



## Si cycling in a forest biogeosystem – the importance of transient state biogenic Si pools

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**Abstract.** The relevance of biological Si cycling for dissolved silica (DSi) export from terrestrial biogeosystems is still in debate. Even in systems showing a high content of weatherable minerals, like Cambisols on volcanic tuff, biogenic Si (BSi) might contribute > 50 % to DSi (Gerard et al., 2008). However, the number of biogeosystem studies is rather limited for generalized conclusions. To cover one end of controlling factors on DSi, i.e., weatherable minerals content, we studied a forested site with absolute quartz dominance (> 95 %). Here we hypothesise minimal effects of chemical weathering of silicates on DSi. During a four year observation period (05/2007–04/2011), we quantified (i) internal and external Si fluxes of a temperate-humid biogeosystem (beech, 120 yr) by BIOME-BGC (version ZALF), (ii) related Si budgets, and (iii) Si pools in soil and beech, chemically as well as by SEM-EDX. For the first time two compartments of biogenic Si in soils were analysed, i.e., phyto-

genic and zoogenic Si pool (testate amoebae). We quantified an average Si plant uptake of 35 kg Si ha<sup>-1</sup> yr<sup>-1</sup> – most of which is recycled to the soil by litterfall – and calculated an annual biosilicification from idiosomic testate amoebae of 17 kg Si ha<sup>-1</sup>. The comparatively high DSi concentrations (6 mg L<sup>-1</sup>) and DSi exports (12 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) could not be explained by chemical weathering of feldspars or quartz dissolution. Instead, dissolution of a relictic, phytogenic Si pool seems to be the main process for the DSi observed. We identified canopy closure accompanied by a disappearance of grasses as well as the selective extraction of pine trees 30 yr ago as the most probable control for the phenomena observed. From our results we concluded the biogeosystem to be in a transient state in terms of Si cycling.

## 1 Introduction

In recent years our understanding of Si cycling in different terrestrial biogeosystems has been improved substantially. Research on steppe (Blecker et al., 2006; Borrelli et al., 2010, White et al., 2012), savannah (Melzer et al., 2010, 2012; Alexandre et al., 2011), forest (Gerard et al., 2008; Cornelis et al., 2010a, 2011a) and wetland biogeosystems (Struyf and Conley, 2009; Struyf et al., 2009) enhanced our understanding of the “plant factor”, i.e., the importance of the biological Si cycling in terrestrial biogeosystems which had been studied in tropical and temperate climates so far (Sommer et al., 2006).

Recently, the complex effects of natural disturbances and anthropogenic perturbations on Si cycling came into focus in Si research. Deforestation (Conley et al., 2008), invading insects (Grady et al., 2007), or fire (Engle et al., 2008) clearly modifies annual DSi exports from catchments. Land use change (LUC) was added as another control on DSi exports at regional scale and conceptual models including those for human activities were developed (Struyf et al. 2010; Carey and Fulweiler, 2012). The results of these studies challenge the steady state assumption implied in most studies on Si budgets.

Although a spatial hierarchy of driving factors for DSi exports is well known (Sommer et al., 2006; Cornelis et al., 2011b), the origin of DSi is still under debate. The major research questions are the following: How large is the contribution of BSi pool to DSi compared to litho-/pedogenic sources? What are the main drivers of the relative importance of biogenic and mineral sources – climate, lithology, state of soil development, soil pattern, land use? To answer these questions Cornelis et al. (2011b) developed a conceptual framework. They defined four different scenarios (“end members”) based on climate, soil, and vegetation: scenario 1 (weathering unlimited) represents optimal climate and soil conditions. Here the DSi exports from soil-plant systems are controlled by climate, i.e., temperature and runoff. In scenario 2 (soil weathering limited) climate is optimal, but soils are poor in weatherable minerals. Consequently, DSi export is controlled by a near-surface biogeochemical cycling in the plant-soil system rather than by geochemical processes in deeper soil layer. In scenario 3 (climate weathering limited) soils contain weatherable minerals. Therefore, DSi might be controlled by either biogenic or pedo-/lithogenic sources. Their relative proportions depend on local conditions (lithology, geomorphology, land use). By scenario 4 (weathering limited) deserts are covered where eolian fluxes are the only relevant Si export from the system. According to Cornelis et al. (2011b) each scenario should be characterized by distinct DSi geochemical signatures ( $\delta^{30}\text{Si}$ , Ge/Si) which can be used in tracing Si pathways in soil-plant systems.

However these signatures have to be combined with Si flux analysis (incl. mass balances) and Si pool quantifications in order to understand the fate of Si in terrestrial bio-

geosystems. Unfortunately, the number of studies in which Si fluxes in/from the soil-plant system are directly linked to a detailed analysis of Si pools is still limited (e.g., Alexandre et al., 1997; Lucas, 2003; Blecker et al. 2006; Gerard et al., 2008; White et al. 2012). Furthermore, no study included a zoogenic Si pool, although first results on testate amoebae dynamics already showed its relevance in terms of biosilicification (comparable to plants, Aoki et al., 2007).

Here we present results on Si cycling in a forest biogeosystem which experienced vegetation shifts due to forest management. We hypothesised that the effects of Si biocycling on DSi must be most pronounced in cases of very low silicate weathering. Therefore, we selected a biogeosystem showing a quartz dominance in soils and parent material (> 95 %). Based on a system approach, we quantified Si fluxes (internal, external) for a 4 yr period and interpreted DSi exports in terms of litho-/pedogenic and biogenic sources as well as vegetation shifts. For the first time the quantification of the biogenic Si pool comprises two compartments, the phyto- and zoogenic pool.

## 2 Materials and methods

### 2.1 Environmental setting

Our study site “Beerenbusch” is located in northern Brandenburg close to the village Rheinsberg (53°09' N, 12°59' E; 78 m a.s.l.; Fig. S1 in Supplement). The climate is characterised by mean annual air temperature of 8.7 °C and average precipitation of 600 mm y<sup>-1</sup> according to long-term measurements (1981–2010) of the meteorological station Neuglobsow/Menz of the German Weather Service (DWD) which is located 4 km E of the forest stand. The mean precipitation during the study period (05/2007–04/2011) was 15 % higher (689 mm) compared to the 30 yr average.

The study site is located at a Weichselian outwash plain (sandur) of the Rheinsberg Basin in the foreland of Late Pleistocene terminal moraines (“Fürstenberger Staffel”, Ginzel and Ertl, 2004). The soil is classified as *Brunic Arenosol (Dystric)* according to WRB (2006) and *Lamellic Udipsamment* according to Soil Taxonomy (Soil Survey Staff, 1999). Humus enriched topsoils extend down to a depth of 35 cm (Fig. S2 in Supplement). Bleached quartz grains indicate a slight podzolization for the first mineral horizon (AE). Brunification leads to a bright brownish color in the subsoil (Bw horizons) down to a depth of 80 cm. Single thin clay lamellae in subsoil and parent material indicate clay translocation macroscopically (down to 120 cm).

### 2.2 Land use and forest management history

The study site is embedded into a large forested area, called “Naturpark Stechlin-Ruppiner Land”. Reconstruction of land use history revealed Beerenbusch to be covered by forests for at least 230 yr (Fig. S3 in Supplement). Before 1888 the site

was used as pine forest (*Pinus sylvestris* L., planted in 1826) (Figs. S4, S5 in Supplement). In 1888 European beech (*Fagus sylvatica* L.) was planted into the pine stand as an understorey. A mixed forest of beech and pine developed since. As canopy closure continued, grasses of the understorey successively disappeared due to light limitation. Already in 1954 only few local spots with grasses in the ground vegetation existed (Table S6 in Supplement). Canopy closure was accomplished in the early 1980s. From 1985 to 1988, the upper pine layer of the mixed forest stand was cut stepwise. The recent plant community classifies as a *Majanthemo-Fagetum* with coverages of 85 % beech in the upper storey, 5 % in the intermediate layer and 20 % in the understorey. The recent ground vegetation shows herbs (1 % coverage), mosses (5 %), and lichens (< 1 %) (F. Becker, personal communication, 2004), but no grasses (Fig. S3 in Supplement). In 2008 the mature beech stand is characterised by a mean tree height of 28 m and a timber volume of 369 m<sup>3</sup> ha<sup>-1</sup>. An ICP Forests site (DE1207) was established and instrumented at a 0.5 ha fenced area in 2001 (Fig. S2 in Supplement).

### 2.3 Sampling and analysis of soils, plants, phytoliths, and testate amoebae

A representative soil and a sediment core had been analysed for basic textural and chemical properties prior to this study (Lachmann, 2002; Jochheim et al., 2007a; Fig. S2 in Supplement). Spatial soil heterogeneity was characterized in a 25 × 25 m raster by Lachmann (2002). From his results a quite similar soil horizonation and sediment layering can be concluded for the 0.5 ha study site. Therefore, we merged results of both samplings (soil pit, core) into one depth function (down to 2.8 m). For Si analysis a resampling from the original soil pit occurred end of September 2009.

Litter and soil material was taken by horizons down to a depth of 1.25 m and stored in plastic bags. Undisturbed soil cores (100 cm<sup>3</sup>) were taken in the middle of soil horizons and dried at 105 °C for determination of bulk density (BD) (Lachmann, 2002). Bulk densities for sediments of the core were calculated by bulk density = (1 – total porosity) × density. Assuming a total porosity of 0.36 (36 Vol. %), which is reasonable for near-surface glaciofluvial sands, and a quartz density of 2.65 g cm<sup>-3</sup> we calculated a bulk density of 1.7 g cm<sup>-3</sup>. Bulk soil samples were air dried, gently crushed and sieved at 2 mm to separate the fine earth fraction (< 2 mm) from gravel (> 2 mm). The particle size distribution of the fine earth was determined by a combined wet sieving (> 63 μm) and pipette (< 20 μm) method (DIN ISO 11277, 1998). Pretreatment for particle size analysis was done by wet oxidation of organic matter using H<sub>2</sub>O<sub>2</sub> (10 Vol. %) at 80 °C and dispersion by shaking the sample end over end for 16 h with a 0.01 M Na<sub>4</sub>P<sub>2</sub>O<sub>7</sub>-solution (Schlichting et al., 1995). Soil pH was measured using a glass electrode in 0.01 M CaCl<sub>2</sub> suspensions at a soil to solution ratio of 1 : 5 (*w/v*) (DIN ISO 10390, 1997) af-

ter a 60 minute equilibration period. The total carbon content was determined by dry combustion using an elemental analyser (Vario EL, Elementar Analysensysteme, Hanau, Germany). CaCO<sub>3</sub> was determined conductometrically following Scheibler (Schlichting et al., 1995). Organic carbon (C<sub>org</sub>) was calculated as the difference between total carbon and carbonate carbon. In soil horizons and sediments without carbonates total carbon equals soil organic carbon. Pedogenic oxides were characterised by dithionite-soluble (DCB) Fe (Fe<sub>d</sub>) and dark acid-oxalate soluble Fe, Al, and Si (Al<sub>o</sub>, Fe<sub>o</sub>, Si<sub>o</sub>) following the procedures of Mehra and Jackson (1960) and Schwertmann (1964), respectively (Schlichting et al., 1995). The element concentrations in solutions were determined by ICP-OES. All basic soil analyses were performed in two lab replicates.

#### 2.3.1 Water extractable Si (Si<sub>H<sub>2</sub>O</sub>)

Water extractable Si was determined by the method of Schachtschabel and Heinemann (1967). Ten grams of dry soil (< 2 mm) were weighted into 80 mL plastic centrifuge tubes and 50 mL distilled water added together with three drops of a 0.1 % NaN<sub>3</sub>-solution to prevent microbial activity. The total extraction lasted seven days in which tubes were shaken by hand twice a day. Tubes were not shaken mechanically to avoid abrasion of coarse mineral particles colliding during shaking (Mc Keague and Cline, 1963). Finally the extraction solution was centrifuged (4000 rpm, 20 min), filtrated (0.45 μm polyamide membrane filters) and Si was measured by ICP-OES. Analyses were performed in three lab replicates.

#### 2.3.2 Tiron extractable Si (Si<sub>Tiron</sub>)

The Tiron (C<sub>6</sub>H<sub>4</sub>Na<sub>2</sub>O<sub>8</sub>S<sub>2</sub>·H<sub>2</sub>O) extraction procedure was developed by Biermans and Baert (1977) and modified by Kodama and Ross (1991). It has been used to quantify amorphous biogenic and pedogenic Si, so-called “pedogenic silica” (Kendrick and Graham, 2004), although a partial dissolution of primary minerals is well known (Kodama and Ross, 1991; Sauer et al., 2006). The extraction procedure is as follows: dilution of 31.42 g Tiron (ACROS Organics, Geel, Belgium) with 800 mL of distilled water, followed by addition of 100 ml sodium carbonate solution (5.3 g Na<sub>2</sub>CO<sub>3</sub> in 100 mL distilled water) under constant stirring. After that the pH of the solution increases from 3.3 to 7.5. Finally, the pH is brought to 10.5 by adding small volumes of a 4 M NaOH-solution. The Tiron solution is transferred into a 1 L volumetric flask and has a final concentration of 0.1 M. For the extraction 30 mg of dry soil is weighted into 80 mL centrifuge tubes and a 30 mL aliquot of the Tiron solution added. The tubes are then heated at 80 °C in a water bath for 1 h. After cooling adhering water on the surface of the tubes is removed, the tubes are weighted and water lost by evaporation is replaced. The extracted solutions were centrifuged at

4000 rpm for 30 min, filtrated (0.45  $\mu\text{m}$  polyamide membrane filters, Whatman NL 17) and elements were measured by ICP-OES. Analyses were performed in three lab replicates.

### 2.3.3 Mineralogy, micromorphology and SEM-EDX analysis

Powder samples of each soil horizon (< 2 mm fraction) were analysed for basic mineral composition using a BRUCKER AXS D5000 Diffractometer (Cu-K $\alpha$  radiation). The relative intensities of the diffraction maxima were used for a semiquantitative estimation of the concentration of mineral species present. The counts from the main reflection peak of all minerals were summed up and the relative proportion of each mineral was calculated as percent of the total. A subsample of the < 2 mm fraction was placed on an Al-stub, fixed by adhesive tape, coated with minimal amount of gold-palladium and analysed on element composition by microprobe analysis (Hitachi S-2700 device, EDX-X-Flash-Detector with SAMX-Software at ZELMI, TU-Berlin).

Four undisturbed soil samples (Kubiena boxes of 8 cm height) were taken from 0–8 cm (AE, Ah), 10–18 cm (AB), 44–52 cm (Bw<sub>1</sub>) and 104–112 cm (2Cwt) (Fig. S2 in Supplement). Air-dried samples were impregnated with Leguval resin under vacuum. After hardening 24  $\mu\text{m}$  thick thin sections were prepared. The micromorphological features were described according to the concepts and terminology proposed by Stoops (2003). For thin section descriptions a SM-LUX-POL (Leitz) microscope with polarisation filter was used. After description thin sections were coated with carbon in a vacuum evaporator, and then examined with electron microprobe analyses (Cameca Camebax Microbeam, ZELMI at TU Berlin), using an accelerating potential of 20 kV. Element distribution maps (Si, Al, Fe, K, Mg, Ca, Na, Ti) were obtained with the same instrument. Feldspar grains and cutans (Fig. 5) were identified by microscopic examination on all thin sections (between 10 and 20 replicates, depending on the number of feldspar grains in each thin section). To obtain information about the weathering state of potassium feldspars, strewn slides of soil material from three different depths (10–20, 40–60, and 100–130 cm) were prepared and analysed using SEM (Fig. 6). For identification of feldspars, samples were analysed with REM-EDX and the potassium distribution in the slides was recorded. Each strewn slide was divided into four subsections and a minimum of ten replicates per subsection were analysed.

### 2.3.4 Phytolith separation from plants and soils, SEM-EDX analysis

Phytoliths were extracted from litterfall, the organic surface layer (L) as well as from the first three mineral horizons (AE, Ah, AB). Litterfall from one sampling date (May to August 2008) was separated into four groups: leaves, bud scales, fruit capsules and wood from twigs and branches. The ex-

traction procedure for litterfall was similar to soil horizons except steps 3 and 4 (see below). From each horizon 10 g of dry soil material (< 2 mm) were processed in four steps (adapted from Alexandre et al., 1997). First organic matter is oxidized using H<sub>2</sub>O<sub>2</sub> (35 Vol. %), HNO<sub>3</sub> (65 Vol. %), HClO<sub>4</sub> (70 Vol. %) at 80 °C until reaction subsides. Secondly, carbonates and Fe oxides are dissolved by boiling the sample in HCl (10 Vol. %) for 30 min. Thirdly, the < 2  $\mu\text{m}$  granulometric fraction is removed by dispersion of remaining solid phase of step 2 with 2 Vol. % sodium hexametaphosphate solution (6–12 h), centrifugation at 1000 rpm for 2–3 min, and subsequent decantation. Finally, the phytoliths are separated by shaking the remaining solid phase of step 3 with 30 mL of sodium polytungstate Na<sub>6</sub>(H<sub>2</sub>W<sub>12</sub>O<sub>40</sub>) · H<sub>2</sub>O (density of 2.3 g cm<sup>-3</sup>), centrifugation at 3000 rpm for 10 min, carefully pipetting the supernatant, and filtering by 5  $\mu\text{m}$  teflon filter. This step was repeated three times. The filter residue was washed with water, bulked, dried at 105 °C, and weighted.

A subsample was placed on an Al-stub, fixed by adhesive tape, and coated with minimal amount of gold-palladium. Ten micrographs per stub were made using a JEOL JSM6060 LV SEM microscope (500  $\times$  magnification). Phytoliths of each micrograph (coverage approx. 200  $\times$  200  $\mu\text{m}^2$ ) were counted. The database comprises a total number of 462 (L), 459 (AE), 422 (Ah), and 238 phytoliths (AB). They were described by shape according to ICPN (Madelá et al., 2005) and assigned to vegetation (Golyeva, 2001) wherever applicable. Further, all counted phytoliths were assigned to one of three classes of phytolith dissolution: (i) plain phytoliths, (ii) phytoliths showing some surface etching, and (iii) phytoliths with strong dissolution features. Phytoliths at stubs were analysed on element composition by microprobe analysis (Hitachi S-2700 device, EDX-X-Flash-Detector with SAMX-Software at ZELMI, TU Berlin).

The Si pool of phytoliths was calculated assuming phytoliths to consist of pure SiO<sub>2</sub>. This results in a slight overestimation of the real Si pools, because biogenic opal contains ca. 10 % H<sub>2</sub>O and some accessory elements (< 1 %) (Bartoli and Wilding, 1980).

### 2.3.5 Plant analysis

The collected plant litter (see Sect. 2.4) was oven dried at 105 °C and milled in a planet type ball mill using milling vessels and balls made from ZrO<sub>2</sub>. Sample aliquots of approximately 200 mg were digested under pressure in PFA digestion vessels using a mixture of 2.5 mL HNO<sub>3</sub> and 1 mL HF at 220 °C and approximately 100 bar (Ultra Clave II, MLS GmbH, Leutkirch, Germany). Silicon was measured by ICP-OES (Vista Pro, Varian Inc., Australia) using a HF-resistant sample introduction system, radial viewing of the plasma, and matrix matched external calibration.

### 2.3.6 Quantification of testate amoebae and related Si pool

For testate amoebae analysis 5 field replicates (20 cm × 20 cm) were placed randomly at Beerenbusch site avoiding areas close to stems (April 2011; Fig. S2 in Supplement). The litter layer (beech leaves) was removed and volumetric soil samples were taken in two segments: 0–2.5 cm and 2.5–5 cm (= sample volume of 1000 cm<sup>3</sup>). Aliquots of 2 g were taken for amoebae analysis in the field and stored in 8 mL formalin (4% aqueous formaldehyde solution). Total soil material of each depth increment and 20 cm × 20 cm area was oven dried at 105 °C and weighted. Bulk densities were calculated by dividing total soil mass by sample volume (1000 cm<sup>3</sup>). Testate amoebae were determined at species level and enumerated directly with an inverted microscope using stained (aniline blue) soil suspensions received from serial dilution (30–500 mg soil in 8 mL water per sample) as reported by Wanner (1999). Thereby living individuals and empty tests were distinguished. All species were assigned either to idiosomic or xenosomic amoebae taxa building up their tests from idiosomes (siliceous platelets synthesised by amoebae from H<sub>4</sub>SiO<sub>4</sub> in soil solution) or xenosomes (extraneous materials such as mineral particles), respectively (e.g., Meisterfeld 2002a, b). Idiosomic Si pools (g m<sup>-2</sup>) of the upper 5 cm were calculated by (1) multiplying Si content of tests (Table 1 in Aoki et al., 2007) with counted individuals of each species, (2) multiplying these data with bulk densities and thickness (2.5 cm), and (3) finally summing up both depth increments.

### 2.4 Site instrumentation for flux determinations

The investigations of the study site water budget started in May 2001. The instrumentation as well as results of the first 4 yr are described in Jochheim et al. (2007a). Precipitation was measured continuously at an open field located 500 m south of the study site using a heated rain gauge (F&C GmbH Gülzow). Gaps in the precipitation data were filled by using open land precipitation data of either (i) the ICP Forests level II plot DE1202, or (ii) the weather station in Neuglobsow/Menz of the German Weather Service (DWD). For silicon analysis (started in May 2007) precipitation water of two rain samplers (RS200, UMS GmbH, Munich; 314 cm<sup>2</sup> each) were collected weekly. Throughfall was measured continuously using a gutter of 0.8 m<sup>2</sup> area with a tipping bucket rain gauge. Additionally, weekly measurements were carried out using 15 rain samplers (see above). Stemflow was measured continuously at one stem with a tipping bucket rain gauge, and additional weekly on 4 stems by sampling the water in barrels. For silicon analysis the weekly samples of open land precipitation, throughfall, and stemflow were bulked to monthly samples. Si concentrations in all waters were determined by ICP-OES.

**Table 1.** Mineralogical composition (wt.-%) of soil horizons and sediment layers by semiquantitative X-ray diffraction.

Depth (cm)	Quartz	Feldspar	Pyroxene
0–10	97.1	2.5	0.4
10–20	96.0	3.6	0.4
30–60	97.7	1.8	0.5
60–80	96.5	3.0	0.5
80–100	93.0	6.6	0.4
100–130	97.4	2.1	0.5
240–260	97.4	2.6	nd
480–500	98.0	2.0	nd

nd = not detected.

Litterfall was collected in 8 inverted pyramidal litter traps (0.25 m<sup>2</sup> each) bi-weekly during four years (05/2006–04/2010) and separated into leaves, flowers, bud scales, beechnuts, fruit capsules, and wood from twigs and branches. Each fraction was bulked into three periods per year (Jan–Apr, May–Aug, Sep–Dec). For silicon analysis see Sect. 2.3.5.

Soil solution was sampled using borosilicate suction probes (EcoTech Bonn GmbH) from 20, 70, and 250 cm soil depths (mean of 150 cm and 250 cm distance to stem; 4 replicates per depth and distance). The samples were collected by applying a suction of –30 kPa in 20 and 70 cm, and –35 kPa in 250 cm. Water was stored within the shafts of the suction probes and sampled bi-weekly. For silicon analysis the 4 field replicates per depth and distance were bulked. Si in soil water was measured by ICP-OES.

Analysis of water content was conducted hourly at different distances to stem (50, 150, 250 cm) of one stem in soil depths 20, 70, and 250 using Theta-probes ML2 (Delta-T Devices Ltd Cambridge, UK). Additionally, close to two further trees using identical distances to stem soil moisture were measured bi-weekly using TDR-probes (FOM/m-TDR, EasyTest Lublin, Poland) at identical soil depths.

Xylem sap flux was measured continuously during the vegetation periods of 2002–2005 on ten selected trees in 1.3 m tree height using the method after Granier (1985) and calculated to representative stand canopy transpiration following Lüttschwager and Remus (2007).

Aboveground woody tree biomass and forest growth were calculated by measuring stem diameter at breast height and tree height of all 108 trees of the plot in spring of 2006, 2008, and 2010 using form factors derived from the beech yield table (Dittmar et al., 1986), wood density (Trendelenburg and Mayer-Wegelin, 1955), bark density (Dietz, 1975), and bark fractions (Altherr et al., 1974).

### 2.5 Calculation and modeling of Si fluxes

Deposition of Si was calculated from the open land precipitation. Si fluxes were calculated by multiplying water fluxes

with Si concentrations. Stand precipitation equals the sum of Si fluxes in throughfall and stemflow, whereas leaching from canopy is the difference of Si fluxes in stand precipitation and open land precipitation. Silicon uptake equals the sum of Si fluxes in litter fall, leaching from canopy, and wood increment. Si export through harvest was calculated as a sum of current accumulation of Si in stem wood including bark.

The simulation of water budget was carried out with the dynamic model Biome-BGC (version ZALF; Jochheim et al., 2007b; Puhlmann and Jochheim, 2007) which is based on Biome-BGC (Thornton et al., 2002). The model runs in daily time steps. It was re-calibrated and validated on the basis of data from intensive forest monitoring sites (Jochheim et al., 2009) as well as on forest yield tables. For this application the model was calibrated based on the measurements of the stand started in 2001. Silicon fluxes with drainage water were calculated for four years (05/2007–04/2011) by multiplying the Si concentrations in soil water from suction probes with simulated soil water fluxes. Vertical distribution of passive Si uptake in soil was estimated from vertical distributed soil water uptake rates multiplied by Si concentrations in soil water. As the Si concentrations in soil water were analyzed in 3 soil depths only (20, 70, 250 cm), they were extrapolated to all other soil depths using the vertical distribution of water extractable Si.

### 3 Results

#### 3.1 Basic soil properties and soil mineralogy

The studied soil is very sandy showing a sand content > 85 % and a dominance of medium sand fraction (0.2–0.63 mm, Fig. 1). In the upper 50 cm a slight increase in silt can be observed. Clay content is always below 3 % with slightly higher values in the upper 1.5 m. The soil is decalcified down to a depth of 1.8 m (Fig. 1). Acidification leads to pH values between 4.3 and 4.5 in the upper 1.6 m. Below that depth pH increases to > 7.0 due to carbonatic sediments (2–4 % CaCO<sub>3</sub>). Quartz is the dominant mineral throughout the soil horizons and sediment layers (Table 1). Only minor additions of feldspars (orthoclase > plagioclase), pyroxene and calcite occurred. Organically complexed Fe, Al and pedogenic oxides (Fe, Al) decreased from C-enriched surface horizons (upper 20 cm) to subsoil horizons as crystallinity of iron oxides increased (Fe<sub>o</sub> : Fe<sub>d</sub> 0.6 → 0.2). The molar Si / Al ratios in oxalate extracts (Si<sub>o</sub> : Al<sub>o</sub>) remained below 0.3 in all soil horizons rendering neof ormation of short-range order minerals, like allophane or imogolite (Si / Al ≈ 0.5), hardly probable.

#### 3.2 Si pools in soils

Water soluble Si shows a decrease from 16 mg kg<sup>-1</sup> in the upper 2 cm to 4 mg kg<sup>-1</sup> in subsoil horizons and a recurring increase in sediments containing carbonates (Fig. 1).

The water-soluble Si pool down to 2.8 m equals 21 g Si m<sup>-2</sup> (= 210 kg Si ha<sup>-1</sup>). The uppermost meter contains 6 g Si m<sup>-2</sup> (= 60 kg Si ha<sup>-1</sup>).

Tiron extractable Si is three orders of magnitude higher compared to water soluble Si. The depth function of Tiron extractable Si shows a continuous decrease from 3 g kg<sup>-1</sup> in topsoil horizons to < 2 g kg<sup>-1</sup> in deepest sediments. The highest Si content can be observed in the uppermost horizon (AE). The Tiron extractable Si pool down to 2.8 m equals 10 kg Si m<sup>-2</sup> (= 100 Mg Si ha<sup>-1</sup>). The uppermost meter contains 4 kg Si m<sup>-2</sup> (= 40 Mg Si ha<sup>-1</sup>).

#### 3.2.1 Phytogenic Si pool in soil

Phytolith contents decrease from litter to mineral soil horizons by one order of magnitude (Table 2a). The upper 20 cm of the soil contain 140 g phytoliths m<sup>-2</sup> (= 1400 kg ha<sup>-1</sup>). Assuming all phytoliths to consist of pure SiO<sub>2</sub> (EDX spectra in Fig. 2c) we calculated 66 g Si m<sup>-2</sup> for the phytolith Si pool in the upper 20 cm (= 660 kg Si ha<sup>-1</sup>). This simplification results in a slight overestimation of the true phytogenic Si pool as biogenic opal contains approx. 10 % H<sub>2</sub>O and some accessory elements (< 1 %) (Bartoli and Wilding, 1980).

The phytolith assemblage of soil horizons – as assignable by shape (ICPN, Madella et al., 2005) – is dominated by elongate (polylobate, fusiform) and short-cell phytoliths (bilobate, trapeziform) with minor contribution of globular and vascular phytoliths. In terms of vegetation a dominance of grass phytoliths were found in mineral soil horizons below 2 cm (Table 2b, Fig. 2b). This fits very well to the vegetation survey from 1954 which showed plants high in Si, such as *Calamagrostis epigejos* (2.2 % Si), *Brachypodium sylvaticum* (3.1 % Si), and *Agrostis capillaris* (1.4 % Si), dominating the ground vegetation (Table S6 in Supplement, Si data from Hodson et al., 2005). Pine and moss phytoliths (rounded particles, Al-rich) can be identified as well (Table 2b, Fig. 2b). However, as there are no more pine and grasses growing at the study site (at least for the last 30 yr) pine and grass phytoliths extracted from the soil represent a relict biogenic Si pool. Surprisingly, clearly identifiable beech phytoliths only account for a minor portion in the upper centimeters. Further, those forms isolated from litterfall (Fig. 2a) are hardly detectable in their original shape in the soil, even not in the upper 2 cm (Fig. 2b). From these findings either a rapid mechanical destruction or a preferential dissolution of beech phytoliths can be concluded. Nevertheless, one has to recognize that 50–75 % of all phytoliths counted cannot be assigned to any vegetation throughout the uppermost 20 cm of the soil (Table 2b).

#### 3.2.2 Zoogenic Si pool in soil

A total number of 6.1 × 10<sup>8</sup> m<sup>-2</sup> testate amoebae (60 % living individuals) was determined in the upper 5 cm of the soil. This number lies in the range of published data (e.g.,

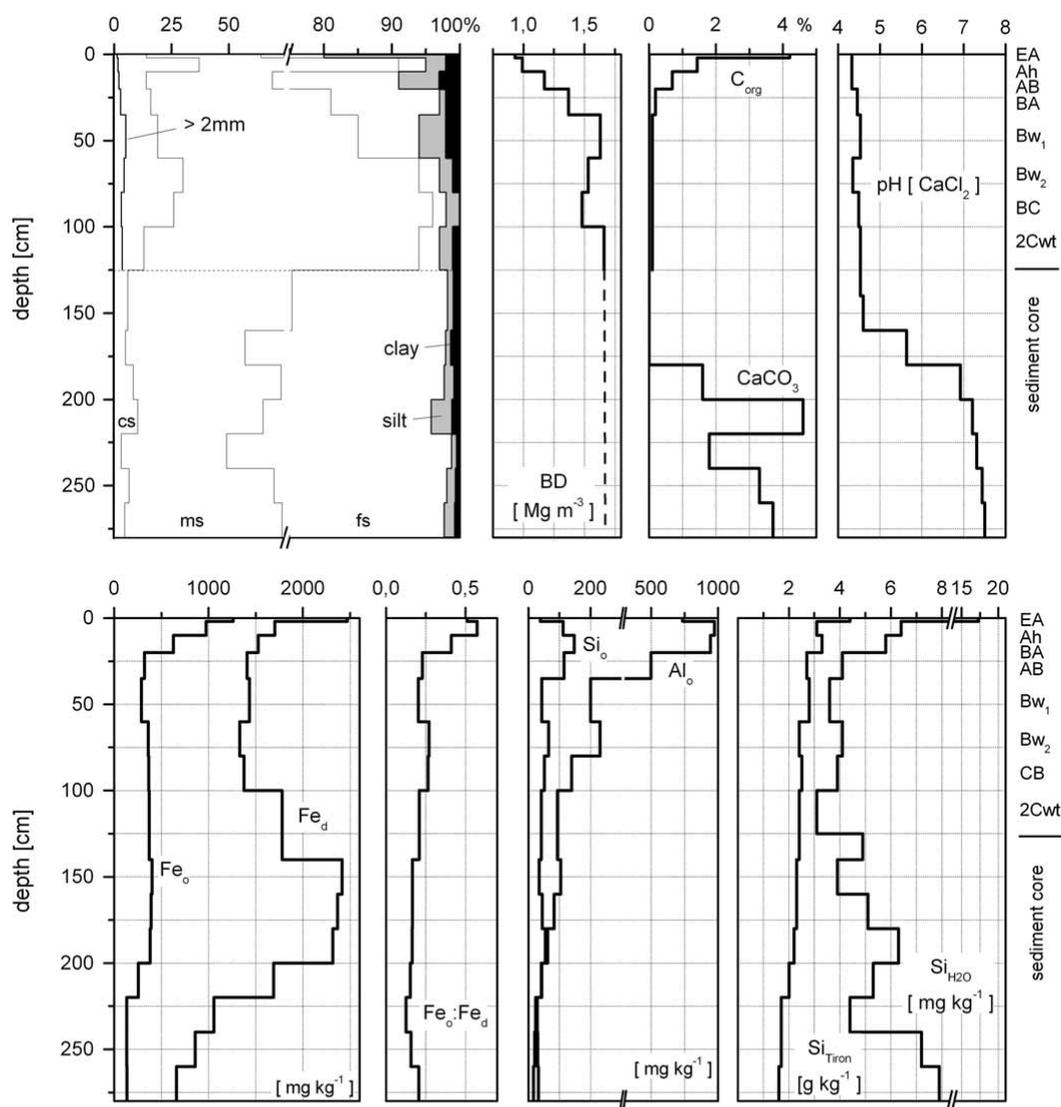


Fig. 1. Depth functions of basic soil properties from the *Brunic Arenosol* (Dystric).

Aoki et al., 2007; Ehrmann et al., 2012; Wanner and Dunger, 2001). Approximately 50 % of all individuals (living and empty tests) belong to either idiosomic or xenosomic taxa of testate amoebae. Xenosomic taxa build up their tests from extraneous materials such as mineral soil particles. Dominant xenosomic taxa at our site are *Phryganella acropodia*, *Plagiopyxis declivis*, and *Centropyxis sphagnicola* (Fig. 3d). Dominant idiosomic taxa embrace *Trinema complanatum*, *Euglypha rotunda*, and *Trinema lineare* (Fig. 3a–c). Only idiosomic taxa can be regarded as a biogenic Si pool influencing dissolved Si (DSi), because they synthesise siliceous platelets for their tests from silica of the soil solution (Anderson, 1994). Therefore, we calculated the zoogenic Si pool only on basis of idiosome-producing amoebae and came up with  $0.19 \text{ g Si m}^{-2}$  in the upper 5 cm of the soil ( $= 1.9 \text{ kg Si ha}^{-1}$ , Table 3). Compared to soil's phytolith Si

pool as derived from density separation, these numbers are two orders of magnitude lower.

### 3.3 Si in aboveground plant biomass

Beech leaves showed the highest Si concentrations followed by bark of branch and stem wood, bud scales, and fruit capsules (Table 4). Beechnuts and woody biomass without bark contain the lowest Si concentrations. Interannual variations in Si concentrations are largest in leaves, bud scales, and fruit capsules. In 2006 lowest Si concentrations in leaves were measured ( $7 \text{ g kg}^{-1}$ ). In litterfall of 2007 Si concentrations nearly doubled ( $13.6 \text{ g kg}^{-1}$ ) which might be an effect of the length of growing period (dry summer in 2006). In terms of intra-annual variations a seasonal trend of Si concentrations can be observed: Si concentrations of leaves from early

**Table 2a.** Phytoliths and related Si pools.

		Depth (cm)	Bulk density (Mg m <sup>-3</sup> )	Phytolith (g kg <sup>-1</sup> )	Phytolith (g m <sup>-2</sup> )	Si content (g kg <sup>-1</sup> )	Phytolith Si, calculated (g kg <sup>-1</sup> )	Phytolith Si, calculated (g m <sup>-2</sup> )
Litterfall (5–8/08)	leaves	–		5.1	–	4.4	2.4	–
	bud scales	–		7.0	–	3.6	3.3	–
	fruit capsules	–		1.1	–	1.1	0.5	–
	branches	–		1.0	–	1.4	0.5	–
Soil	litter layer (L)			8.7			4.1	
	AE	0–2	0.92	0.9	17		0.4	8
	Ah	2–10	0.99	1.2	92		0.5	43
	AB	10–20	1.17	0.3	32		0.1	15
	<i>sum</i>	0–20				140		66

**Table 2b.** Morphological description and identification of phytoliths; n.a. = not assignable.

Soil Horizon	Depth (cm)	Count-% of total phytolith number (10 SEM micrographs)									
		Globular	Elongate	Short cell	Vascular	n.a.	Grasses	Beech	Mosses	Pine	n.a.
Litter layer (L)		7	25	18	1	49	7	12	4	1	75
AE	0–2	7	25	20	2	45	16	5	5	3	72
Ah	2–10	0	38	15	0	47	44	0	1	2	53
AB	10–20	0	27	8	0	64	45	0	0	1	54

**Table 3.** Si pools of idiosomic testate amoebae; mean values and standard deviation (in brackets),  $n = 5$  field replicates.

Depth (cm)	Bulk density (Mg m <sup>-3</sup> )	Living mg Si m <sup>-2</sup>	Empty shells mg Si m <sup>-2</sup>	Total mg Si m <sup>-2</sup>
0–2.5	0.10 (0.04)	81(110)	54 (48)	135 (158)
2.5–5	0.36 (0.11)	34 (25)	22 (11)	57 (23)
<i>sum</i> 0–5		116 (106)	76 (43)	192 (148)

stages of vegetation development (May–Aug) were lowest (4.4–6.8 g kg<sup>-1</sup>), whereas during autumn/winter (Sep–Dec) or winter/spring (Jan–Apr) Si concentrations range from 6 to 14 g kg<sup>-1</sup> (depending on single year).

Taking into account the biomass of each plant compartment the total Si pool in aboveground biomass summarises up to 83 kg Si ha<sup>-1</sup> (Table 4). The stem bark contributes the largest fraction (50 %) followed by leaves (36 %), branch bark (6 %), and stem wood (3 %).

### 3.4 Internal and external Si fluxes

Si uptake by plants contributes the largest internal Si flux in the biogeosystem (35 kg Si ha<sup>-1</sup> yr<sup>-1</sup>, Fig. 4). The major part is transported into the leaves (30 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) rendering autumn litter fall the most important annual flux component to the soil. Minor fluxes are related to annual litterfall of twigs, bud scales, fruits, and flowers (4 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) or

dendromass increments (0.7 kg Si ha<sup>-1</sup> yr<sup>-1</sup>). Although the Si pool size of testate amoebae is very small their relevance for internal Si cycling cannot be neglected. Due to relatively short generation times of idiosome-producing amoebae, e.g., Euglyphida with 2–16 days resulting in 12–130 generations per year (Schönborn, 1975, 1982; Lousier, 1984; Aoki et al., 2007), the annual biosilicification by idiosomes are in the order of the cumulative annual Si uptake by plants: using a conservative estimate of 15 generations per year we calculated an annual biosilicification of 17 kg Si ha<sup>-1</sup> yr<sup>-1</sup> (Si pool of living idiosomic taxa). The turnover rates of idiosomic Si pool must be much higher compared to phytolithic Si pool as can be deduced from (only) 47 % empty tests of total idiosomic Si pool. Consequently, testate amoebae can be regarded as a temporal Si pool on a very short, monthly timescale.

The total Si input with open land precipitation is rather low (< 1 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) which fits to the data from literature (Sommer et al., 2006; Cornelis et al., 2011b). The Si export by seepage equals 12 kg Si ha<sup>-1</sup> yr<sup>-1</sup> showing average silica concentrations of approx. 6 mg Si L<sup>-1</sup> (Table 5). The high temporal variability between years (CV = 50–64 %) can be related to varying seepage (Table 5), while temporal variability in silica concentrations were rather small (CV increase with depth: 4 → 11 %). Mean silica concentration increases only slightly from acid soil horizons (0.2 m, 0.7 m) to the calcareous parent material in 2.5 m (4.9 → 5.7 mg Si L<sup>-1</sup>,

**Table 4.** Pools and increments of plant biomass, Si concentrations (mass weighted mean values), Si pools in aboveground biomass, Si fluxes with increment of tree biomass; \*: calculated from litter fall of 05/2006–04/2010; \*\*: calculated from stem volume of 2010 or stem growth of 2006–2009; no. in brackets = std.dev. ( $n = 4$  yr).

	Biomass pools (t DM ha <sup>-1</sup> )	Litterfall/increment (t DM ha <sup>-1</sup> yr <sup>-1</sup> )	Si content (mg kg <sup>-1</sup> )	Si pools (kg ha <sup>-1</sup> )	Si fluxes (kg ha <sup>-1</sup> yr <sup>-1</sup> )
Leaves*	3.4	3.4 (0.30)	8952 (3101)	29.7	29.7 (7.5)
Bud scales*	0.5	0.5 (0.04)	2395 (431)	1.2	1.2 (0.3)
Beechnuts*	0.7	0.7 (0.80)	280 (252)	0.2	0.2 (0.3)
Fruit capsules*	1.6	1.6 (1.20)	913 (275)	1.4	1.4 (1.1)
Other*		0.1 (0.03)	10229 (2979)		0.6 (0.4)
Branches/twigs including bark*		0.4 (0.30)	1649 (183)		0.7 (0.5)
<i>Total litterfall</i>		6.7			33.7
Branch wood without bark**	24.4	0.4	17	0.4	0.01
Stem wood without bark**	196.1	3.1	17	3.4	0.05
Bark of branch wood**	2.0	0.0	2565	5.2	0.08
Bark of stem wood**	16.3	0.2	2565	41.9	0.59
<i>Dendromass increment</i>		3.7			0.70
<i>Aboveground biomass</i>	245.0			83.4	

Table 5). This increase goes along with an increase in the water-soluble Si fraction (Fig. 1).

We set the annual accumulation rate of Si in stem wood and bark (0.6 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) as the (annual) harvest export. Although not realized yet, at the end of a rotation period stem harvest leads to a complete Si export of this compartment. By adding this Si export to seepage losses we come up with a gross Si loss of 13 kg Si ha<sup>-1</sup> yr<sup>-1</sup>. Taking into account the inputs by deposition our biogeosystem reveals a net loss of 12 kg Si ha<sup>-1</sup> yr<sup>-1</sup>.

#### 4 Discussion

The low atmospheric Si input is in accordance with reported values from other forested biogeosystems (< 2 kg ha<sup>-1</sup> yr<sup>-1</sup>, Sommer et al., 2006; Cornelis et al., 2011b). Throughfall and stemflow can also be neglected in terms of Si fluxes. The small annual increase in Si stored in the vegetation (biomass increment = 0.7 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) is lower compared to reported data (European beech = 3.5 kg Si ha<sup>-1</sup> yr<sup>-1</sup>, Cornelis et al., 2010a) which might be explained by the lower forest growth of our mature beech stand (120 yr). The Si uptake by beech (35 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) and return flux by litterfall (34 kg Si ha<sup>-1</sup> yr<sup>-1</sup>) lies in the range of other European beech stands in temperate climates (19–47 kg Si ha<sup>-1</sup> yr<sup>-1</sup>; Pavlov, 1972; Bartoli and Souchier, 1978; Ellenberg et al., 1986; Cornelis et al., 2010a). We calculated an averaged passive Si-uptake of 17 kg Si ha<sup>-1</sup> yr<sup>-1</sup> as the product of vertically distributed mean Si concentrations and modeled transpiration flux (239 mm yr<sup>-1</sup>). When compared to the measured

**Table 5.** Water fluxes, Si concentrations, and Si fluxes in water (m = measurements, s = simulation); all fluxes are expressed as mean annual values for a four year period (5/2007–4/2011), number in brackets = std.dev ( $n = 4$  yr).

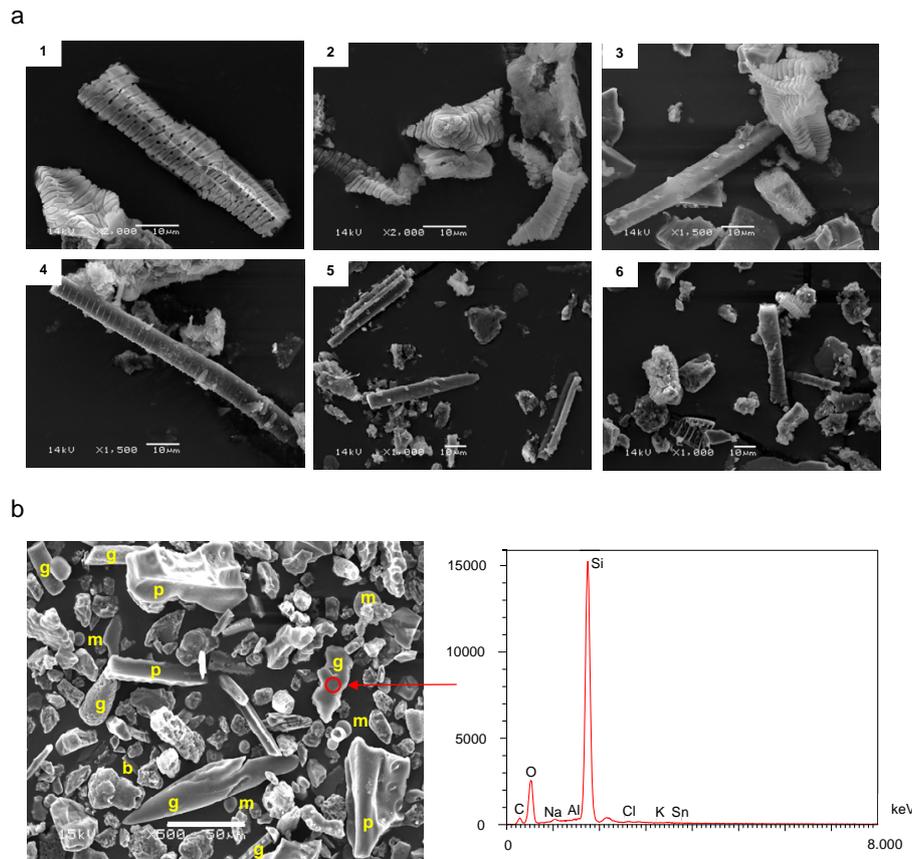
	Water flux (L m <sup>-2</sup> yr <sup>-1</sup> )	Si concentr. (mg L <sup>-1</sup> )	Si flux (kg ha <sup>-1</sup> yr <sup>-1</sup> )
Precipitation (m)	689 (195)	0.1 (0.03)	0.6 (0.2)
Leaching from canopy (m)			0.3 (0.1)
Canopy evaporation (s)	168 (28)		
Transpiration (s)	239 (24)		
Evapotranspiration (s)	476 (62)		
Throughfall (m)	491 (189)	0.1 (0.04)	0.8 (0.3)
Stemflow (m)	32 (13)	0.3 (0.03)	0.1 (0.0)
Stand precipitation (m)	523 (203)		0.9 (0.3)
Drainage from 20 cm (s)	316 (152)	4.9 (0.20)	15.4 (7.6)
Drainage from 70 cm (s)	262 (139)	5.3 (0.30)	13.9 (8.0)
<b>Drainage from 250 cm (s)</b>	<b>214 (133)</b>	<b>5.7 (0.60)</b>	<b>12.3 (7.9)</b>

Si-uptake of 34 kg Si ha<sup>-1</sup> yr<sup>-1</sup> an active uptake of 50 % of total uptake can be inferred.

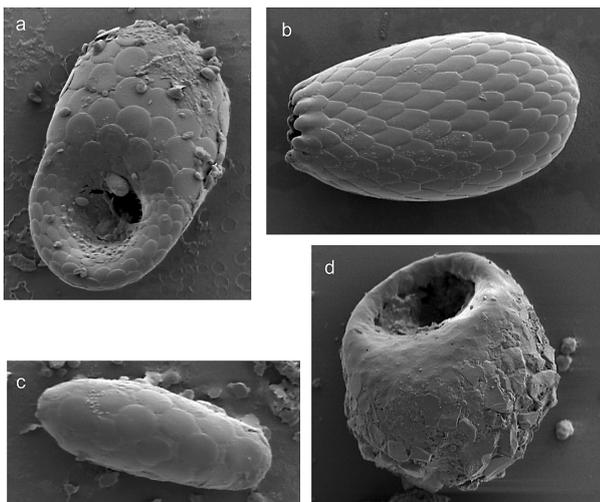
Generally, the magnitude of Si uptake at Beerenbusch surprises, when considering the very low content of weatherable minerals in the rooting zone (soils, sediments). In principal three processes might cause the silica concentrations observed: (i) quartz dissolution, (ii) weathering of silicates, and (iii) dissolution of the biogenic Si pool (non-steady state).

#### 4.1 Quartz dissolution and silicate weathering

Quartz has a water solubility of 1–7 mg L<sup>-1</sup> (36–250 μmol L<sup>-1</sup>) depending on particle size and temperature (Iler, 1979; Bartoli and Wilding, 1980; Dove, 1995). The



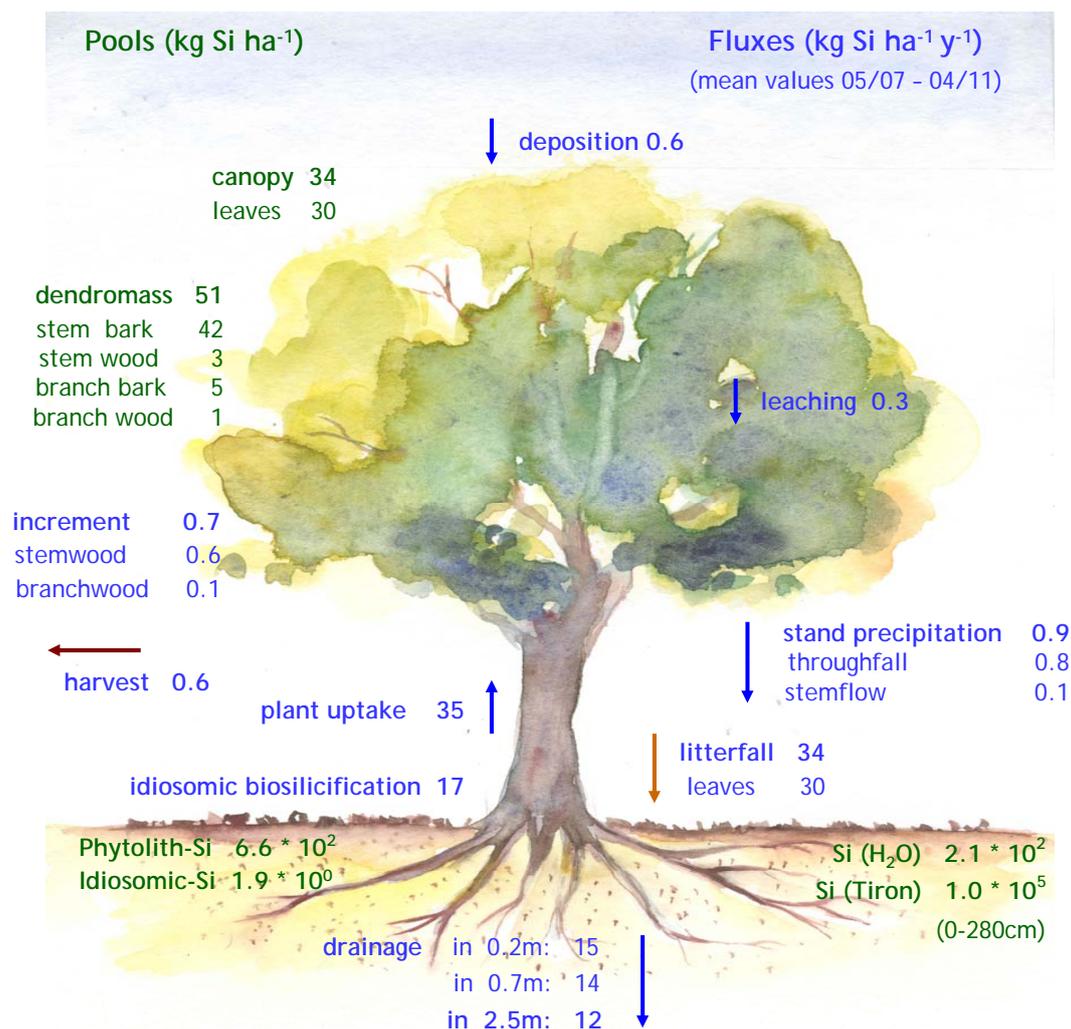
**Fig. 2.** (a) SEM micrographs of opal phytoliths in *Fagus sylvatica* litterfall (1, 2, 3 leaves, 4, 5 bud scales, 6 fruit capsules), scale bars at all micrographs = 10 μm; (b) SEM micrographs of opal phytoliths in AE horizon, 0–2 cm, scale bar = 50 μm (b = beech, p = pine, g = grass, m = moss); (c) element distribution of a grass opal phytolith (SEM-EDX).



**Fig. 3.** SEM micrographs of dominant testate amoebae; idiosomic taxa: (a) *Trinema complanatum* (length = 46 μm), (b) *Euglypha rotunda* (length = 49 μm), (c) *Trinema lineare* (length = 23 μm); xenosomic taxa: (d) *Centropyxis sphagnicola* (length = 65 μm).

lower value is more realistic for soils. This is supported by soil water data from quartz-rich, non-redoximorphic soils developed on quartzitic or granitic lithologies in the Black Forest (Podzols, Cambisols) showing silica concentrations always  $< 2.5 \text{ mg L}^{-1}$  (Sommer et al., 2006). Studies from tropical soils with absolute quartz dominance also confirm low silica concentrations in soil waters ( $< 1.2 \text{ mg Si L}^{-1}$ ; Cornu et al., 1998; Lucas, 2001; Patel-Sorentino et al., 2007; do Nascimento et al., 2008).

There are two reasons for observed lower in situ silica concentrations compared to lab experiments on pure phases: (i) lab studies on quartz dissolution kinetics mostly use samples ground to silt size which leads to fresh mineral surfaces and high surface: volume ratios compared to our site showing sand as the dominating particle size class. (ii) Surface coatings (Fe oxides, organic matter) protect quartz grains from intense dissolution, because of the reduced access of soil solution to quartz surfaces. Chemical interactions between soil solution and solid phase mainly take place at the surface of coatings. The latter occur at all soil depths as depicted from thin sections (Fig. 5, upper). Coatings show dark to light brown colors and an average thickness

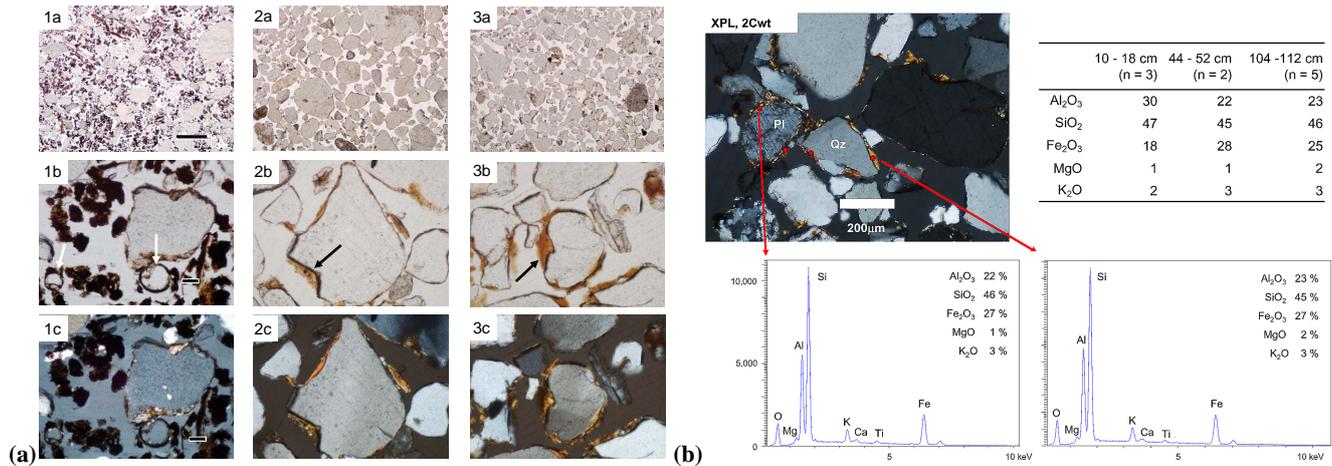


Painting: C. Höhn

**Fig. 4.** Si pools and fluxes at Beerenbusch (mean values 05/2007–04/2011) (painting: Cornelia Höhn, Müncheberg).

of 10–20 μm. From element distribution (EDX analysis on several coatings, Fig. 5 lower) two different mineral phases can be inferred – an iron bearing oxyhydroxide, most probably goethite, and kaolinite. Subsoil horizons show a higher percentage of Fe in the coatings (“ferri-argillans”, compare Chartres, 1987; Stoops, 2003; Li et al., 2008). Because feldspar and quartz coatings showed a similar chemical composition they cannot be interpreted as weathering rinds. Instead, a vertical redistribution of fine material by clay translocation (see macroscopic lamellae in 2Cwt) and podzolization (precipitation of organic Fe-Al-complexes), combined with a μm-scale horizontal redistribution of fines during desiccation (water films around grains) are the most probable explanations. From these findings we conclude that quartz dissolution will contribute only a minor fraction to the silica concentrations observed.

Weatherable silicates (mainly K-feldspars, few plagioclase) contribute 3 % (average) to the mineral assemblage of our soil and sediments. Due to the low content and a higher stability of orthoclase, we expected a comparative low contribution of feldspar weathering to DSi. Nevertheless, we checked feldspar weathering intensity by thin sections and SEM. In thin sections some feldspars have been found as part of compact multiminerall sand grains (Fig. S2 in Supplement), probably derived from glacier grinding of granites and subsequent glaciofluvial sorting processes. Here the accessible surface area for chemical weathering is very limited. Further, single feldspar grains appear only slightly weathered in SEM micrographs compared, e.g., to feldspars in White et al. (2008). Furthermore there is no trend with depth (Fig. 6). These findings are surprising, because the high input of organic acids by podzolization should enhance acid hydrolysis.



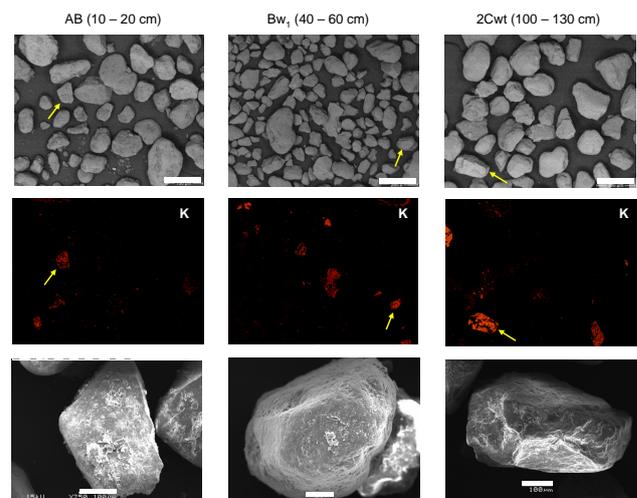
**Fig. 5.** (a) Micromorphological properties of soil horizons; microstructure (a, b = PPL, c = XPL) of selected soil horizons: (1) AE (0–2 cm), arrows point to testate amoebae; (2) Bw<sub>1</sub> (44–51 cm), arrow points to Fe-oxides/clay coating; (3) 2Cwt (104–112 cm), arrow points to Fe-oxides/clay coating; widths of photo: 1a, 2a, 3a = 6 mm; other = 0.6 mm. (b) SEM-EDX micrographs of cutans in 2Cwt (104–112 cm); average chemical composition (wt.-%) of coatings of different depths analysed by electron microprobe; number in brackets = no. of cutans analysed by EDX in each thin section.

Coatings on feldspar grains (Fig. 5) most probably explain this apparent contradiction as coatings reduce accessibility of soil water to feldspar surfaces, hence chemical reactions. Soil solution chemistry measured from 2001 to 2012 supports our SEM findings (mean concentrations at 20, 70, 250 cm in mg L<sup>-1</sup>): Neither Na (5.8, 5.5, 5.9) nor Cl (7.9, 7.4, 8.7) give evidence for a silicate weathering front. Instead, the depth trends of DOC (23.6, 15.1, 6.7 mg L<sup>-1</sup>), Al (1.5, 0.5, 0.03 mg L<sup>-1</sup>), and Fe (41, 12, 5 μg L<sup>-1</sup>) confirms podzolization to be the main pedogenic process. In summary, we conclude feldspar weathering to be of very low influence on DSi concentrations.

#### 4.2 Dissolution of biogenic Si pool

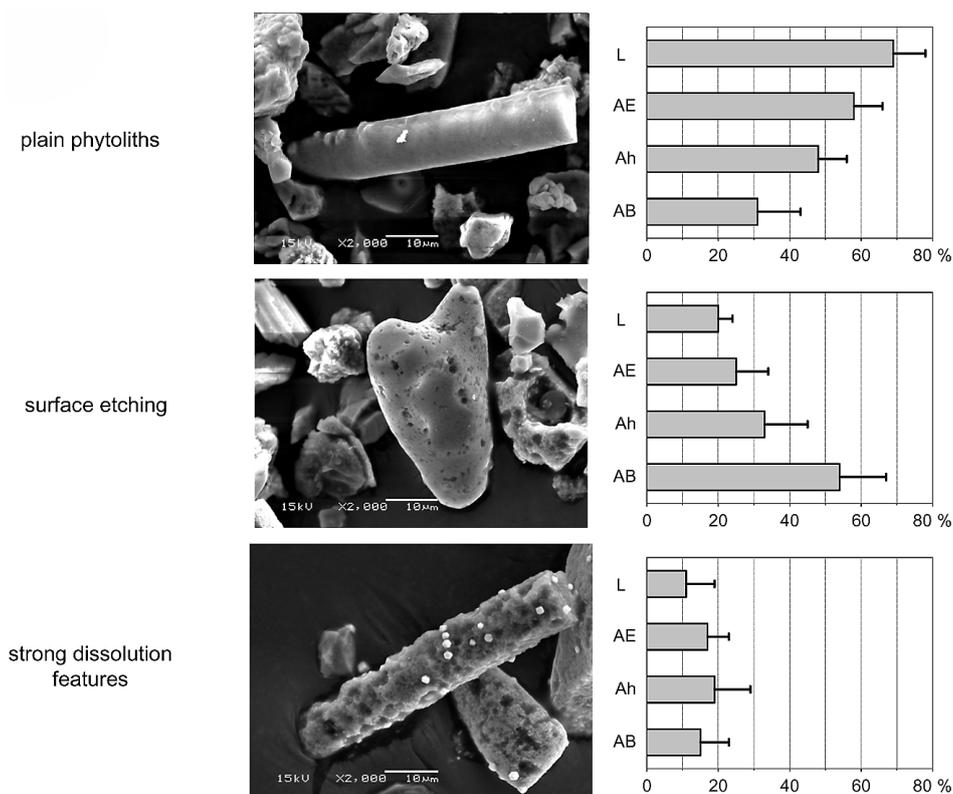
The biogenic Si pool contains the phytogenic and the zoogenic Si pool, both of which differ remarkably in dynamics and turnover rates. The annual biosilicification of idiosomic amoebae by binary fission adds up to 17 kg Si ha<sup>-1</sup> which is in the order of magnitude of the Si flux by litterfall, hence phytogenic silicification. On the other hand, the idiosomic Si pool of 2 kg Si ha<sup>-1</sup> is comparatively small. Empty tests comprise only 40 % (0.8 kg Si ha<sup>-1</sup>) of total idiosomic Si pool, while living amoebae make up 60 % (1.2 kg Si ha<sup>-1</sup>). This leads to our conclusion of a very high solubility of the idiosomic Si pool. Consequently, turnover rates are too high for interannual pool changes to become relevant for DSi exports.

Annual biosilicification by plants sums up to 35 kg Si ha<sup>-1</sup>, most of which is returned to the soil by litterfall. Compared to the phytolith Si pool of 660 kg Si ha<sup>-1</sup> this flux is relatively large, hence residence time is rather short (pool/flux = 19 yr). Therefore decadal changes of phytogenic



**Fig. 6.** SEM-micrographs and EDX element mapping (K) of soil material (< 2 mm) from three different depths; scale bars: upper row = 700 μm; lower row, left and right = 100 μm, middle = 50 μm; yellow arrows point to orthoclase mineral grains shown in lower row.

Si pools are relevant for DSi exports in principal. Early experiments on phytolith dissolution in distilled water showed silicic acid equilibrium concentrations of 2–10 mg Si L<sup>-1</sup> (Bartoli and Wilding, 1980) depending on plant species as well as surface area. These concentrations were higher than those determined for quartz (1 mg Si L<sup>-1</sup>), but lower than synthesised pure silica gels (56 mg Si L<sup>-1</sup>). The latter might be closer to solubilities of nm-sized phytogenic Si. Recent experiments on the reactivity of plant phytoliths in soil solutions have shown the solubility product to be close



**Fig. 7.** Categories and depth functions of phytolith dissolution (based on counting in SEM micrographs): fresh, plain phytoliths (upper), phytoliths showing clear signs of surface etching (middle), and phytoliths showing strong dissolution features (lower); horizontal bars = mean  $\pm$  SD ( $n = 10$  micrographs), total number of phytoliths counted: L = 462, AE = 459, Ah = 422, AB = 238.

to that of amorphous silica and up to one order of magnitude higher than those of clay minerals (Frayse et al., 2009, 2010). Half-life times of the studied phytoliths range from 10–12 yr ( $\text{pH} < 3$ ) to  $< 1$  yr ( $\text{pH} > 6$ ). From these studies it is concluded that phytoliths represent a very reactive Si pool in soil solutions which contributes substantially to DSi.

To check the status of phytolith dissolution in our soil we defined three classes of increasing dissolution and counted assigned phytoliths in SEM micrographs (Fig. 7). The percentage of plain phytoliths showing no signs of dissolution or surface etching significantly decreased from 69 % at the soil surface to 31 % in the AB horizon (10–20 cm). Simultaneously phytoliths showing slight surface etching increased to 54 %. Strongly dissolved phytoliths revealed a maximum of 19 % in Ah (2–10 cm), but no clear trend with depth. Combining these findings with (i) the observation of missing fresh beech phytoliths in soil horizons (see Sect. 3.2), and (ii) the parallel increase of water-soluble Si with phytolith Si pool (upper 25 cm), we regard dissolution of phytogenic Si as the most important driver of DSi concentrations, hence DSi exports.

The influence of phytogenic Si on DSi must be regarded even stronger as the total phytogenic Si pool in soils is definitely higher than the phytolith Si pool. Only the  $> 5 \mu\text{m}$  frac-

tions of soil horizons are quantified by phytolith separation procedure (see Sect. 2.3). Comparing the Si content of leaves calculated on basis of the phytolith content with measured Si content supports this consideration: the calculated Si content of  $2.4 \text{ g kg}^{-1}$  (Table 2a) only comprises 55 % of the measured Si content of  $4.4 \text{ g kg}^{-1}$  (Table 2a). BSi in the  $< 2 \mu\text{m}$  fraction probably explains this difference, because this fraction is lost during phytolith extraction procedure. In their early work Wilding and Drees (1971) quantified 50 % of total leaf opal (*Fagus grandifolia*) in the  $< 2 \mu\text{m}$  fraction, another 22 % in the 2–5  $\mu\text{m}$  fraction. Watteau and Villemin (2001) provided evidence for nm-size phytogenic Si granules in leaves and soils. This fraction has a higher reactivity in soil solutions compared to phytoliths due to its higher surface : volume ratio.

Our conclusion about BSi as the main driver for the DSi observed is indirectly supported by process-based modeling of the Si cycle in a forest biogeosystem (Gerard et al., 2008). In a Cambisol from volcanic tuff – containing less quartz (30 %) and higher percentages of clays and weatherable minerals (K-feldspars) compared to our site – the BSi still account for 60 % of DSi after all. Finally, as grasses – which are now absent at the beech stand – contribute a major part to recent phytolith pool, we concluded the phytogenic Si pool

not to be at steady state, but transient state: the continuous decomposition of the relictic phytogenic Si pool is actually not compensated by an equivalent upbuilding.

## 5 Conclusions

Our studied forest biogeosystem exhibits surprisingly high DSi concentrations and exports compared to the very low content of weatherable minerals in soil and sediments. From our findings we excluded geochemical weathering processes as a major control on DSi, but concluded a strong biogenic footprint on DSi. The disappearance of Si-rich grasses during canopy closure as well as pine logging 30 yr ago seems to be the ultimate reason for the phenomena observed. The related phytogenic Si pool is in disequilibrium with recent vegetation and is dissolving successively. This has to be confirmed by Si isotope analysis of the different phases like it has been done in Cornelis et al. (2010b), Engström et al., (2010), Opfergelt et al. (2010), or Steinhöfel et al. (2011). Furthermore, phytolith dissolution experiments on grasses are needed in future research to understand the dissolution kinetics of Si-rich species, like *Calamagrostis epigejos*.

We regard our study as another example highlighting the importance of disturbances and perturbations in Si cycling (Ittekkot et al., 2006; Laruelle et al., 2009; Struyf et al., 2010; Clymans et al., 2011; Struyf and Conley, 2012). Our study complements the conceptual model of Struyf et al. (2010). It adds vegetation changes by forest management, i.e., without dramatic LUC, as another driver for transient state Si cycling. Future research on Si cycling should explicitly consider decadal transient states and their main drivers.

**Supplementary material related to this article is available online at: <http://www.biogeosciences.net/10/4991/2013/bg-10-4991-2013-supplement.pdf>.**

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