



## Several pathways of hydrogen peroxide action that damage the *E. coli* genome

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### Abstract

Hydrogen peroxide is an important reactive oxygen species (ROS) that arises either during the aerobic respiration process or as a by-product of water radiolysis after exposure to ionizing radiation. The reaction of hydrogen peroxide with transition metals imposes on cells an oxidative stress condition that can result in damage to cell components such as proteins, lipids and principally to DNA, leading to mutagenesis and cell death. *Escherichia coli* cells are able to deal with these adverse events via DNA repair mechanisms, which enable them to recover their genome integrity. These include base excision repair (BER), nucleotide excision repair (NER) and recombinational repair. Other important defense mechanisms present in *Escherichia coli* are OxyR and SosRS anti-oxidant inducible pathways, which are elicited by cells to avoid the introduction of oxidative lesions by hydrogen peroxide. This review summarizes the phenomena of lethal synergism between UV irradiation (254 nm) and H<sub>2</sub>O<sub>2</sub>, the cross-adaptive response between different classes of genotoxic agents and hydrogen peroxide, and the role of copper ions in the lethal response to H<sub>2</sub>O<sub>2</sub> under low-iron conditions.

**Key words:** hydrogen peroxide, cross-adaptive response, lethal synergism, copper and iron.

Received: February 13, 2004; Accepted: February 19, 2004.

### General Aspects

The appearance of aerobic forms of life was an important step in the evolutionary process, since oxygen consumption leads to the production of ten-fold more energy from glucose than does anaerobic metabolism (Meneghini, 1987). However, this process imposes constraints on cell viability, because of the generation of reactive oxygen species during respiration.

The consecutive univalent reduction of molecular oxygen to water produces three active intermediates: superoxide anion (O<sub>2</sub><sup>-</sup>), hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and hydroxyl radical (OH<sup>•</sup>). These intermediates, collectively referred to as reactive oxygen species (ROS) are potent oxidants of lipids, proteins, and nucleic acids (Halliwell and Gutteridge, 1984; Mello-Filho and Meneghini, 1985; Me-

neghini, 1988). Among the oxidative DNA lesions, one of the major classes of DNA damage leads to modification in purine and pyrimidine bases, together with oligonucleotide strand breaks, DNA-protein cross-links and abasic sites. Increasing evidence suggests that the cumulative damage caused by ROS contributes to numerous degenerative diseases associated with aging, such as atherosclerosis, rheumatoid arthritis and cancer (Ames *et al.*, 1993; Halliwell and Gutteridge, 1999).

Living organisms have developed specific mechanisms to prevent the production and effects of ROS. The reduction of O<sub>2</sub> by cytochrome oxidase without yielding ROS, the superoxide dismutase catalysis of O<sub>2</sub><sup>-</sup> into H<sub>2</sub>O<sub>2</sub> through a dismutation reaction, the decomposition of H<sub>2</sub>O<sub>2</sub> by catalase and peroxidases, and the scavenging of ROS by some vitamins comprise part of the set of cellular antioxidant defenses (Halliwell and Gutteridge, 1999). Synthesis of the enzymes that catalyze these reactions is a part of the adaptive response triggered by the stress posed by ROS.

Despite these cellular defenses, many lesions can be produced in cellular targets, mainly in the DNA molecule, leading to mutagenesis and cell death. However, cells are able to deal with these adverse events via DNA repair mechanisms, which enable them to recover their integrity. These include base excision repair (BER), nucleotide excision repair (NER) and recombinational repair (Miles and Sancar, 1989).

Several studies indicate that the killing of *E. coli* cells exposed to H<sub>2</sub>O<sub>2</sub> is mainly due to damage to DNA (Imlay and Linn, 1986; Imlay and Linn, 1988; Hagensee and Moses, 1989; Asad and Leitão, 1991) and a wide variety of DNA lesions are formed (Halliwell and Gutteridge, 1999) as a by-product of the H<sub>2</sub>O<sub>2</sub> reaction. Some of these lesions are miscoding acting as an important source of mutagenesis in aerobically growing cells.

Participation of OH<sup>•</sup> as the main damaging agent has been suggested by studies using scavengers of OH<sup>•</sup>. Repine *et al.* (1981) and Brakely *et al.* (1990) have demonstrated that dimethyl sulfoxide partially inhibits DNA base damage by H<sub>2</sub>O<sub>2</sub>, and Brandi *et al.* (1989) have noted that *E. coli* bacteria are partially protected against the lethal effects of H<sub>2</sub>O<sub>2</sub> by pretreatment with ethanol, dimethyl sulfoxide, or thiourea.

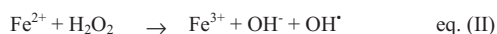
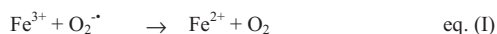
The induction of DNA damage by H<sub>2</sub>O<sub>2</sub> in *E. coli* as well as in mammalian cells can be either impaired or enhanced by the presence of transition metal ion chelators (Mello-Filho and Meneghini, 1985; Asad and Leitão, 1991). In addition, H<sub>2</sub>O<sub>2</sub> can cause membrane lesions through lipid peroxidation and, by promoting alterations in several amino acids can lead to the inactivation of enzymes (Farr and Kogona 1991).

Recent reviews on the mechanisms of oxidative DNA damage and repair have focussed on prevention and repair (Cooke *et al.*, 2003), biochemical features (Cadet *et al.*, 2003), substrate specificities for glycosylases (Dizdaroglu, 2003), biological consequences (Wallace, 2002; Bjelland and Seeberg, 2003) and the role of iron (Kruszewski, 2003).

In this review, we discuss several aspects of DNA damage and cellular inactivation induced by hydrogen peroxide; the importance of transition metals such as iron and copper in this context; DNA repair pathways involved in the cellular response to H<sub>2</sub>O<sub>2</sub>, and related antioxidant cell defenses.

## H<sub>2</sub>O<sub>2</sub> and transition metals

H<sub>2</sub>O<sub>2</sub> reacts with O<sub>2</sub><sup>•-</sup> resulting in OH<sup>•</sup> production through the so-called Haber-Weis cycle. However, this reaction depends on the presence of transition metals such as Cu<sup>+</sup> and/or Fe<sup>2+</sup>, which work as reducing agents, according to the reactions below (Kehrer, 2000).



H<sub>2</sub>O<sub>2</sub>, per se, is considered a weak oxidant agent. Nevertheless, it easily crosses the cellular membrane and reacts with transition metals, generating OH<sup>•</sup>. Evidence for the importance of transition metals comes from studies with chelators such as dipyrindyl, phenanthroline and desferrioxamine (Fe chelators) and neocuproine (a Cu chelator) (Brandi *et al.*, 1989; Brakely *et al.*, 1990; Hallywell and Gutteridge, 1999). Pretreatment with iron chelators protects both prokaryotic and eukaryotic cells against the lethal effects of H<sub>2</sub>O<sub>2</sub> (Mello-Filho and Meneghini, 1985; Imlay and Linn, 1988; Asad and Leitão, 1991). Additionally, cultures of *Staphylococcus aureus* are more sensitive to H<sub>2</sub>O<sub>2</sub> when iron is added to the culture media (Repine *et al.*, 1981) and in *E. coli* cultures the same is observed (Touati *et al.*, 1995).

In *E. coli* the metabolism of iron is involved in the cellular antioxidant response. A transcription factor denoted Fur, the global repressor of ferric ion uptake, regulates about 30 genes implicated in iron uptake from the environment (Braun, 1997; Braun *et al.*, 1998). Most Fur-regulated genes are derepressed in low iron concentrations and repressed when a high concentration of iron is present (Hantke, 1981; Hantke, 2002). The finding that *fur* mutants are sensitive to H<sub>2</sub>O<sub>2</sub> and that they suffer an increase in oxidative DNA damage leading to mutations under aerobic conditions supports the hypothesis that Fur has a role in the defense against oxidative stress (Touati *et al.*, 1995). Furthermore, the regulators of *E. coli* responses to oxidative stress, OxyR and SoxRS, activate the expression of Fur and suggest that control of iron metabolism in *E. coli* is an integral part of the antioxidant response (Zheng *et al.*, 1999)

## DNA Repair Pathways and Antioxidant Defense Systems

### Base excision repair (BER)

Lesions produced by H<sub>2</sub>O<sub>2</sub> are typically repaired by base excision repair (BER) mechanisms. This kind of DNA repair is initiated by DNA glycosylases, enzymes that recognize the modified bases and act by cleaving the glycosylic bond, thereby removing the damaged base from the sugar phosphate backbone and, as a result, producing an apurinic/apyrimidinic (AP) site (Sancar and Sancar, 1988).

Some DNA glycosylases also display a class I AP lyase activity that incises the phosphodiester linkage on the 3' side of the AP lesion and generates a 5'-phosphate group and a 3'-terminus that needs removal by a class II AP endonuclease/3'-diesterase prior to repair synthesis and ligation (Piersen *et al.*, 2000). The cleavage of AP sites can also be catalyzed by class II AP endonucleases which incise the 5' side of the AP site, leaving a 3'-OH terminus and a 5'-abasic residue that is removed by a deoxyribosephosphodiesterase (dRPase) (Mol *et al.*, 2000).

In *E. coli* two enzymes are representative of the class I AP lyases, the Fpg/MutM DNA glycosylase, product of the

*fpg/mutM* gene, which recognizes primarily oxidized purines, and the DNA endonuclease III, product of the *nth* gene, which recognizes primarily oxidized pyrimidines. Interestingly, the *E. coli* repair-deficient *fpg/mutM*, as well as the *nth* mutants, are not more sensitive to killing by H<sub>2</sub>O<sub>2</sub> than wild-type cells (Cunningham and Weiss, 1985; Boiteux and Huisman, 1989). Subsequently, endonuclease VIII and IX were purified from endonuclease III-deficient *E. coli* cells (Wallace *et al.*, 1988; Melamede *et al.*, 1994). Endonuclease VIII was found to exhibit a thymine glycol DNA glycosylase activity as well as an AP lyase activity. Endonuclease IX recognizes urea residues and  $\beta$ -ureidobutyric acid in DNA; however, DNA containing thymine glycol or dihydrothymine is not a substrate for this enzyme (Friedberg *et al.*, 1995). Later, the gene for endonuclease VIII (*nei*) was isolated (Jiang *et al.*, 1997; Saito *et al.*, 1997). Endonuclease VIII is present in extracts of *E. coli* at 5 to 10% of the level of endonuclease III and is responsible for the repair of 10% of the thymine glycol in *E. coli* (Wallace *et al.*, 1988; Wallace, 1988).

The *nth nei* double mutants are hypersensitive to H<sub>2</sub>O<sub>2</sub> (Jiang *et al.*, 1997). Moreover, endonuclease VIII can recognize 8-oxo-7,8-dihydroxyguanine (8-oxoG) lesions, a kind of lesion also recognized by Fpg protein (Blaisdell *et al.*, 1999). The multiplicity of DNA glycosylases that recognize and attack sites of oxidative damage in DNA confirms the importance of this form of base damage.

Class II AP endonucleases of *E. coli* are mainly represented by the exonuclease III, the *xthA* gene product, and endonuclease IV, the *nfo* gene product.

The role of exonuclease III in repairing oxidative damage was highlighted by the demonstration that *xth* mutants are extremely sensitive to H<sub>2</sub>O<sub>2</sub> (Demple *et al.*, 1983). On the other hand, *nfo* mutants are not sensitive to H<sub>2</sub>O<sub>2</sub> (Cunningham *et al.*, 1986), and deletion of *nfo* increases the killing of *xth* mutants to H<sub>2</sub>O<sub>2</sub>, indicating that many of the repair activities of exonuclease III and endonuclease IV overlap.

Endonuclease IV normally represents about 10% of the total endonuclease activity; superoxide-generating agents induce a 10- to 20-fold increase in the level of this enzyme through the *soxRS* response (Chan and Weiss, 1987).

#### Nucleotide excision repair (NER) in response to H<sub>2</sub>O<sub>2</sub>

In *E. coli*, a complex of proteins encoded by the *uvrA*, *uvrB* and *uvrC* genes is required for lesion recognition and the dual incisions. This complex eliminates DNA lesions that cause significant distortions in the phosphodiester backbone of the molecule (Friedberg *et al.*, 1995; Hanawalt, 2001).

This kind of repair is of fundamental importance for the correction of UV (254 nm) lesions, mainly cyclobutane pyrimidine dimers. It appears not to be related to the repair

of H<sub>2</sub>O<sub>2</sub> lesions, since the *uvrABC* mutants are not sensitive to H<sub>2</sub>O<sub>2</sub> (Imley and Linn, 1987).

The finding that the triple mutant *uvrA nfo xthA* cannot be constructed, despite the fact that the double mutants are viable (Saporito *et al.*, 1989), has raised the question of the connection between BER and NER DNA repair pathways. Additionally, the UvrABC complex is able to remove AP sites generated by oxidative lesions in the DNA molecule (Lin and Sancar, 1989). So, these results suggest that both repair systems correct the lesions produced by oxidant agents produced during cellular respiration.

These findings suggest that some intermediate products of base excision repair may be substrates for the UvrABC complex. This hypothesis was confirmed by Kow *et al.* (1990), who demonstrated the role of this complex in the repair of thymine glycols in the DNA of replicative form of phage  $\Phi$ X174. The survival of phage containing thymine glycols is lower in the *nth uvrA* double mutant than in the *nth* single mutant.

#### SOS response

The best-studied transcriptional response to DNA damage is the SOS response (Friedberg *et al.*, 1995; Walker, 1996). Single-stranded DNA produced by several DNA-damaging agents and repair mechanisms can be bound by RecA protein, resulting in conversion of this protein to its activated form. Once activated, RecA interacts with LexA protein, the repressor of the SOS genes (Wagner *et al.*, 1999). This interaction triggers the autocatalytic cleavage of LexA and consequent destruction of its ability to function as a repressor, which results in the derepression of SOS genes (Mustard and Little, 2000; Fernandez De Henestrosa *et al.*, 2000). By using DNA microarray techniques Courcelle *et al.* (2001) have shown that in *E. coli* the expression of 43 genes is controlled by LexA.

The expression of LexA-controlled genes allows the increased phenotypic expression of mutagenesis (*umuDC* genes), nucleotide excision repair (*uvrA* and *uvrB* genes), genetic recombination (*recA*, *recN*, *recQ* and *recD* genes), cellular filamentation (*sulA* and *sulB* genes), and survival and mutagenesis of irradiated phages (W-reactivation and W-mutagenesis) (Friedberg *et al.*, 1995).

A low concentration of H<sub>2</sub>O<sub>2</sub> (1-3 mM) results in SOS gene induction in wild-type cells (Imley and Linn, 1987; Goerlich *et al.*, 1989). However, H<sub>2</sub>O<sub>2</sub> can induce some SOS responses without SOS induction. An example is the cell filamentation induced by H<sub>2</sub>O<sub>2</sub> in *sulA* and *recA* mutants and the mutagenesis that occurs at the same level in wild-type and *umuC* mutant cells treated with H<sub>2</sub>O<sub>2</sub> (Imley and Linn, 1987). On the other hand, the induction of SOS by H<sub>2</sub>O<sub>2</sub> is an important event, since *recA* and *recBC* mutant cells are very sensitive to H<sub>2</sub>O<sub>2</sub> treatment probably due to the lack of recombinational repair necessary for the repair of H<sub>2</sub>O<sub>2</sub>-induced lesions (Imley and Linn, 1987). Additionally, Konola *et al.*, (2000) have shown that the *ruvA*



mutants are 10- to 15-fold more sensitive to H<sub>2</sub>O<sub>2</sub> (1-3 mM) than the wild-type cells. Together with RuvB, the RuvA protein generates the RuvAB complex, which stimulates strand migration in the Holliday junctions (West, 1996).

#### Other inducible events: anti-oxidant and cross-adaptive responses triggered by H<sub>2</sub>O<sub>2</sub>

Most genes encoding DNA repair enzymes that act on oxidative damage appear to be expressed constitutively in actively growing cells. This is presumably because oxidative DNA damage is continuously produced by ROS, which are normal by-products of aerobic metabolism (Demple and Harrison, 1994; Henle and Linn, 1997). However, in order to deal with elevated levels of peroxide in the environment, cells have evolved changes in metabolism that help to protect DNA from ROS.

Most inducible genes that respond to oxidative damage prevent, rather than repair DNA damage. However, a notable exception is endonuclease IV, the *nfo* gene product, an AP endonuclease that repairs 3' phosphate residues to 3'OH groups that can prime DNA synthesis (Chaudhry *et al.*, 1999; Izumi *et al.*, 2000).

Two key protective responses have been described in *E. coli* - one controlled by *soxRS* genes and the other by *oxyR* (Tsaneva and Weiss, 1990; Storz and Imlay, 1999; Gonzalez-Flecha and Demple, 2000).

#### The SoxRS

Low concentrations of superoxide-generating compounds such as paraquat and menadione render the cells resistant to higher doses of these agents (Geenberg and Demple, 1989) in a manner dependent on the integrity of the *soxRS* locus.

The *soxRS* regulatory system acts in two steps, with SoxR serving both as a sensor and as an activator protein. When activated by the univalent oxidation of the 2Fe-2S clusters of the protein through a not yet explained mechanism (Storz and Imlay, 1999) SoxR induces transcription of *soxS*, a positive regulator that stimulates transcription of more than 16 other superoxide responsive genes (Wu and Weiss, 1992; Hidalgo *et al.*, 1995). Although this system responds to oxidative stress when cells are exposed to superoxide radical-generating agents, it is not induced by H<sub>2</sub>O<sub>2</sub> (Chan and Weiss, 1987; Tsaneva and Weiss, 1990; Hidalgo *et al.*, 1997). However, it was demonstrated that in some conditions H<sub>2</sub>O<sub>2</sub> as well as singlet molecular oxygen could activate the SoxRS regulon *in vivo* (Manchado *et al.*, 2000; Agnez-Lima *et al.*, 2001).

The products of the induced *soxRS* regulon include: Mn-superoxide dismutase (*sodA*), DNA repair endonuclease IV (*nfo*), glucose-6-phosphate dehydrogenase (*zwf*), aconitase (*acnA*), stable fumarase (*fumC*), ferredoxin reductase (*fpr*), toxin and antibiotic efflux pumps (*acrAB*), an anti-sense RNA for the ompF porin mRNA (*micF*) and an iron-binding repressor of iron transport (*fur*) (Amabile-

Cuevas and Demple, 1991; Gaudu and Weiss, 1996; Gaudu *et al.*, 1997; Pomposiello and Demple, 2001).

#### Regulation of the OxyRS response to oxidative damage

Bacterial cells possess an adaptive response to oxidizing agents, which means that exposure to low levels of H<sub>2</sub>O<sub>2</sub> allows bacterial cells to survive further toxic doses of H<sub>2</sub>O<sub>2</sub> (Demple and Halbrook, 1983; Demple, 1991). The expression of nine proteins induced by H<sub>2</sub>O<sub>2</sub> treatment is under the control of the *oxyR* gene (Christman *et al.*, 1985). Several proteins whose expression is regulated by *oxyR* have been identified, including catalase and an alkyl hydroperoxide reductase (Morgan *et al.*, 1986; Storz *et al.*, 1990).

The level of OxyR protein does not change with H<sub>2</sub>O<sub>2</sub> treatment, indicating that it is activated post-translationally (Storz *et al.*, 1990). H<sub>2</sub>O<sub>2</sub> activates the transcriptional activity of OxyR by oxidizing two of its cysteine residues (Zheng *et al.*, 1998; Aslung *et al.*, 1999; Storz and Tolodano, 1999). When activated, OxyR activates transcription of genes that include *katG* (catalase hydroperoxidase I), *ahpCF* (alkylhydroperoxide-NADPH oxidoreductase), *grxA* (glutaredoxin), *gorA* (glutathione reductase), *dps* (a protein that protects DNA from peroxide damage) (Altuvia *et al.*, 1994; Martinez and Kolter, 1997) and *fur* (an iron-binding repressor of iron transport). Under oxidative-stress conditions with the influence of OxyR and SoxRS, the number of Fur molecules per cell increases from 5,000 to about 10,000 (Zheng *et al.*, 1999).

In addition, OxyR activates the synthesis of *oxyS* (Altuvia *et al.*, 1997; Zhang *et al.*, 1997), which encodes an untranslated mRNA that appears to regulate as many as 20 additional genes, possibly by an antisense mechanism (Altuvia *et al.*, 1997; Argaman and Altuvia, 2000). The *oxyS* gene product is independent of OxyR in limiting the endogenous production of H<sub>2</sub>O<sub>2</sub> in *E. coli* (Gonzalez-Flecha and Demple 1999).

Through random transcriptional fusions some new genes, the expression of which require OxyR, have been detected. These include *henF* (coproporphyrinogen III oxidase), which participates in the synthesis of photoheme IX, which is required for activity of both HPI (*katG*) and HPII (Mukhopadhyay and Schellhorn, 1997). DNA microarray techniques have detected several other new OxyR-activated genes including the *henH* heme biosynthetic gene; the six-gene *suf* operon, which may participate in Fe-S cluster assembly of repair; and four genes of unknown function (Zheng *et al.*, 2001).

#### Cross-Adaptive Responses

Cross-adaptive response occurs when cells exposed to doses of a sub-lethal agent develop resistance against challenging doses of another lethal agent. For instance, it is

well known that *E. coli* cells exposed to low doses of H<sub>2</sub>O<sub>2</sub> develop resistance against heat shock, ethanol (Jenkins *et al.*, 1988), ultraviolet A (UVA) (Tyrrell, 1985), formaldehyde (Nunoshiba *et al.*, 1991), menadione and cumene hydroperoxide (Christman *et al.*, 1985). In contrast, prior exposure of *E. coli* to low doses of H<sub>2</sub>O<sub>2</sub> has shown little or no effect on the resistance of these cells to UVC or alkylating agents (Demple and Halbrook, 1983). This section is meant to provide a general view into the cross-adaptive response induced by H<sub>2</sub>O<sub>2</sub> against effects of UV (254 nm), N-methyl-N'-nitro-N-nitrosoguanidine (MNNG) and cumene hydroperoxide.

#### Cross-adaptive response between H<sub>2</sub>O<sub>2</sub> and UV (254 nm)

Demple and Halbrook (1983) have shown that prior exposure of *E. coli* to low doses of H<sub>2</sub>O<sub>2</sub>, in the micromolar range, engenders little or no effect on the resistance of these cells to UV. However, pretreatment with 2.5 mM H<sub>2</sub>O<sub>2</sub> protected wild-type cells against UV-irradiation. This protection is independent of the SOS response, since it is also observed in a *lexA* mutant, and is dependent on DNA excision repair, since this protection is not observed in an *uvrA* mutant (Asad *et al.*, 1994). The possibility that DNA repair activities other than those known to be involved in SOS response could be induced during H<sub>2</sub>O<sub>2</sub> pretreatment, was verified by examining the ability of H<sub>2</sub>O<sub>2</sub>-pretreated cells to repair incoming damaged DNA. In this way, experiments similar to those described by Weigle (1953), but involving treatment of the cells with 2.5 mM H<sub>2</sub>O<sub>2</sub> instead of UV, were carried out with UV-irradiated phages. An enhanced survival of UV-irradiated phages in the wild-type, H<sub>2</sub>O<sub>2</sub>-pretreated cells has been shown. The same results were obtained with the *lexA1* mutant, indicating that the response is not dependent on the SOS system; nevertheless RecA and UvrA proteins are involved, since there is no UV-damaged phage reactivation in *recA* and *uvrA* mutant cells pretreated with 2.5 mM H<sub>2</sub>O<sub>2</sub>. The *oxyΔ3* mutant, in which the adaptive response to H<sub>2</sub>O<sub>2</sub> does not occur (Imlay and Linn, 1987), presented similar results to those observed in wild-type and the *lexA1* mutant, indicating that the *oxyR* regulon does not participate in the UV-damaged phage repair which takes place in H<sub>2</sub>O<sub>2</sub>-treated cells.

The induction of some proteins after UV irradiation in *lexA* (Def) mutants has been observed (Lesca *et al.*, 1991), and these proteins may act in mutagenesis and in the DNA repair of UV-irradiated bacteriophage lambda (Calsou *et al.*, 1987). In fact, Petit *et al.* (1993) have characterized an *E. coli* gene, *dinY*, whose induction does not require the cleavage of LexA repressor, although it may be considered a member of the SOS regulon (Petit *et al.*, 1993; Friedberg *et al.*, 1995). The protective effect induced by H<sub>2</sub>O<sub>2</sub> against UV is shown to be independent of the induction of *dinY* gene or other genes under the same control as *dinY* (Asad *et al.*, 2000). Besides, this cross-protection re-

sponse is not induced when the SOS regulon is constitutively expressed. The most consistent view of these observations seems to be that the induction of this response may be related to the induction of genes that are not under the control of LexA, and are inhibited by the expression of SOS genes (Asad *et al.*, 2000).

#### Cross-adaptive response between H<sub>2</sub>O<sub>2</sub> and MNNG

The Ada and Ogt proteins are involved in the DNA repair of some lesions caused by alkylating agents, as O<sup>6</sup>-methylguanine and O<sup>4</sup>-methylthymine. Moreover, Ada induces the expression of different genes such as *ada*, *alkA* and *aidB* in cells treated with alkylating agents, leading to the development of adaptive responses induced by pretreatment with sublethal concentrations of alkylating agents (Friedberg *et al.*, 1995).

A cross-adaptive response against lethal effects caused by alkylating agents does not occur when cells of *E. coli* are exposed to low doses of H<sub>2</sub>O<sub>2</sub> (micromolar) (Demple and Halbrook, 1983), but on the other hand, pretreatment with 5 mM H<sub>2</sub>O<sub>2</sub> protects the wild-type strain, as well as *ada*, *ogt*, *ada-ogt*, *alkA* and *aidB* mutants against the lethal effect of MNNG (Assad *et al.*, 1997). Since H<sub>2</sub>O<sub>2</sub> is able to oxidize thiols (Halliwell and Gutteridge, 1999), which are necessary to convert MNNG into the mutagenic methylnitrosamine (Sedgwick and Robins, 1980), similar experiments were performed with N-nitroso-N-methylurea and N-nitroso-N-ethylurea, which do not require activation by thiols, and the results were similar to those obtained with MNNG (Assad *et al.*, 1997). This protection is accompanied by a reduction in the mutation frequency in the wild-type cells and in the *ogt* mutant, but not in the *ada* mutant. So, Ada protein is able to decrease the mutagenic lesions induced by MNNG in H<sub>2</sub>O<sub>2</sub>-pretreated cells. Horsfall *et al.* (1990) have shown that the DNA context influences both the distribution and reparability of alkylation damage. This observation may provide information on the cross-adaptive response between H<sub>2</sub>O<sub>2</sub> and MNNG, since the DNA alkylation pattern induced by MNNG can be altered when the DNA has already been oxidized.

#### Cross-adaptive response between H<sub>2</sub>O<sub>2</sub> and cumene hydroperoxide

Ahp protein expression is under the control of OxyR protein. Ahp plays an important role in protecting bacterial cells against alkyl hydroperoxides, such as cumene hydroperoxide (Storz *et al.*, 1989), and the active enzyme requires the presence of two subunits with molecular weights of 22 kDa (AhpC) and 57 kDa (AhpF).

It has been shown that 2.5 mM H<sub>2</sub>O<sub>2</sub> pre-treatment protects wild-type cells against cumene hydroperoxide killing (Asad *et al.*, 1998). Unexpectedly, this protection is observed in the *oxyΔ3* mutants, meaning that it is an OxyR-independent phenomenon. The protection is mediated by Ahp protein and there is no requirement for novel

protein synthesis (Asad *et al.*, 1998). Electrophoretic profile studies of proteins from wild-type cells treated with  $H_2O_2$  (2.5 mM) for 20 min have shown that a 22 kDa protein (possibly Ahp protein) observed in untreated wild-type cells is faintly visible in  $H_2O_2$ -treated cells. In the latter, instead of the 22 kDa band a 24 kDa band appears. This modification was not observed when wild type was treated with lower concentrations of  $H_2O_2$  (micromolar).

It is known that  $H_2O_2$  causes modification in several *E. coli* proteins (Farr and Kogona, 1991). So, these results suggest that millimolar concentrations of  $H_2O_2$  might induce an alteration in the electrophoretic profile of the smaller subunit of AhpC (22 kDa). It is important to bear in mind that inhibition of protein synthesis does not prevent the appearance of the 24-kDa band promoted by  $H_2O_2$ . These results are interpreted to mean that AhpC 22-kDa subunit undergoes conversion to different oxidized forms when challenged with higher doses of  $H_2O_2$ , resulting in a small alteration of its molecular weight. Since Ahp may prevent oxidative damage and repair lesions caused by active oxygen species (Farr and Kogona, 1991), the AhpC 24 kDa form can be interfering with the repair of lesions caused by cumene hydroperoxide (Asad *et al.*, 1998).

#### DNA damage and repair in *E. coli* challenged by $H_2O_2$ in the presence of iron chelators

The protection conferred by iron scavengers against the effect of  $H_2O_2$  was first demonstrated by Imlay and Linn (1988). In 1991, Asad and Leitão showed that prior treatment with iron chelators such as o-phenanthroline, dipyridyl and desferrioxamine protected the cells against the  $H_2O_2$  lethal effect, but the number of induced DNA-single strand breaks (DNA-SSB) was similar even in the presence of iron chelators. More significantly, the

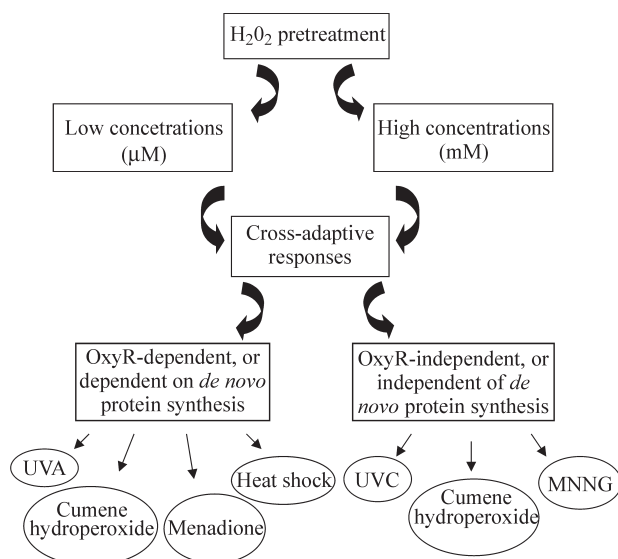
breaks observed after treatment with metal chelators and  $H_2O_2$  are repaired 60 min after  $H_2O_2$  elimination in *xthA* but not *polA* mutant cells (Asad and Leitão, 1991).

Since then, in a series of experiments spanning a full decade, DNA damage and the genetic repair mechanisms of the  $H_2O_2$ -induced DNA lesions in a low-iron condition, which means cells previously treated with metal-ion chelators, have been examined. Asad *et al.* (1995) demonstrated a striking hypersensitivity of the *fpg* and *uvrA* mutant strains to  $H_2O_2$  under low-iron conditions, suggesting that the processing of DNA lesions induced in this situation may occur in a different way from that found under physiological iron conditions.

Other investigations have demonstrated common pathways in the cell response to  $H_2O_2$  challenge in a low-iron condition and in physiological iron conditions. SOS and the OxyR pathways are the ones involved in the response of *E. coli* cells subjected to both challenges (Asad *et al.*, 1997).

Evidence for the participation of endonuclease IV (the *nfo* gene product) in the repair of DNA lesions generated by  $H_2O_2$  under low-iron conditions came from studies of the processing of DNA strand breaks in different *E. coli* strains (Galhardo *et al.*, 2000). Survival experiments with *xthA*, *nfo*, and *xthA nfo* mutant strains in which the cultures were treated with 5 mM  $H_2O_2$  confirm data previously described in the literature, so that the *xthA* mutant was sensitive to  $H_2O_2$  (Demple *et al.*, 1983) while the *nfo* strain was not (Cunningham *et al.*, 1986), and the *xthA nfo* double mutant was even more sensitive than the *xthA* mutant. Experiments with cultures pretreated with dipyridyl showed a distinct pattern of dependence on AP endonucleases for bacterial survival. Neither the *xthA* nor the *nfo* single mutants were significantly inactivated by  $H_2O_2$  under low-iron conditions. However, the *xthA nfo* double mutant was highly sensitive to this treatment, indicating that both exonuclease III and endonuclease IV may act in the repair of the oxidative lesions generated under such conditions. Analysis of the sedimentation profile in alkaline sucrose gradients also demonstrated that both *xthA* and *nfo* mutants, but not the *xthA nfo* double mutant, could carry out complete repair of DNA-SSB generated by  $H_2O_2$  under low-iron conditions (Galhardo *et al.*, 2000). Thus, these findings support the idea that both exonuclease III and endonuclease IV act in the repair of DNA damage induced by  $H_2O_2$  in iron-depleted *E. coli*, and then strongly suggest that the lesions caused by  $H_2O_2$  in the presence of dipyridyl are qualitatively different from those found in the absence of this iron chelator. It is interesting to note that despite the severe repair defect of the *xthA nfo* strain, significant levels of repair were observed under both physiological and low-iron conditions. The exact nature of this repair mechanism remains to be elucidated.

Further studies indicated that the formation of substrates for exonuclease III and endonuclease IV is mediated



**Figure 1** - Cross-adaptive responses induced by low or high  $H_2O_2$  concentration operate through different mechanisms.



by the Fpg DNA glycosylase, since the *fpg* mutation increases cell survival and repair of DNA strand breaks in a null AP endonuclease background (*xthA nfo* double mutant). Recently, Speck *et al.* (2002) showed that the same phenomenon is observed for nitric oxide treatment, that is, in a null AP endonuclease background, *fpg* and *ung* mutations are able to increase cell survival.

#### The role of copper ions in the lethality induced by H<sub>2</sub>O<sub>2</sub> in low-iron conditions

Most of the work on the effects of H<sub>2</sub>O<sub>2</sub> in living organisms has reported that DNA damage induced by H<sub>2</sub>O<sub>2</sub> can be explained basically by the generation of hydroxyl radicals through the iron-mediated Fenton reaction. Some authors have attributed the same role to copper ions, also present in biological systems (Sagripanti and Kraemer, 1989; Aruoma *et al.*, 1991; Dizdaroglu *et al.*, 1991). The suggested mechanism for this effect takes into account the formation of a DNA-Cu<sup>2+</sup> complex in which this metal would be reduced and then the DNA-Cu<sup>+</sup> would react with H<sub>2</sub>O<sub>2</sub> to produce oxidative damage, via Fenton-type reactions (Aruoma *et al.*, 1991; Byrnes *et al.*, 1992; Lloyd *et al.*, 1997; Lloyd and Phillips, 1999). Although copper ions seemed not to participate in the genotoxicity of H<sub>2</sub>O<sub>2</sub> in *E. coli* under physiological iron conditions (Asad and Leitão, 1991), these ions have been shown to take part in the genotoxicity of H<sub>2</sub>O<sub>2</sub> under conditions of low iron availability. In fact, neocuproine (copper ion chelator) can inhibit cell inactivation and DNA strand breakage caused by H<sub>2</sub>O<sub>2</sub> in the presence of iron chelators (Almeida *et al.*, 2000). This phenomenon can only be detected in high concentrations of H<sub>2</sub>O<sub>2</sub> (15 mM), suggesting that these ions only interact with H<sub>2</sub>O<sub>2</sub> in the intracellular environment in the absence of iron and under severe oxidative stress.

Copper-induced DNA damage has been studied in several systems. It has been demonstrated that such damage is targeted preferentially to adjacent polyguanosines (Aruoma *et al.*, 1991). Analysis of the oxidative base lesions generated by copper-mediated Fenton reactions showed that purine residues are the preferred targets of the DNA-damaging species (Sagripanti and Kraemer 1989; Dizdaroglu *et al.*, 1991; Lloyd and Phillips, 1999], with 8-oxoG being the most abundant lesion formed. Frelon *et al.* (2003) have shown that only 8-oxoG is formed upon incubation of DNA with Cu(II) ions and H<sub>2</sub>O<sub>2</sub> and suggested that *in vitro* the Fenton reaction triggered by copper ions generates singlet oxygen as the predominant reactive species, with hydroxyl radical being produced predominantly when the Fenton reaction is triggered by iron ions. In fact, 8-oxoG is the only lesion excised from H<sub>2</sub>O<sub>2</sub>/Cu<sup>2+</sup>-treated DNA at detectable levels by the yOgg1 glycosylase from *Saccharomyces cerevisiae* (Karahalil *et al.*, 1998). The finding of a remarkable increase in mutagenesis in *uvrA fpg*-strain cultures pretreated with dipyrindyl, and its inhibition by neocuproine (Almeida *et al.*, 2000), strongly sug-

gest that 8-oxoG is formed in large amounts by such treatment. The data suggesting a significant role for Fpg and UvrA proteins in the repair of DNA lesions correlate well with the hypothesis of copper participation, since it was demonstrated that these proteins are important in the repair of lesions induced by singlet oxygen, another well-known guanine-damaging agent. Taken together these results may indicate the production of singlet oxygen as the predominant ROS when *E. coli* cultures are challenged with H<sub>2</sub>O<sub>2</sub> under low-iron conditions.

On the other hand some interesting results were obtained from mutagenesis assays performed by our group (manuscript in preparation) regarding the nature of lesions produced by H<sub>2</sub>O<sub>2</sub> in cells previously treated with dipyrindyl. Using an assay based on *lac*<sup>-</sup> reversion through a single base change in mutated *lacZ* codon (Cupples and Miller, 1989), we found that concentrations of H<sub>2</sub>O<sub>2</sub> above 15 mM induce almost exclusively A:T→ T:A transversions. However, in cells previously treated with dipyrindyl a massive and significant presence of G:C→ A:T transitions is detected, a clearly different profile of induced mutations. Considering that this transversion is not reported to be induced by 8-oxoG the authors have suggested that H<sub>2</sub>O<sub>2</sub> under low-iron conditions may generate the lesion 5-hydroxy-2'-deoxycytidine, a highly mutagenic product of cytosine oxidation that also constitutes a substrate for Fpg repair protein (Hatahet *et al.*, 1994; Feig *et al.*, 1994).

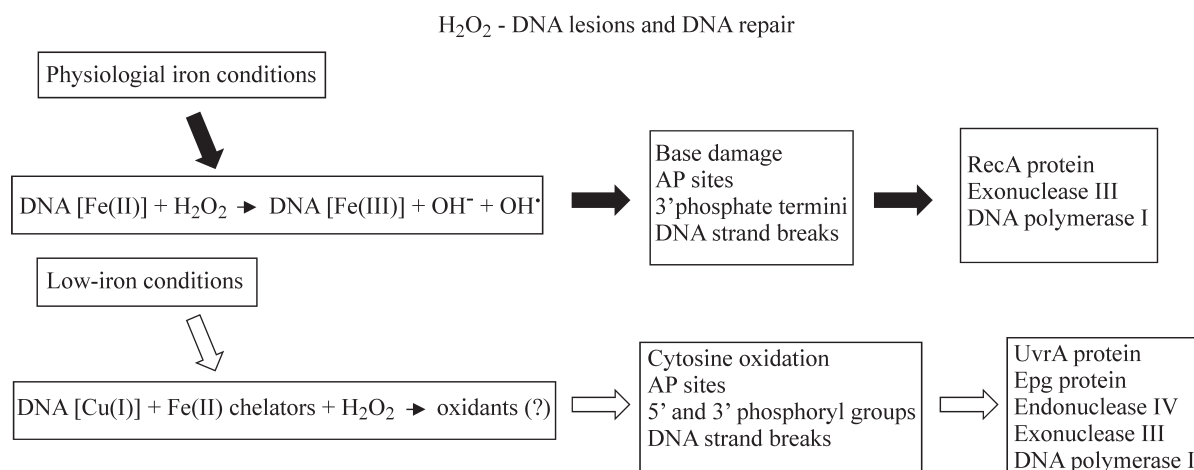
A qualitative difference exists between the DNA lesions generated by H<sub>2</sub>O<sub>2</sub> plus copper (in the presence of iron chelators) and iron (physiologic iron condition), which can be deduced by the different DNA repair requirements for cell survival in these situations. The reason for this difference is not known, and there exists an as yet unsolved question about the participation of hydroxyl radical in the damage to DNA caused by copper plus H<sub>2</sub>O<sub>2</sub> (Sagripanti and Kraemer, 1989; Yamamoto and Kawanishi, 1989). Free-radical scavengers partially inhibit *E. coli* inactivation by H<sub>2</sub>O<sub>2</sub> (Imlay *et al.*, 1988), and we observed that thiourea protects the cells against the lethality produced by H<sub>2</sub>O<sub>2</sub> both in the presence of dipyrindyl (our unpublished results) and in physiological iron conditions (Asad and Leitão, 1991).

#### Synergistic lethal interactions

In the last three decades, many studies have been published regarding synergistic lethal effects between H<sub>2</sub>O<sub>2</sub> and different physical and chemical agents. Some of them will be described in this section.

#### Lethal synergism between UV (254 nm) and H<sub>2</sub>O<sub>2</sub>

Synergistic lethal interaction between UV (254 nm) and X-ray irradiation, a well known system of generating H<sub>2</sub>O<sub>2</sub> and DNA strand breaks was described three decades



**Figure 2** - DNA lesions and DNA repair in physiological and low-iron conditions.

ago in *E. coli* (Haynes, 1966; Martignoni and Smith, 1973). However, synergistic lethal interaction was not observed between UV (254 nm) and H<sub>2</sub>O<sub>2</sub> (Hartman and Eisenstark, 1980). On the other hand, synergistic killing of *E. coli* and *S. typhimurium* by near-UV (300-400 nm) radiation and H<sub>2</sub>O<sub>2</sub> has been described (Hartman and Eisenstark, 1978; Hartman and Eisenstark, 1980; Kramer and Ames, 1987).

An analysis of the UV and X-ray effects (Martignoni and Smith, 1973), the results of repair of H<sub>2</sub>O<sub>2</sub> lesions in several mutant strains (Ananthaswamy and Eisenstark, 1977), and our own results for UV and reductone (an H<sub>2</sub>O<sub>2</sub>-generating keto-aldehyde) (Leitão *et al.*, 1981a; Leitão *et al.*, 1981b) suggested that a synergistic lethal interaction between UV and H<sub>2</sub>O<sub>2</sub> (254 nm) should be observed. Indeed, Leitão and Carvalho (1988) observed that prior UV (254 nm) irradiation strongly increased the sensitivity to H<sub>2</sub>O<sub>2</sub> of wild-type *E. coli* cells, and this synergistic lethal interaction was also observed to a reduced extent in a *polA* mutant, suggesting that UV lesions are potentiated by the additional damage produced by H<sub>2</sub>O<sub>2</sub>, a result similar to those observed by Martignoni and Smith (1973) for UV and X-rays.

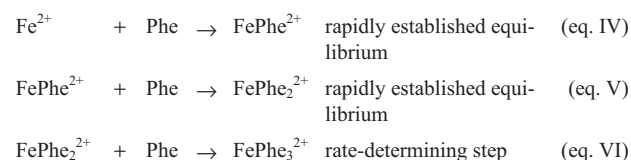
The detection of DNA double-strand breaks by DNA sedimentation on neutral sucrose gradients (Leitão and Carvalho, 1988) clearly indicates a mechanism responsible for the synergistic lethal interaction observed. Although H<sub>2</sub>O<sub>2</sub> can produce DNA strand breaks with low efficiency (Demple and Linn, 1982), the repair of UV lesions by the action of the *uvr* gene products generated single-stranded DNA regions and it was supposed that, in the presence of H<sub>2</sub>O<sub>2</sub>, DNA double-strand breaks might arise in these regions, produced by the action of exonuclease III (the *xthA* gene product). This hypothesis was confirmed by the absence of synergistic lethal interaction between UV (254 nm) and H<sub>2</sub>O<sub>2</sub> in *xthA* as well as in *uvrA* mutant strains (Leitão and Carvalho, 1988). It seems that the same kind of mechanism may be operating in the synergisms observed

with near-UV and H<sub>2</sub>O<sub>2</sub>, UV and reductone and UV and X-rays.

#### Lethal synergism between phenanthrolines and H<sub>2</sub>O<sub>2</sub>

As described earlier, prior treatment with iron chelators such as desferrioxamine, dipyrpyridyl or o-phenanthroline (1,10-phenanthroline) protects *E. coli* cells against the lethal effects of H<sub>2</sub>O<sub>2</sub>. However, Asad *et al.* (1994) detected a strong lethal interaction when *xthA* mutant cells were treated simultaneously with H<sub>2</sub>O<sub>2</sub> and o-phenanthroline. In the same way Almeida *et al.* (1999) have also detected a lethal synergistic interaction between neocuproine (2,9-dimethyl 1,10-phenanthroline) and H<sub>2</sub>O<sub>2</sub>. In both cases the phenomenon of synergism was accompanied by an increase in the number of DNA strand breaks.

In the case of the synergism described by Asad *et al.* (1994), it was argued that the formation and dissociation equilibrium of Fe<sup>2+</sup>-Phe complex, which follows equations IV, V and VI (Lee *et al.*, 1948) shown below, would explain the observed phenomenon.



Since the equilibrium of the bis and mono complexes of Fe<sup>2+</sup>-Phe are rapidly established and since these bis and mono complexes react quickly with H<sub>2</sub>O<sub>2</sub> (Burgers and Prince, 1965), there would be H<sub>2</sub>O<sub>2</sub>-mono and H<sub>2</sub>O<sub>2</sub>-bis complex formation and these complexes may be extremely lethal to the cells, therefore justifying the lethal interaction observed. In this case, the ability of o-phenanthroline to penetrate into the DNA duplex, acting as a shuttle for the Fe<sup>2+</sup> ions would increase the efficiency of OH generation close to DNA (Furtado *et al.*, 1997). Recently Furtado (2002) showed that this interaction also occurs *in vitro*. By



using plasmid DNA, he showed that the high number of breaks, as measured by the transformation of supercoiled to relaxed form, is obtained when iron and o-phenanthroline are added to the reaction medium immediately before H<sub>2</sub>O<sub>2</sub>.

As mentioned above, Almeida *et al.* (1999) showed that prior incubation of *fpg*, *uvrA* and *lexA* mutant strains with neocuproine led to an increased sensitivity to H<sub>2</sub>O<sub>2</sub>. The chemistry of this synergistic lethal interaction was suggested to occur via the iron-mediated Fenton reaction, which would be responsible for displacement of copper ions from the complex NeO<sub>2</sub>Cu<sup>+</sup>, as Florence *et al.*, (1985) have reported to occur *in vitro*. In fact, neocuproine, as well as other phenanthroline derivatives, have the ability to penetrate into DNA, guiding metal ions to this site, which can contribute to radical formation in the vicinity of this molecule (Furtado *et al.*, 1997). Indeed, it is possible that neocuproine can guide copper ions to the DNA molecule, thereby promoting the occurrence of radical generation at this site as H<sub>2</sub>O<sub>2</sub> reacts with these ions.

## Conclusions

The most remarkable outcome from the genetic studies on H<sub>2</sub>O<sub>2</sub>-mediated genotoxicity is the striking sensitivity of *xthA* mutant *E. coli* cells to H<sub>2</sub>O<sub>2</sub>. In fact, this observation provides some insights into the nature of H<sub>2</sub>O<sub>2</sub>-induced lesions. Nevertheless, it is important to realize that the oxidative stress produced by H<sub>2</sub>O<sub>2</sub> results in the induction of a diverse set of physiological responses, which include some paradoxical effects. In this context, the most unexpected phenomenon investigated so far is the bimodal pattern of inactivation of *E. coli* by H<sub>2</sub>O<sub>2</sub> described by Imlay and Linn (1986). In the same way, it is now evident that although pre-treatment of *E. coli xthA* cultures with iron chelators confers protection against the lethal effects of H<sub>2</sub>O<sub>2</sub>, DNA lesions can still be formed under these conditions. We have to keep in mind that in contrast with what is observed in the repair of damage induced by H<sub>2</sub>O<sub>2</sub> under physiological iron conditions, after treatment with H<sub>2</sub>O<sub>2</sub> in low-iron conditions: (i) the lesions observed may be repaired in *xthA* mutant cells (ii) both endonuclease IV and exonuclease III as well as Fpg and UvrA proteins participate in the repair of these lesions. Such a difference in the repair suggests a qualitative difference in the formation of DNA lesions that is independent of the presence of iron. Iron chelators partially inhibit *E. coli* inactivation by H<sub>2</sub>O<sub>2</sub>. However, the precise mechanism for the genotoxic effect of H<sub>2</sub>O<sub>2</sub> under low-iron conditions remains to be elucidated. It is also curious to see that some metal-ion chelators such as o-phenanthroline and neocuproine, which should generally inhibit the lethal effect of H<sub>2</sub>O<sub>2</sub> can sometimes enhance cell inactivation. Moreover, the data available so far suggest that H<sub>2</sub>O<sub>2</sub> (2.5 mM) can induce protection against UV light, MNNG and cumene hydroperoxide independently of the

adaptive response, indicating that H<sub>2</sub>O<sub>2</sub> can produce cross-protection responses through various mechanisms that might not involve the induction of *de novo* protein synthesis.

## Acknowledgements

We thank C.F.M. Menck, D.P. Carvalho and M. Sorenson for critical reading of this manuscript. Research supported by CNPq, CAPES, FUJB, FAPERJ and PRONEX.

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Editor Asociado: Carlos F.M. Menck