72(*d*,

J=15,5, 1H), 7,71(*d*, *J*=1,2, 1H), 7,66(*d*, *J*=8,5, 2H), 7,61(*d*, *J*=8,4, 1H), 7,50(*d*, *J*=8,4, 2H), 7,35-7,45(*m*, 2H), 7,08(*d*, *J*=8,4, 1H), 5,20(*s*, 2H), 3,81(*s*,3H) ¹³C NMR(300MHZ, DMSO) α =189,71, 149,7, 149, 145,2,141,2, 140,1, 136, 131,3(2C), 129,4(2C),129(2C), 128,5, 127,7, 127,10(2C), 121,44, 119,7, 115,2, 110,6, 70,9, 55,90. IR (KBr): 31203122, 3033, 2822, 1677, 1632, 1513, 1451, 1360, 1159, 801, 745, 714,665 Anal. calcd. For C₂₃H₁₉BrO₃C, 65.26; H, 4.52; Br, 18.88; O, 11.34. Found C, 65.06; H, 4.62; Br, 18.90; O, 11.42.

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-(4-nitrophenyl)prop-2-en-1-one (2h)

White solid, mp 152–156°C ¹H NMR (300MHZ, DMSO) α =8,22(*d*, *J*=8,5, 2H), 7,97(*d*, *J*=8,5, 2H), 7,88(*d*, *J*=15,5, 1H), 7,74(*d*, *J*=15,5, 1H), 7,72(*d*, *J*=1,2, 1H), 7,69(*d*, *J*=8,5, 2H), 7,64(*d*, *J*=8,4, 1H), 7,52(*d*, *J*=8,4, 2H), 7,34-7,46(*m*, 2H), 7,12(*d*, *J*=8,4, 1H), 5,22(*s*, 2H), 3,81(*s*,3H) ¹³C NMR(300MHZ, DMSO) α =190,11, 148,7, 149, 145,71,142,2, 140,16, 136, 131,33(2C), 129,48(2C),129(2C), 128,59, 127,77, 127,22, 121,44, 119,7, 111,24, 74,66, 55,96. IR (KBr): 3111,3101, 2919, 2713, 1643, 1645, 1563, 1412, 1368, 1334,1110, 801, 712, 701. Anal. calcd. For C₂₃H₁₉NO₅C, 70.94; H, 4.92; N, 3.60; O, 20.54 Found. C, 70.54; H, 5.02; N, 3.70; O, 20.74.

(E)-3-(3-(benzyloxy)-4-methoxyphenyl)-1-(2-chlorophenyl)prop-2-en-1-one (2i)

Yellowish crystals, mp 160–165°C ¹H NMR (300MHZ, DMSO) α =7,80(*d*, *J*=15,5, 1H), 7,46-7,75 (*m*, 4H), 7,56(*d*, *J*=15,5, 1H), 7,19(*m*, 5H),6.61-6.75(*m*, 3H), 5.20(*s*, 2H), 3,81(*s*,3H) ¹³C NMR(300MHZ, DMSO) α =190,11,149,7, 149, 145,11, 141,12, 137,3, 136,4, 134,2, 131,3, 129,4, 129(2C),128,5, 127,7, 127,4, 127,2(2C),120, 115, 111,5, 71,2, 56,4 IR (KBr): 3111,3101, 2901, 2821, 1693, 1652, 1531,1512, 1411, 1345, 1144, 801, 715, 701 Anal. calcd. for C₂₃H₁₉ClO₃C, 72.72; H, 5.16; Cl, 9.41; O, 12.72 Found C, 72.90; H, 5.08; Cl, 9.38; O, 12.69.

Antibacterial bioassay

Derivatives **2 (a-i)** were tested for in vitro anti-microbial activity against five different bacterial species (Gram negative and Gram positive) namely: *Staphylococcus aureus* ATCC, *Klebsiella pneumonia* ATCC, *Escherichia coli* ATCC, *Pseudomonas*

aeruginosa ATCC, *Proteus mirabilis* ATCC using: the diffusion method and the minimum inhibitory concentration (MIC).

The diffusion méthode (methode of the disk)

Each disk contain 100mg of the test compound for this method Muller Hinton agar was melted at 100C° and after cooling to 56 C° was poured into Petri plates of 9cm diameter in quantities of 18 ml, left on the flat surface to solidify and the surface of the medium was dried at 37C°, then the culture of each bacteria and yeast strain after being kept in Mueller –Hinton broth to 10⁻⁵ cfu ml⁻¹ were pipetted into the Mueller-Hinton agar plate prepared as described above, the surface of the medium was allowed to dry . The 10 mg ml⁻¹ in DMSO compound impregnted discs were applied to the surface of incubated plates. The Petri plates were placed in an incubator at 37C° after 18h of incubation the Petri plates were examined and it was found that all the test compounds exhibited different degrees of antibacterial activity or inhibitory action (table3).

The minimum inhibitory concentration (MIC)

The MIC of these compounds was determined by the micro-broth dilution technique using Muller-Hinton Broth. Serial tow –fold dilution ranged from 2500 to 2.4 ¼g⁻¹ for compounds.

The inoculums was prepared in broth which had been kept overnight at 37C° and which had been diluted with Muller –Hinton Broth to give a final concentration of 10⁻⁵ mg.ml⁻¹ in the test tray. The trays were covred and placed in plastique bags to prevent drying after incubation at 37C° for 18-24h. The MIC was defined as the lowest concentration of compound giving complet inhibition of visible growth (table4).

Antibacterial evaluation

The antibacterial evaluation data for compounds (2a-i) is presented in Table-3 and Table-4. The zone of inhibition was measured in mm,

Minimal inhibitory activity was observed for 2500 ¼Àg/mL to 2.4 ¼Àg⁻¹ and compounds showed their effect in a dose dependant manner. The antibacterial activity of the different compound is moderate, good and excellent

From table-3, it is observed that compound 2i with 2-Chloro substitution, compound 2c with 4-methoxy substitution and 2d with 4-Methyl, exhibited excellent antibacterial activity against the gram positive and gram negative bacterial species. All the other tested compounds exhibited different degrees of antibacterial activities, and the inhibition actions were between 10 to 18mm. The moderate antibacterial activity was recorded for the compound 2g with all bacterial tested species.

CONCLUSION

Nine novel chalcone derivatives (2a–i) were synthesized by the Claisen-Schmidt condensation the structural confirmation of these derivatives (**2a-i**) was accomplished by spectroscopic techniques, including 1H NMR, 13C NMR, IR and elemental analyses.

The antibacterial activitie of the synthesized compound (**2a- i**) were carried out by well diffusion and MIC method. The obtained results proved that the synthesized chalcones analogues have diffrent antimicrobial effects against all the bacterial specis, and some product (**2c**, **2d**, **2i**) exhibited excellent antibacterial activity against the gram positive and gram negative bacterial species.

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