

# Nitrous Oxide Emissions from Open-Lot Cattle Feedyards: A Review

Heidi M. Waldrip,\* Richard W. Todd, David B. Parker, N. Andy Cole, C. Alan Rotz, and Kenneth D. Casey

## Abstract

Nitrous oxide ( $\text{N}_2\text{O}$ ) emissions from concentrated animal feeding operations, including cattle feedyards, have become an important research topic. However, there are limitations to current measurement techniques, uncertainty in the magnitude of feedyard  $\text{N}_2\text{O}$  fluxes, and a lack of effective mitigation methods. The objective of this review was to assess  $\text{N}_2\text{O}$  emission from cattle feedyards, including comparison of measured and modeled emission rates, discussion of measurement methods, and evaluation of mitigation options. Published annual per capita flux rates for beef cattle feedyards and open-lot dairies were highly variable and ranged from 0.002 to 4.3 kg  $\text{N}_2\text{O}$  animal<sup>-1</sup> yr<sup>-1</sup>. On an area basis, published emission rates ranged from 0 to 41 mg  $\text{N}_2\text{O}$  m<sup>-2</sup> h<sup>-1</sup>. From these studies and Intergovernmental Panel on Climate Change emission factors, calculated daily per capita  $\text{N}_2\text{O}$  fluxes averaged  $18 \pm 10$  g  $\text{N}_2\text{O}$  animal<sup>-1</sup> d<sup>-1</sup> (range, 0.04–67 g  $\text{N}_2\text{O}$  animal<sup>-1</sup> d<sup>-1</sup>). This variation was due to inconsistency in measurement techniques as well as irregularity in  $\text{N}_2\text{O}$  production and emission attributable to management, animal diet, and environmental conditions. Based on this review, it is clear that the magnitude and dynamics of  $\text{N}_2\text{O}$  emissions from open-lot cattle systems are not well understood. Further research is required to quantify feedyard  $\text{N}_2\text{O}$  fluxes and develop cost-effective mitigation methods.

## Core Ideas

- The magnitude of published  $\text{N}_2\text{O}$  fluxes from open-lot cattle systems was highly variable.
- $\text{N}_2\text{O}$  fluxes varied with measurement methods, management, and environmental conditions.
- Methods to mitigate feedyard  $\text{N}_2\text{O}$  emissions are available but have yet to be thoroughly evaluated.
- Improved animal performance and reducing N intake are best options to reduce feedyard  $\text{N}_2\text{O}$  losses.

**N**ITROUS OXIDE ( $\text{N}_2\text{O}$ ) is a greenhouse gas (GHG) with a 100-yr global warming potential 265 to 298 times greater than carbon dioxide ( $\text{CO}_2$ ) and an atmospheric lifetime of ~114 yr (Myhre et al., 2013; USEPA, 2013). Current tropospheric  $\text{N}_2\text{O}$  concentrations (326 ppb) are lower than that of other GHGs ( $\text{CO}_2$ , 395 ppm; methane [ $\text{CH}_4$ ], 1893 ppb) (Blasing, 2014); however, small  $\text{N}_2\text{O}$  increases have major impact on climate change due to the ability of  $\text{N}_2\text{O}$  to trap solar energy (i.e., the “greenhouse effect”). In 2007, approximately 60% of global  $\text{N}_2\text{O}$  emissions were attributed to fertilized cropland and livestock production (Smith et al., 2007).

Approximately 51% of anthropogenic  $\text{N}_2\text{O}$  produced in the United States originates from cropland; however, the USEPA estimated that 17.3 Tg  $\text{CO}_2$  equivalents ( $\text{CO}_2$  eq) of  $\text{N}_2\text{O}$  were derived from livestock manure in 2015, of which 8 Tg  $\text{CO}_2$  eq were attributed to beef cattle (USEPA, 2015). The estimated contributions of major livestock sectors to manure-derived  $\text{N}_2\text{O}$  in the United States are presented in Fig. 1a, where the impact of beef and dairy industries is clear.

In 2009, the USEPA proposed a mandatory reporting rule for livestock facilities that annually generate 25,000 t of  $\text{CO}_2$  eq as  $\text{CH}_4$  and/or  $\text{N}_2\text{O}$  from manure (USEPA, 2009). The estimated threshold cattle population for generation of such a large quantity of GHG at a single beef operation was 88,923 head. Approximately 80 to 90% of beef cattle in the United States are finished in open-lot feedyards with holding capacities of more than 1000 animals: in 2007 a total of 774 feedyards had capacities greater than 2500 cattle (USDA–NASS, 2007). Many of the top 20 beef producing companies in the United States have

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**Abbreviations:** AOA, ammonia-oxidizing archaea; AOB, ammonia-oxidizing bacteria; CAFO, concentrated animal feeding operation; CP, crude protein; CTC, chlortetracycline; DCD, dicyandiamide; DEA, denitrification enzyme activity; DMPP, 3,4-dimethylpyrazole phosphate; DNDC, Denitrification and Decomposition Model; EF, emission factor; EC, electrical conductivity; GC, gas chromatograph; HIP, “hole-in-the-pipe”; IFSM, Integrated Farm Systems Model; IPCC, Intergovernmental Panel on Climate Change; MM, micrometeorological method; NA, nitrification activity; NBPT, *N*-(*n*-butyl) thiophosphoric triamide; NFT-NSS, non-flow through/non-steady state; NOB, nitrate-oxidizing bacteria; OM, organic matter; WDGS, wet distillers grain plus solubles.

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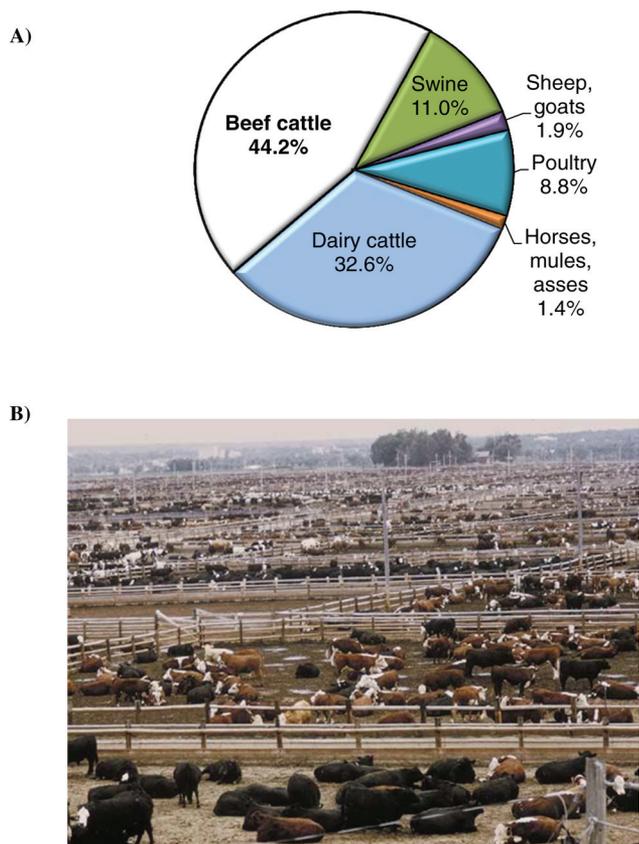
J. Environ. Qual. 45:1797–1811 (2016)  
doi:10.2134/jeq2016.04.0140

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Received 12 Apr. 2016.

Accepted 2 Aug. 2016.

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**Fig. 1. (A) Estimated contributions of major livestock sectors to manure-derived  $N_2O$  in the United States. Data adapted from USEPA (2013). (B) Commercial beef cattle feedyard in the Texas Panhandle. Finishing cattle are typically housed in earthen-surfaced pens for approximately 180 d at a density of approximately  $15\text{ m}^2\text{ animal}^{-1}$ .**

feedyards with capacities  $>100,000$  cattle (NCBA, 2015). Most large feedyards are located in the southern High Plains of Texas, Nebraska, and Kansas (USDA-ERS, 2012). The number of large feedyards and open-lot dairies is continually expanding in the southern High Plains (USDA-NASS, 2015). An example of a large beef feedyard in Texas is presented in Fig. 1b.

## Regulating $N_2O$ Emissions from Cattle Feedyards

Approximately 23 million cattle are fed annually in US feedyards with capacities  $>1000$  animals (USDA-NASS, 2013). Most are housed in open-lot, soil-surfaced pens. The manure (a mixture of feces, urine, soil, dropped feed, and scurf) that accumulates in pens is heterogeneous and dynamic, consisting of both freshly excreted and older material that is continually changing compositionally. Manure is a complex mixture of  $H_2O$ , organic matter (OM), nutrients, microbes, enzymes, minerals, and pharmaceutical residues. Manure undergoes numerous (bio)chemical processes, including mineralization, hydrolysis, nitrification, denitrification, and fermentation. To date, it is assumed that most feedyard  $N_2O$  is produced via nitrification and denitrification, although the complex nature of the manure pack itself, the high spatial variability of manure characteristics within pens, and interactions among environmental and management factors complicate this interpretation (Cole et al., 2009a,b; Waldrip et al., 2015a,b,c,d).

Environmental issues occur when cattle are housed in high densities, largely due to manure accumulation within a relatively small area. In general, less than 20% of the nitrogen (N) consumed by cattle is converted into animal product, with the remainder excreted in urine and feces (Cole and Todd, 2009; Mosier et al., 1998). Imbalances between feedyard N imports (e.g., purchased feed and animals) and exports (e.g., animal product and manure) can cause N losses to air and water. This review focuses on  $N_2O$  emissions; however, Waldrip et al. (2015a,b) recently conducted detailed reviews on general feedyard N sustainability and ammonia ( $NH_3$ ) emission.

In the United States, concentrated animal feeding operations (CAFOs) have been subjected to regulation, reflecting heightened concern over pollution from livestock. Manure is estimated to be the source of  $\sim 17\%$  of N entering US waters (Smith et al., 1997). Clean Water Act (33 U.S.C. §1251 et seq. [1972]; USEPA, 2002) regulations require that CAFOs with a National Pollutant Discharge Elimination System permit develop and implement management plans for fields that receive manure (USEPA, 1990). Such a plan, which must meet USDA Natural Resources Conservation Service (NRCS) standards, limits nutrient application to land and specifies control measures to prevent nutrient losses. However, implementing and enforcing feedyard adoption of nutrient management plans may be cost prohibitive. The USDA Economic Research Service estimates that reductions in net returns in the livestock and poultry sectors due to regulation would be about \$1.4 billion per year, with subsequent economic welfare decline for producers and consumers by almost \$2 billion per year (Ribaldo et al., 2003). A benefit of regulation would be improved air, but at a cost to both producers and consumers.

Organic and inorganic N in manure is converted to forms that are readily transported by hydrologic and atmospheric processes. Focusing strictly on one issue, such as nitrate ( $NO_3^-$ ) leaching, could increase emissions of  $N_2O$ ,  $NH_3$ , or other gaseous N compounds involved in environmental degradation. As an example, a market for  $N_2O$  offsets may reduce total N emission to air but increase N losses to water. Pollution swapping tradeoffs complicate development of best management practices to prevent N losses.

Quantification of feedyard  $N_2O$  emission is needed for inventory purposes and to estimate the impact of cattle production on the environment. Programs, such as NRCS's Conservation Innovation Grants (USDA-NRCS, 2002), may provide funding for CAFO to initiate measures to mitigate N losses; however, it is unclear to what extent feedyard operators have sought funding for  $N_2O$  mitigation.

## $N_2O$ from Beef Cattle Feedyards

Nitrous oxide losses from open-lot feedyards are typically very small compared with emissions of  $NH_3$  and  $N_2O$  from agricultural soils and  $NH_3$  emission from animal housing. More compelling aspects of controlling and understanding feedyard  $N_2O$  are increased atmospheric GHG concentrations and potential CAFO reporting and regulation standards. Beef cattle researchers and producers have traditionally focused on increasing production efficiency, with less emphasis on long-term consequences; this approach may need revising to preserve economic viability of the beef industry while maintaining the environment.

## N<sub>2</sub>O Formation Processes

A substantial body of work exists on N<sub>2</sub>O formation and emission from soils, which was reviewed by Butterbach-Bahl et al. (2013). Less is known about manure-derived N<sub>2</sub>O, although some studies have analyzed N<sub>2</sub>O emission and related processes during manure composting or after soil application (Fortuna et al., 2012; Maeda et al., 2010). Various biotic and abiotic mechanisms produce N<sub>2</sub>O from soils, including denitrification, nitrifier-denitrification, nitrification, coupled nitrification–denitrification, dissimilatory nitrate ammonification, chemical decomposition, and chemodenitrification (Butterbach-Bahl et al., 2013). It is assumed that manure-derived N<sub>2</sub>O is primarily produced through microbial nitrification and denitrification (Jungbluth et al., 2001; Maeda et al., 2010). There is increasing evidence that abiotic N<sub>2</sub>O production via hydroxylamine (NH<sub>2</sub>OH) oxidation and chemodenitrification is important in soils (Zhu-Barker et al., 2015). To our knowledge, no research has assessed the role of abiotic processes in manure-derived N<sub>2</sub>O emission.

### Nitrification

Nitrification is the aerobic oxidation of reduced N (usually ammonium [NH<sub>4</sub><sup>+</sup>] or NH<sub>3</sub>) by ammonia-oxidizing bacteria (AOB) and ammonia oxidizing archaea (AOA) to nitrite (NO<sub>2</sub><sup>-</sup>) (Fig. 2). Nitrite is then oxidized by nitrate-oxidizing bacteria (NOB) to NO<sub>3</sub><sup>-</sup>. Fungi and other heterotrophic nitrifiers also produce NO<sub>3</sub><sup>-</sup> from organic material (Laughlin et al., 2009). In these processes, nitric oxide (NO) and N<sub>2</sub>O are gaseous intermediates that may be released from incomplete reactions. The formation of NO<sub>2</sub><sup>-</sup> requires two enzymes: ammonia monooxygenase (AMO; *amoA*) and hydroxylamine oxidoreductase (Norton, 2008). Reduction of NO<sub>2</sub><sup>-</sup> to NO<sub>3</sub><sup>-</sup> is performed by nitrite oxidoreductase. The initial oxidation of NH<sub>4</sub><sup>+</sup> to NO<sub>2</sub><sup>-</sup> is usually the rate-limiting step for nitrification (Gruber and Galloway, 2008); therefore, chemical compounds that block this reaction are one method to reduce N<sub>2</sub>O emissions (Dalal et al., 2003).

Current understanding of manure nitrification is largely derived from agricultural soils. This notion may be invalid due to soil and manure differences in OM and N contents and physicochemical properties. Nitrification potential, an estimate of the maximum capacity of microbes to transform NH<sub>4</sub><sup>+</sup> into NO<sub>3</sub><sup>-</sup>, has been determined on manure-amended soils (Fortuna et al., 2012; Laanbroek and Gerards, 1991). The quantitative polymerase chain reaction technique has elucidated the nitrifier

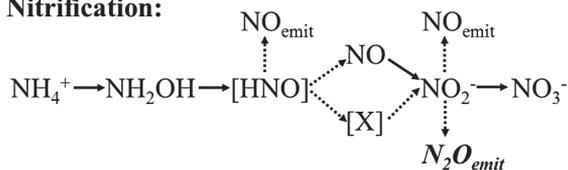
community structure in manure, allowing for direct assessment of *amoA* and AOA- and AOB-specific 16S rRNA gene copies. Yamada et al. (2013) found that AOA were the predominant nitrifiers during the early stages of beef manure composting. Rice et al. (2007) characterized the microbial community of beef feedyard manure, where both community structure and population numbers of AOB and NOB differed both with season and depth within the manure pack.

In soils, nitrification is controlled by oxygen and H<sub>2</sub>O content, temperature, and NH<sub>4</sub><sup>+</sup>/NH<sub>3</sub> availability (Norton, 2008). Recent research indicates that the same may be true for beef manure (Ayadi et al., 2015). The optimum H<sub>2</sub>O content for soil nitrifiers is about 60% of available pore space (Brady and Weil, 2002), but studies on H<sub>2</sub>O content effect on manure nitrification rates are sparse. Waldrip et al. (2015d) saw that N<sub>2</sub>O emission from feedyard manure at 20% saturation was 41% higher than at 80% saturation, suggesting enhanced nitrifier activity when manure was dry. In contrast, Miller and Berry (2005) reported negligible N<sub>2</sub>O emission from incubated soil/manure mixtures (25% soil and 75% beef cattle manure), despite variations in H<sub>2</sub>O content. Further studies are warranted to determine the effect of H<sub>2</sub>O content on N<sub>2</sub>O emissions from feedyards. Soil nitrification rates increase with temperature, with an optimum between 27 and 35°C, but are reduced at temperatures <10°C (Sabey et al., 1956). Similarly, Ayadi et al. (2015) observed that potential nitrification activity in packed beef manure was 120% higher at 40°C than at 10°C.

Soil nitrifier community structure and nitrification rates are influenced by N availability (Fortuna et al., 2012). For example, AOB populations tended to increase with increased NH<sub>4</sub><sup>+</sup> concentrations in N-rich grassland soils (Di et al., 2010), whereas AOA were more predominant relative to AOB under N-limited conditions (Hatzenpichler, 2012). Specific activities of AOB are generally greater than AOA; however, AOA activity can be higher under N-limited conditions (Herrmann et al., 2008; Taylor et al., 2010). Fortuna et al. (2012) used quantitative polymerase chain reaction on a diverse soil set amended with dairy slurry, where AOB copy numbers were an order of magnitude greater than AOA. There was a tendency for AOB, but not AOA, to increase with slurry application; however, results were inconsistent among the soils studied. Overall, soil nitrification potential was related to AOB copy numbers and total organic C and total N concentrations.

Little is known about the fungal contribution to nitrification in manure. Rice et al. (2007) saw low fungal diversity in the wet-pack, the dense layer of accumulated pen manure that has a higher H<sub>2</sub>O content than overlying manure (Cole et al., 2009b), and proposed that fungal nitrification could be limited by bacterial competition for substrate and low O<sub>2</sub> concentrations. Maeda et al. (2011) pinpointed a need for research on fungal N<sub>2</sub>O production in composts. Antibiotics are commonly used at subtherapeutic levels in commercial feedyards to promote weight gain and improve animal health. Chlortetracycline (CTC), a broad-spectrum antibiotic, is active against a range of gram-positive and gram-negative bacteria and protozoa and is administered to cattle for disease prevention; however, approximately 75% of CTC administered to cattle may be excreted (Elmund et al., 1971). Monensin sodium (Rumensin, Elanco Animal Health) is an ionophore used to control coccidiosis and rumen acidosis and

### Nitrification:



### Denitrification:

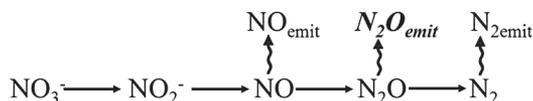


Fig. 2. Pathways of nitrification and denitrification leading to N<sub>2</sub>O emission from soils and manures. Dashed lines and square brackets indicate incompletely understood processes and intermediates, respectively.

to improve beef cattle feed efficiency. According to the manufacturer, estimated monensin sodium excretion rates were  $9.5 \text{ mg kg}^{-1}$  manure when supplied to finishing cattle at high levels ( $480 \text{ mg animal}^{-1} \text{ d}^{-1}$ ) (Elanco, 2005). Tylosin phosphate (Tylan, Elanco Animal Health) is a macrolide antibiotic used to treat pneumonia and foot rot, promote growth, prevent liver abscesses, and control other common cattle diseases. Manure from beef cattle that received  $11 \text{ mg d}^{-1}$  tylosin phosphate contained concentrations of  $107 \text{ mg tylosin phosphate kg}^{-1}$  manure (Amarakoon et al., 2016).

Antibiotic residues may reduce the contribution of bacterial-derived  $\text{N}_2\text{O}$ , causing fungi to predominate  $\text{N}_2\text{O}$  production on feedyards. Hao et al. (2011) measured  $\text{N}_2\text{O}$  emissions from composting manure where cattle were fed diets containing CTC, CTC plus sulfamethazine, or tylosin phosphate. These researchers noted that  $\text{N}_2\text{O}$  from compost made from manure from cattle treated with CTC/sulfamethazine, CTC, and tylosin phosphate tended to be greater than that from untreated cattle, but the results were inconsistent for the 2 yr of the study period. To date, no comprehensive study has assessed the role of fungi on feedyard  $\text{N}_2\text{O}$  emissions.

In contrast to soils, nitrification may not be a predominant process leading to  $\text{N}_2\text{O}$  emissions from feedyard manure because some common nitrifiers are sensitive to high  $\text{NH}_4^+/\text{NH}_3$  concentrations. Suwa et al. (1994) found that  $\text{NH}_3$  oxidizers in activated sewage sludge were inhibited when  $\text{NH}_4^+$  concentrations were greater than  $5 \text{ mmol L}^{-1}$  ( $90 \text{ mg NH}_4^+ \text{ L}^{-1}$ ). Koper et al. (2010) reported nitrification inhibition in soils amended with dairy manure compost. In contrast, Antileo et al. (2002) observed nitrifier acclimatization to  $\text{NH}_3\text{-N}$  concentrations as high as  $13 \text{ mg L}^{-1}$  in sewage sludge. Typical  $\text{NH}_3/\text{NH}_4^+\text{-N}$  concentrations in Texas feedyard pens averaged approximately  $6755 \text{ mg kg}^{-1}$  in fresh urine spots and  $2281 \text{ mg kg}^{-1}$  in nonurine spots (Cole et al., 2009a,b). It is unknown how these high  $\text{NH}_3/\text{NH}_4^+$  concentrations affect nitrification in feedyards.

### Denitrification

Denitrification generally occurs under anaerobic and microaerophilic conditions, where  $\text{NO}_3^-$  is reduced to  $\text{N}_2$  by a variety of microbial groups, including heterotrophic and autotrophic bacteria, Archea, Proteobacteria, and eukaryotic fungi (Zumft, 1997). The denitrification pathway involves the sequential action of dissimilatory  $\text{NO}_3^-$  reductase, dissimilatory  $\text{NO}_2^-$  reductase, NO reductase, and  $\text{N}_2\text{O}$  reductase (Fig. 2). Although manure denitrification has been studied, as with nitrification, most understanding comes from the considerable body of soil denitrification research. In both soils and manures, denitrification rates are largely controlled by temperature, pH,  $\text{H}_2\text{O}$ , and  $\text{NO}_3^-$  contents, and  $\text{N}_2\text{O}$  reductase activity (Galbally, 1989; Tsutsui et al., 2013; Willers et al., 1993).

Bremner and Shaw (1958) reported that soil denitrification rates increased with temperatures from 2 to  $25^\circ\text{C}$ . Optimal soil denitrification temperatures range from 25 to  $45^\circ\text{C}$  (Brady and Weil, 2002; Lensi and Chalamet, 1982). Woodbury et al. (2001) measured denitrification enzyme activity (DEA) and nitrification activity (NA) in feedyard manure collected from different depths (loose surface manure, manure pack at depths of 0–0.1 and 0.10–0.20 m) and locations within feedyard pens (near feedbunk, manure mounds, and near downgradient fenceline). The highest DEA rates were in loose surface manure near the

downgradient fenceline ( $132 \text{ mmol L}^{-1} \text{ g}^{-1} \text{ h}^{-1}$ ), whereas NA was greatest in the 0- to 0.1-m depth of the manure pack at the base of manure mounds ( $462 \text{ mmol L}^{-1} \text{ g}^{-1} \text{ h}^{-1}$ ). Thus, spatial variability in pen manure plays a role in determining the predominant mechanism producing feedyard  $\text{N}_2\text{O}$ .

The catabolism of fecal carbohydrates produces residual fatty acids that create anaerobic conditions when oxidized (Paul and Beauchamp, 1989). Nielsen et al. (1996) reported that high microbial respiration rates also cause anaerobic conditions in manure. High bulk density ( $\sim 1.7\text{--}1.9 \text{ g cm}^{-3}$ ) and low air and  $\text{H}_2\text{O}$  permeability ( $\sim 1.9 \times 10^{-6}$  to  $6.2 \times 10^{-7} \text{ cm s}^{-1}$ ) contribute to anoxic or anaerobic conditions favorable for denitrification in feedyard pens (McCullough et al., 2001; Mielke and Mazurak, 1976). However, these same conditions can inhibit diffusion of  $\text{NH}_3$  and  $\text{N}_2\text{O}$  from the manure pack to the atmosphere (Williams et al., 2008).

Substrate pH influences whether  $\text{N}_2\text{O}$  or  $\text{N}_2$  is the final denitrification product. In soil,  $\text{N}_2\text{O}$  is the major product at  $\text{pH} < 7$ , whereas  $\text{N}_2$  predominates at  $\text{pH} > 8$  (Simek and Cooper, 2002; Wijler and Delwiche, 1954). Feedyard manure pH usually ranges between 7.1 and 8.4 and increases with urine application (Cole et al., 2009a,b; Ward et al., 1978); thus,  $\text{N}_2$  should be the primary denitrification product. However, pH influence on denitrification is complicated by high salt concentrations (electrical conductivity [EC],  $\sim 1.0$  vs.  $0.3 \text{ dS m}^{-1}$  for soil underlying feedyard pens) (Cole et al., 2009a), which can inhibit denitrification (Woodbury et al., 2001). Further study is needed to elucidate the interactions between pH and EC on feedyard N transformations.

### Interaction between Nitrification and Denitrification in $\text{N}_2\text{O}$ Production

The particular pathway that produces  $\text{N}_2\text{O}$  in soils is largely controlled by  $\text{H}_2\text{O}$  content, where nitrification predominates when water-filled pore space is  $< 60\%$  (Pihlatie et al., 2004). Pihlatie et al. (2004) reported that nitrification in peat soils accounted for 76, 42, and 22% of total  $\text{N}_2\text{O}$  production at 60, 80, and 100% water-filled pore space, respectively. Denitrification rates increase exponentially with soil  $\text{H}_2\text{O}$  content  $> 55\%$  and are highest near saturation (Parton et al., 2001). The same trend appears to be true for manure because nitrification was the source of  $\text{N}_2\text{O}$  from dairy manure at high redox potential (low water content) and denitrification predominated at low redox potential (Brown et al., 2000).

Evidence links nitrification and denitrification to total  $\text{N}_2\text{O}$  production in soil (Nielsen et al., 1996). Firestone and Davidson (1989) conceptualized the “hole-in-the-pipe” (HIP) soil model that relates production of NO and  $\text{N}_2\text{O}$  to N availability, where the specific production mechanism is a function of  $\text{H}_2\text{O}$  content. In the HIP model,  $\text{N}_2\text{O}$  escapes from the theoretical holes in the pipe: the size of the holes and extent of  $\text{N}_2\text{O}$  emission is primarily based on soil  $\text{H}_2\text{O}$  content. If HIP is valid for feedyard manure, high spatial and temporal variability in pen manure  $\text{H}_2\text{O}$  and N content complicates determination of the primary  $\text{N}_2\text{O}$  production mechanism and complicates development of effective mitigation strategies to control feedyard  $\text{N}_2\text{O}$  emission.

Increasing manure  $\text{H}_2\text{O}$  content could slow nitrification, accelerate denitrification, or result in a combination of these processes. Petersen et al. (2004) and Monaghan and Barraclough (1992) noted little accumulation of  $\text{NO}_3^-$  in pasture soil during

the first 2 to 5 d after urine application. Limmer and Steele (1982) saw increased soil  $\text{NO}_3^-$  concentrations 6 to 11 d after urine application, with decreased denitrification potential after 21 d. In contrast, Petersen et al. (2004) noted increased soil  $\text{NO}_3^-$  content that accounted for 17 to 23% of applied N just 4 d after urine application. Monaghan and Barraclough (1992) hypothesized that the combined inhibitory effects of elevated pH, high  $\text{NH}_3$  concentrations, and osmotic stress reduced nitrification rates in pasture urine spots. Similar conditions exist in feedyard pens; however, high stocking densities result in more frequent urine application within a limited area than in grazing systems.

One example of  $\text{H}_2\text{O}$  content effects on feedyard  $\text{N}_2\text{O}$  emission comes from Aguilar et al. (2011), who used static flux chambers on a commercial Kansas feedyard and noted that flux rates and magnitude of  $\text{N}_2\text{O}$  emissions varied greatly depending on pen conditions. The greatest ( $25 \text{ mg N}_2\text{O m}^{-2} \text{ h}^{-1}$ ) and most variable emission rates were from moist/muddy spots, as compared with dry spots or flooded areas. Moist/muddy spots are considered “hot spots” (localized micro-sites with conditions favoring microbial activity) for  $\text{N}_2\text{O}$  production and emission (Matthews et al., 2010; Woodbury et al., 2001). Excessive  $\text{H}_2\text{O}$  content (e.g., flooded conditions) reduces redox potential and decreases air-filled pore space, which prevents gas diffusion and subsequent emissions. Temperature is an important factor in  $\text{N}_2\text{O}$  production due to its influence on OM decomposition and N mineralization (Saggar et al., 1998). Woodbury et al. (2001) proposed that the loose, well-aerated surface manure in feedyard pens was highly organic and favorable for both organic N mineralization and nitrification, whereas compacted manure at lower depths contained anaerobic zones promoting denitrification. Thus, feedyard  $\text{N}_2\text{O}$  is likely a result of combined denitrification and nitrification processes, similar to observations from compost and soil (Ma et al., 2008; Maeda et al., 2010).

#### Enteric $\text{N}_2\text{O}$ Production

In addition to manure-derived  $\text{N}_2\text{O}$ , some enteric  $\text{N}_2\text{O}$  may be produced by ruminant animals (Kaspar and Tiedje, 1981; Petersen et al., 2015; Stackhouse et al., 2011). The contribution of enteric  $\text{N}_2\text{O}$  to the environmental footprint of cattle has not been firmly established. Kaspar and Tiedje (1981) suggested that ruminal microbes produce small amounts of  $\text{N}_2\text{O}$  as a byproduct of dissimilatory  $\text{NO}_3^-$  reduction to  $\text{NH}_4^+$ . In contrast, Reynolds et al. (2010) reported no enteric  $\text{N}_2\text{O}$  emissions from cattle in respiration chambers.

To date, enteric  $\text{N}_2\text{O}$  has been largely ignored in models and GHG inventory calculations. Petersen et al. (2015) saw enteric  $\text{N}_2\text{O}$  emissions ranging from 0.4 to 67  $\text{g CO}_2\text{eq kg}^{-1}$  dry matter (DM) intake from dairy cows in respiration chambers. Stackhouse et al. (2011) reported that beef calves and steers emitted from 0.7 to 19.9  $\text{mg N}_2\text{O h}^{-1}$  in respiration chambers, with higher values emitted by larger animals. In a study to evaluate the effects of supplemental dietary  $\text{NO}_3^-$  on  $\text{CH}_4$  emissions from dairy cows,  $\text{N}_2\text{O}$  ranged from 0.4 to 67.2  $\text{g CO}_2\text{eq kg}^{-1}$  DM intake (Petersen et al., 2015). Supplementation at 21  $\text{g NO}_3^- \text{ kg}^{-1}$  DM decreased  $\text{CH}_4$  emission by 19 to 47%, but  $\text{N}_2\text{O}$  increased by as much as 66  $\text{g CO}_2\text{eq kg}^{-1}$  DM intake. More research is needed to quantify and evaluate diet effects on enteric  $\text{N}_2\text{O}$  production.

## Quantifying Feedyard $\text{N}_2\text{O}$ Emissions

Emissions of  $\text{N}_2\text{O}$  from open-lot cattle systems are not continuous and are highly variable (Table 1). Heterogeneity in measured  $\text{N}_2\text{O}$  fluxes is due to interactions among the chemical, physical, and biological properties that control nitrification and denitrification. In addition, the specific method used to measure and calculate  $\text{N}_2\text{O}$  flux can lead to variability in reported per capita  $\text{N}_2\text{O}$  emissions for cattle, which complicates comparison among published studies. Calculated daily per capita  $\text{N}_2\text{O}$  emissions measured with different methods, emission factors (EFs), and model predictions are presented in Fig. 3, where the differences in  $\text{N}_2\text{O}$  estimates among studies and methods used are clear. Data from Intergovernmental Panel on Climate Change (IPCC) EFs were converted to daily  $\text{N}_2\text{O}$  production based on average N excretion of cattle fed crude protein (CP) at the recommended level of 12.5 to 13.5% dry matter intake (NRC, 2000) or fed higher CP diets containing wet distillers grain plus solubles (WDGS). Data presented on an area basis (Aguilar et al., 2014; Boadi et al., 2004; Redding et al., 2015) were assumed to house finishing cattle at a density of 15  $\text{m}^2$  per animal.

### $\text{N}_2\text{O}$ Measurement Methods

Most frequently,  $\text{N}_2\text{O}$  from livestock operations is measured with nonflow through/nonsteady state (NFT-NSS) chambers where  $\text{N}_2\text{O}$  concentrations are measured with a gas chromatograph (GC) equipped with an electron capture detector (Cortus et al., 2015; Costa et al., 2014). An example of NFT-NSS chambers installed in a feedyard pen is presented in Fig. 4a. The NFT-NSS chambers were used to measure  $\text{N}_2\text{O}$  production from simulated manure packs by Ayadi et al. (2015), where the effects of temperature were demonstrated after application of fresh feces and urine (Fig. 5).

Using the NFT-NSS method, a chamber (generally polyvinyl chloride or some other inert material) with a known volume is inserted into the manure and capped for short periods (generally 30 min) to collect emitted compounds. Unlike active chambers and wind tunnels where sweep air is passed through the chamber at a known rate, no sweep air is passed through NFT-NSS; rather,  $\text{N}_2\text{O}$  diffuses from the manure into the chamber headspace. Headspace  $\text{N}_2\text{O}$  concentration increases over time, and air samples are withdrawn periodically and analyzed with a GC–electron capture detector. Flux rates are calculated from the change in  $\text{N}_2\text{O}$  concentration over time, usually according to the USDA Agricultural Research Service GRACEnet (Greenhouse gas Reduction through Agricultural Carbon Enhancement) protocol developed for soils (Hutchinson and Mosier, 1981; Parkin and Venterea, 2010), although the specific calculation method may vary.

Typical issues with NFT-NSS chambers are lateral diffusion and leakage when the surface is irregular (e.g., surface roughness), high spatial variation where coefficients of variation commonly exceed 100% (Parkin and Venterea, 2010), difficulty inserting chambers to optimal depths (Hutchinson and Livingston, 2001, 2002), and creation of micro-environments within the chamber (e.g., temperature, humidity and pressure perturbations, gas mixing, and manure disturbance). Chamber flux measurements are usually determined on hourly to daily scales because of labor-intensive sample collection and the need for numerous chambers to account for spatial heterogeneity. Thus, short-term  $\text{N}_2\text{O}$  fluxes, such as from urine spots, may not be captured with this

**Table 1. Selected studies of N<sub>2</sub>O emissions from open-lot cattle production systems. All data are presented in units reported in original literature. A graphical comparison of calculated annual per capita N<sub>2</sub>O emissions is presented in Fig. 3.**

Reference	Cattle production system	N <sub>2</sub> O emission rate	Technique
IPCC (2006)	emission factor developed for dry lot and bedded cattle	0.02 kg N <sub>2</sub> O–N kg <sup>-1</sup> N excreted (dry lot); 0.01 kg N <sub>2</sub> O–N kg <sup>-1</sup> N excreted (deep bedded manure, no mixing); 0.07 kg N <sub>2</sub> O–N kg <sup>-1</sup> N excreted (deep bedded manure, active mixing)	average values derived from literature review
Pchetteplace et al. (2001)	simulated US feedyard	4.3 kg N <sub>2</sub> O animal <sup>-1</sup> yr <sup>-1</sup>	modeled with IPCC method
Boadi et al. (2004)	Manitoba, Canada research feedyard	0.06 kg N <sub>2</sub> O animal <sup>-1</sup> yr <sup>-1</sup>	vented small chambers (20.3 cm diameter, 0.03 m <sup>2</sup> footprint)
Borhan et al. (2011)	Texas panhandle feedyard	0.21 kg N <sub>2</sub> O animal <sup>-1</sup> yr <sup>-1</sup> ; 3.4 g animal d <sup>-1</sup> (summer)	dynamic chamber (0.19 m <sup>2</sup> footprint) and portable gas chromatograph
Leytem et al. (2011)	southern Idaho open-lot dairy	3.7 kg N <sub>2</sub> O animal <sup>-1</sup> yr <sup>-1</sup> (0.2–33 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup> )	photoacoustic field gas monitor and inverse dispersion method
Redding et al. (2015)	beef feedyards in Northern and Southern Australia	0.35 kg N <sub>2</sub> O animal <sup>-1</sup> yr <sup>-1</sup> (Southern Australia); 0.002 kg N <sub>2</sub> O animal <sup>-1</sup> yr <sup>-1</sup> (Northern Australia)	large vented nonflow-through chamber (20 m <sup>2</sup> footprint)
Rahman et al. (2013)	North Dakota feedyard	0.03 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup>	portable wind tunnel (0.80 × 0.40 m) with airflow rate of 1.75 m <sup>3</sup> s <sup>-1</sup>
Costa et al. (2014)	Brazilian feedyard	0.02 g ± 27.1 g N <sub>2</sub> O–N animal <sup>-1</sup> d <sup>-1</sup>	static chambers
Bonfacio et al. (2015)	simulated Idaho open-lot dairy	14 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup>	modeled with the Integrated Farm Systems Model
Cortus et al. (2015)	monoslope beef barns	0.89–1.0 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup>	gas filter correlation analyzer
Stackhouse-Lawson et al. (2013)	California research feedyard	0.24–0.29 g N <sub>2</sub> O animal <sup>-1</sup> h <sup>-1</sup> (~6.4 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup> )	enclosed 185 m <sup>2</sup> cattle pens vented to a photoacoustic field gas monitor.
Zhu et al. (2014)	milking cows and heifers on an open-lot dairy in China	1.5–2.8 g animal <sup>-1</sup> h <sup>-1</sup> (~ 37 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup> ; milking cows); 0.3–2.2 g animal <sup>-1</sup> h <sup>-1</sup> (~ 24 g N <sub>2</sub> O animal <sup>-1</sup> d <sup>-1</sup> ; heifers).	gas samples collected from nine inlet ports along 80 m length of tubing into a mixing chamber and analyzed with gas chromatography.
Boadi et al. (2004)	Manitoba, Canada feedyard	0.134 mg N <sub>2</sub> O m <sup>-2</sup> h <sup>-1</sup>	vented small static chambers (20.3 cm diameter, 0.03 m <sup>2</sup> footprint)
Aguilar et al. (2014)	Kansas feedyard under differing pen conditions	0.0–41.4 mg N <sub>2</sub> O m <sup>-2</sup> h <sup>-1</sup> (median values: moist/muddy, 2.0; dry/loose, 0.16; dry/compacted, 0.13; flooded, 0.10 mg N <sub>2</sub> O m <sup>-2</sup> h <sup>-1</sup> )	static chambers
Redding et al. (2015)	beef feedyards in Northern and Southern Australia	0.004 kg N <sub>2</sub> O ha <sup>-1</sup> d <sup>-1</sup> (Southern Australia); 0.43 kg N <sub>2</sub> O ha <sup>-1</sup> d <sup>-1</sup> (Northern Australia)	large vented nonflow-through chamber (20 m <sup>2</sup> footprint)

method. However, NFT-NSS chambers are suitable for evaluating mitigation methods and determining the effects of changing manure physicochemical characteristics and environmental conditions on N<sub>2</sub>O fluxes (Hensen et al., 2013).

Selection of an appropriate mathematical method to relate measured N<sub>2</sub>O concentrations to flux is essential for NFT-NSS data. Different methods can produce quite disparate results, creating considerable uncertainty in N<sub>2</sub>O flux estimates (Levy et al., 2011; Venterea, 2013). Some calculation methods include linear or quadratic regression (Hutchinson and Mosier, 1981), the non-linear diffusive flux estimator method (Livingston et al., 2006), the chamber bias correction method (Venterea, 2010), and the revised Hutchinson and Mosier method (Pedersen et al., 2010). Parkin and Venterea (2010) developed a comprehensive manual for using NFT-NSS chambers for soil emissions, but it is unclear to what degree these guidelines are used for CAFO N<sub>2</sub>O studies. Standardized protocols with NFT-NSS chambers are needed that include strategies for typical complications in feedyards (e.g., difficulty inserting chambers into the dense manure pack, surface irregularity, spatial variability, and inability to leave chambers in pens occupied by cattle because of potential damage to equipment by animals or animal injury). Many of these same concerns have been an issue for quantification of NH<sub>3</sub> and other emissions from cattle pens with chambers and wind tunnels (Cole et al., 2007; Parker et al., 2009, 2013).

Micrometeorological methods (MMs) rely on measured atmospheric gas concentrations and a description of atmospheric processes (e.g., wind speed and direction, air temperature,

turbulence) near the emitting surface. Advantages of MMs include noninterference with emission processes, integration over large areas (e.g., m<sup>2</sup> or ha), and measurements on time scales <1 h. Some disadvantages of MMs include the requisite specialized equipment, technical expertise, and relatively large source areas. Calculation of trace gas emission rates using MMs may be complicated when the emission source is mobile, such as grazing or feedyard cattle. Feedyard N<sub>2</sub>O concentrations are often barely detectable above background; thus, current MMs may not capture small and episodic N<sub>2</sub>O fluxes from urine spots or areas with high DEA or NA.

Open-path Fourier-transform infrared spectroscopy, tunable diode laser spectroscopy, eddy covariance, and other MMs have been used to measure NH<sub>3</sub>, N<sub>2</sub>O, and CH<sub>4</sub> emissions from open-lot cattle systems (Baum, 2008; Baum and Ham, 2009; Leytem et al., 2011, 2013; Todd et al., 2011). Singurindy et al. (2007) used eddy covariance to determine N<sub>2</sub>O emission from soil after dairy manure application. Eddy covariance systems, such as pictured in Fig. 4b, have potential as a sensitive method to quantify N<sub>2</sub>O fluxes from open-lot systems but have yet to be deployed in feedyards.

Using a novel approach, Zhu et al. (2014) collected air samples through nine inlet ports along an 80-m length of polyethylene tubing at an open-lot dairy in China. Line-averaged N<sub>2</sub>O concentrations were determined with a GC, and flux rates were calculated with the backward Lagrangian stochastic inverse dispersion technique (Flesch et al., 2009).

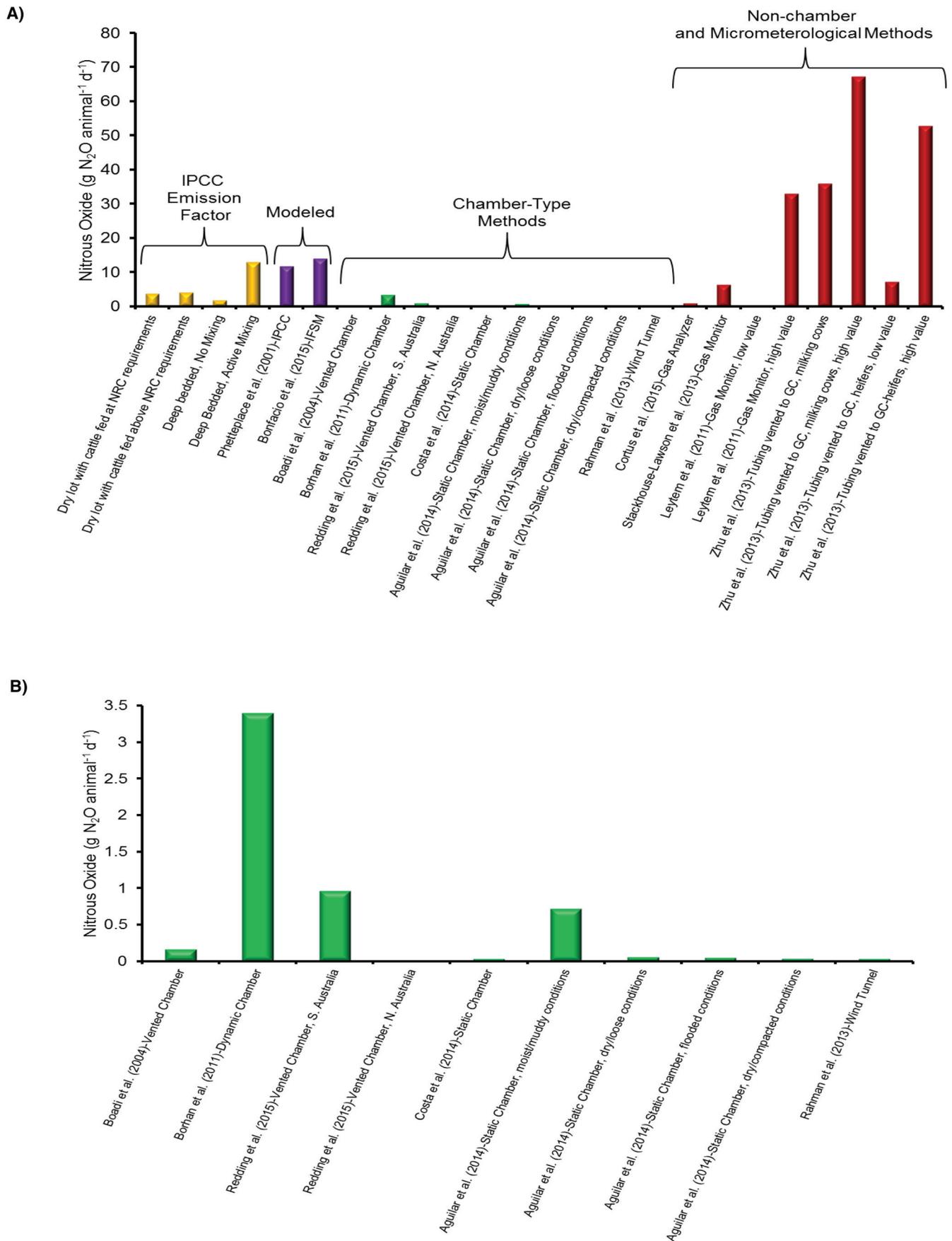
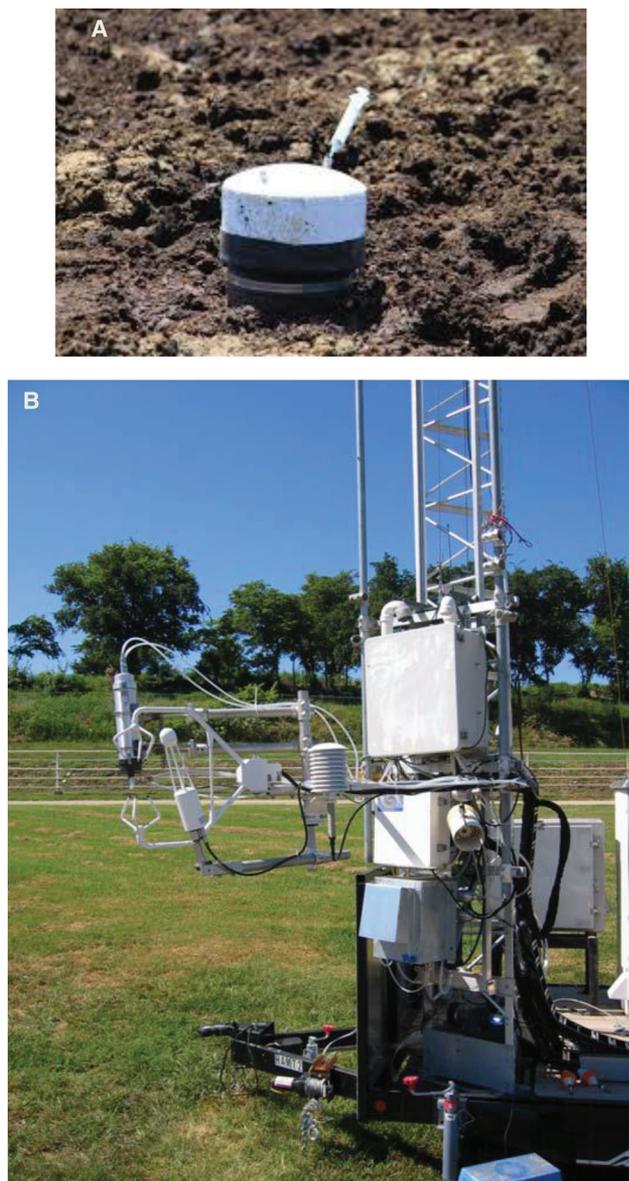


Fig. 3. Comparison of Intergovernmental Panel on Climate Change (IPCC) emission factors, model predictions, and measurement methods on annual per capita N<sub>2</sub>O emission from open-lot cattle production systems. All data were converted from the original units presented in Table 1 to kg N<sub>2</sub>O animal<sup>-1</sup> yr<sup>-1</sup>. GC, gas chromatography; NRC, National Research Council.



**Fig. 4.** Examples of available methods to measure  $N_2O$  and ammonia in livestock production. (A) Sealed nonflow-through/nonsteady state chamber deployed in a feedyard pen. (B) Relaxed eddy covariance system used by Baum (2008).

Emission rates for milking cows ranged from 1.5 to 2.8  $g N_2O animal^{-1} h^{-1}$ , whereas emission from heifers were 0.3 to 2.2  $g N_2O animal^{-1} h^{-1}$  (Fig. 6) (Zhu et al., 2014). Differences between  $N_2O$  fluxes in milking cows and heifers were attributed to manure management, with weekly manure removal from milking cow housing versus twice monthly removal for heifers. Another factor could be higher urea-N excretion by milking cows than heifers due to greater feed demand and N intake by lactating animals (Cassel et al., 2005). Higher temperatures were recorded in milking cow areas ( $\sim 27^\circ C$ ) than that for heifers ( $\sim 12^\circ C$ ) by Zhu et al. (2014), which could lead to greater  $N_2O$  production and emission.

### Scale of Feedyard $N_2O$ Emissions

Limited research has been conducted to directly measure  $N_2O$  emission from cattle feedyards; thus, the impact of  $N_2O$  from open-lot cattle systems on global climate change remains

unclear. Some work suggests that open-lot systems may not be a significant  $N_2O$  source. Stackhouse-Lawson et al. (2013) used a photoacoustic infrared field gas monitor to measure  $N_2O$  emission from steers housed in four completely enclosed pens (40 steers per each 185- $m^2$  soil-surfaced pen). Per capita  $N_2O$  emissions were  $\sim 0.24 g N_2O animal^{-1} h^{-1}$ , or  $6 g animal^{-1} d^{-1}$  (Table 1). Using similar instrumentation, Leytem et al. (2011) found that annualized emissions from a 10,000 cow open-lot dairy in Idaho were  $10 g animal^{-1} d^{-1}$  in a study conducted from March to November 2008. The highest emissions ( $\sim 36 g N_2O animal^{-1} d^{-1}$ ) were observed in spring (April to late May), when the manure in pens was drying and air temperatures were increasing (average temperature,  $21^\circ C$ ).

Other studies (Rahman et al., 2013; Zhu et al., 2014) used different methods (GC coupled with inversion modeling and a custom wind tunnel, respectively) and found emission rates ranging from 17 to 38  $g animal^{-1} d^{-1}$  from a North Dakota feedyard and an open-lot dairy in China. Costa et al. (2014) reported much lower  $N_2O-N$  emissions from a Brazilian beef cattle feedyard that averaged  $0.02 g animal^{-1} d^{-1}$ ; however, spatial variations ranged between 55 and 123%. In this study, the highest  $N_2O$  fluxes occurred from recent urine spots. Waldrip et al. (2015c) conducted eight studies with NFT-NSS chambers in two Texas feedyards and found low, but highly variable,  $N_2O$  flux rates that ranged from below detectable levels to  $3.1 g animal^{-1} d^{-1}$ .

We calculated daily per capita  $N_2O$  emissions from published data obtained from constant EF, model simulations, chambers, and MMs, which averaged  $18 \pm 10 g N_2O animal^{-1} d^{-1}$  (range,  $0.04-67 g N_2O animal^{-1} d^{-1}$ ) (Fig. 3). This variation was due to inconsistency in the measurement technique used and spatial and temporal irregularity in  $N_2O$  production and release attributable to management, animal diet, and environmental conditions. Calculated  $N_2O$  fluxes in studies that used chamber-type methods were very low ( $0.15 \pm 0.28 g N_2O animal^{-1} d^{-1}$ ) (Fig. 3a and 3b) compared with those that used MMs ( $25.4 \pm 25.6 g N_2O animal^{-1} d^{-1}$ ) (Fig. 3a). Further study and comparison of differing methodologies on measured and calculated  $N_2O$  flux rates from feedyards are needed.

Diel differences in  $N_2O$  emissions have been reported from open-lot dairies during summer and fall in eastern China (Zhu et al., 2014), where fluxes tended to be highest at mid-day and lowest in early morning (Fig. 6). This pattern was likely due to increased manure temperature and consequent enhanced  $N_2O$  solubility (Blackmer et al., 1982). However, increased  $N_2O$  emission at higher temperatures was not observed by Leytem et al. (2011) on an Idaho open-lot dairy, from feces and urine in controlled temperature incubators (Pereira et al., 2012), or on a Kansas feedyard (Aguilar et al., 2014). Waldrip et al. (2015c) reported that feedyard  $N_2O$  collected from NFT-NSS chambers was positively related to manure temperature. However, relationships between  $N_2O$  fluxes and numerous other variables were identified, including manure  $H_2O$ , OM,  $NO_3^-$ ,  $NH_4^+$ , and dissolved N and C concentrations as well as spectral characteristics related to OM humification. Thorough manure characterization could improve understanding of the driving forces for feedyard  $N_2O$  production and emission.

Rahman et al. (2013) used a portable wind tunnel at a North Dakota feedyard, where atmospheric  $N_2O$  concentrations were highest in spring (March;  $0.99 ppm N_2O$ ) and lowest in early

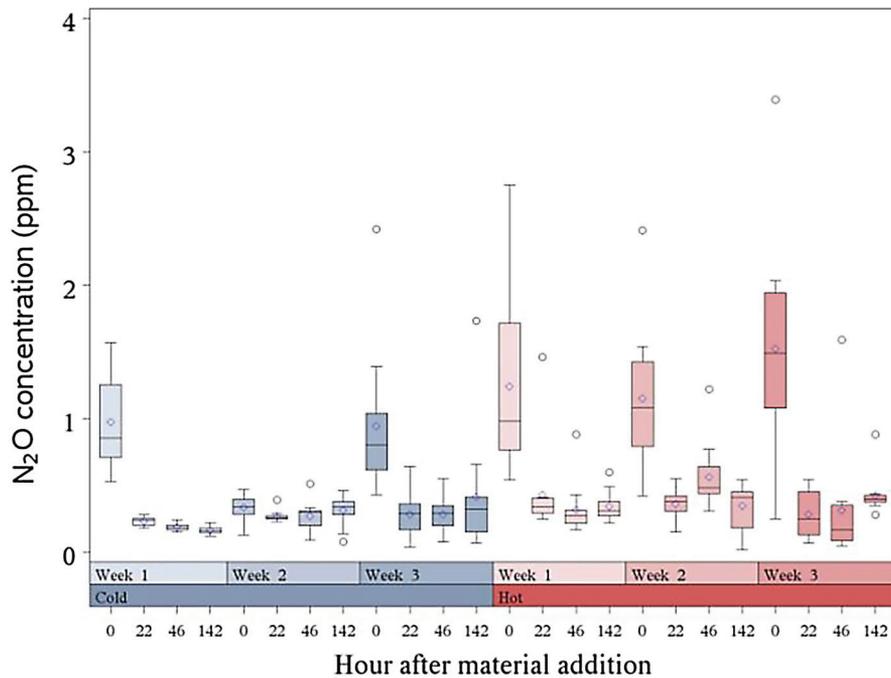


Fig. 5. Concentrations of  $N_2O$  in headspace of nonflow-through/nonsteady-state chambers inserted into simulated beef manure packs. Feces, urine, and bedding were incubated for three consecutive 142-h periods under cold ( $10^\circ C$ ) or hot ( $40^\circ C$ ) temperatures. Manure moisture content was maintained at 70% throughout the experiment. This work showed that most  $N_2O$  was produced immediately after addition of fresh material and that temperature played a major role in the magnitude of  $N_2O$  production. Source: Ayadi et al. (2015).

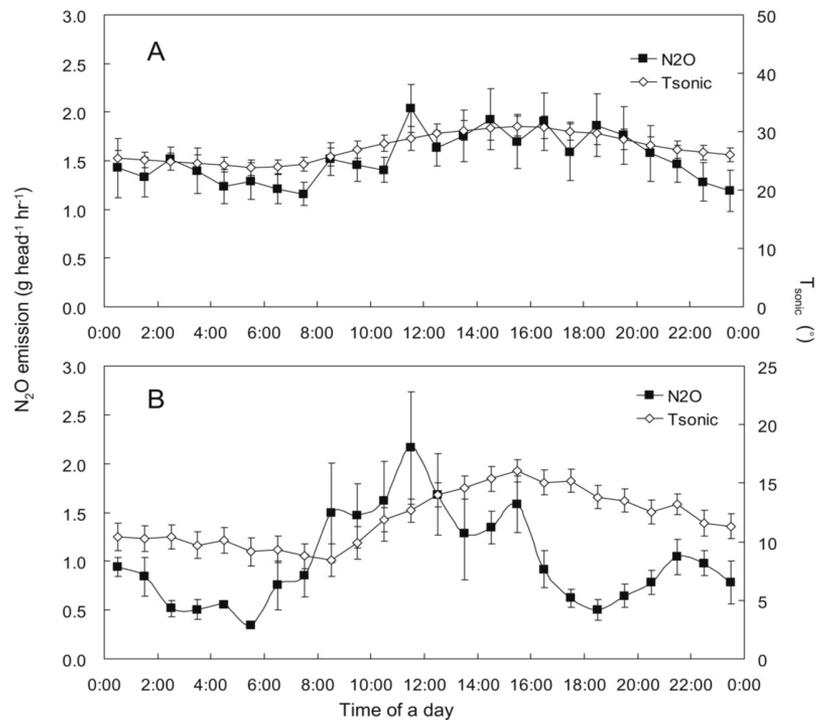


Fig. 6. Diurnal patterns of  $N_2O$  emissions from (A) milking cows and (B) heifers on an open-lot dairy in China. Temperature ( $T_{sonic}$ ;  $^\circ C$ ) was more closely related to  $N_2O$  fluxes for milking cows than heifers. Source: Zhu et al. (2014).

summer (June) and winter (January;  $\sim 0.45$  ppm  $N_2O$ ). Manure  $H_2O$  content was proposed as the likely cause of this observation because drier conditions facilitated nitrification in spring. However, it should be noted that estimated fluxes were  $\sim 46$  times higher than those measured from dynamic chambers on a Texas feedyard by Borhan et al. (2011). Rahman et al. (2013) attributed this contrast between studies to the specific measurement

method used and questioned the suitability of flux chambers for feedyard  $N_2O$  measurements.

Most inventory calculations and models for estimating  $N_2O$  emission include information regarding the amount of manure N excreted; however, Redding et al. (2015) and Waldrip et al. (2015c) found no relationship between  $N_2O$  emissions and  $NO_3^-$  or total N concentrations in feedyard manure. Redding et al. (2015) measured

$N_2O$  emissions from Australian feedyards using a large chamber and a Fourier-transform infrared monitor. Fluxes of  $N_2O$  increased with manure density, pH, temperature, and manure mass and decreased with manure  $H_2O$  and organic C content. In short, the interplay among factors involved in  $N_2O$  production and emission processes considerably complicates quantitative analyses of the role of feedyards in global climate change. Continued research is needed at the feedyard scale and in isolated micro-sites (i.e., small chamber and laboratory incubation studies) to supplement current understanding of feedyard  $N_2O$ .

## Estimating Feedyard $N_2O$ Emissions

Estimates of  $N_2O$  emission from CAFOs are required for regional and national GHG assessments and may soon be required for regulatory reporting purposes. Improved emission estimates will assist in evaluation of mitigation practices for feedyards and will help to forecast changes in  $N_2O$  emission due to changing climate and management. The USEPA and IPCC use constant EF to estimate  $N_2O$  production (0.02 kg  $N_2O-N$  kg<sup>-1</sup> manure N excreted for open-lot systems) (IPCC, 2006). Deep-bedded cattle housing (e.g., mono-slope barns) were expected to produce half the  $N_2O$  emitted from open-lot pens, with an EF of 0.01 kg  $N_2O-N$  kg<sup>-1</sup> N excreted. However, there were large variations in the data used to develop these EFs; thus, uncertainty is high (−16% to +24%) (USEPA, 2013). One key limitation of EFs is a failure to account for changes in microbial and chemical processes due to temperature,  $H_2O$ , and other factors affecting  $N_2O$  production and emission.

Process-based models provide an alternative that better represent site-specific conditions and interactions among factors influencing emissions. Several models are available to estimate soil  $N_2O$  emissions. They include, but are not limited to, the Daily Century Model (DayCent) (Del Grosso et al., 2006), the Denitrification and Decomposition Model (DNDC) (Li, 2007; Li et al., 1992), and the Integrated Farm Systems Model (IFSM) (Chianese et al., 2009; Rotz et al., 2012). These models simulate soil processes to predict  $N_2O$  release based on environmental conditions (e.g., temperature, precipitation), management (e.g., fertilization regime, land use type, crop growth), and soil properties (e.g., pH, nutrient concentrations, bulk density, porosity).

Despite progress in modeling soil  $N_2O$  emissions, considerably less work has been conducted to develop and improve process-based models for predicting  $N_2O$  emissions from open-lot CAFOs. Li et al. (2012) added a manure component to DNDC, calling the new model Manure-DNDC. In Manure-DNDC, denitrification is based on biochemical reaction kinetics as a function of denitrifier growth rate, mass balance between  $N_2O$  production and consumption, and gas diffusion through manure. Nitrification rate is predicted according to Michaelis–Menten kinetics, where nitrifier activity is a function of dissolved organic C and  $NH_4^+$  concentrations, temperature, redox potential, and pH in manure. Enteric  $N_2O$  is modeled as a linear function of daily N consumed in feed, diet composition, and animal growth and milk production.

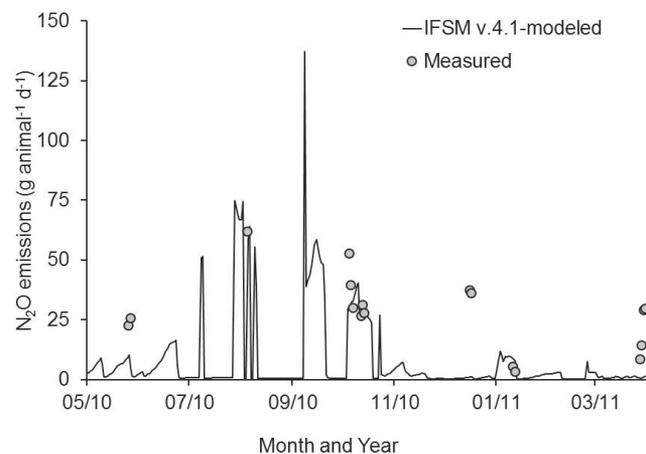
Costa et al. (2014) evaluated Manure-DNDC predictions against measured data from a Brazilian feedyard. Manure-DNDC predictions were within the standard deviations of measured  $N_2O$  emissions, and model predictions mirrored measured

changes in  $N_2O$  fluxes due to precipitation and temperature. However, Manure-DNDC tended to underestimate when measured  $N_2O$  emissions were high, perhaps due to overestimation of  $NO_3^-$  leaching through manure. Li et al. (2012) evaluated Manure-DNDC predictions against 2 yr of measured data where cattle slurries were applied to soil in Scotland. Manure-DNDC effectively simulated peak  $N_2O$  fluxes after slurry application and differences in  $N_2O$  emissions due to C to N ratios of the applied slurries.

The IFSM is another process-based model that has been evaluated for open-lot systems (Rotz et al., 2012). Nitrification and denitrification are predicted with IFSM using relationships from the DayCent soil model (Del Grosso et al., 2006). Factors considered in estimates of  $N_2O$  emissions include manure  $NH_4^+$ ,  $NO_3^-$ , and  $H_2O$  concentrations; porosity and pH; and air temperature. The IFSM has been used successfully to evaluate potential environmental impacts of different dairy management systems, where  $N_2O$  emission estimates varied with climate and agronomic practices (Belflower et al., 2012; Bonfacio et al., 2015; Stackhouse-Lawson et al., 2013).

Bonfacio et al. (2015) predicted mean daily per capita  $N_2O$  emission of 14 g animal<sup>-1</sup> at an Idaho open-lot dairy, as compared with mean measured emission of 17 g  $N_2O$  animal<sup>-1</sup> by Leytem et al. (2011) (Table 1; Fig. 7). Overall, Bonfacio et al. (2015) saw 80% agreement between IFSM predictions and measured emissions. The IFSM accurately simulated changes in denitrification rates due to manure  $H_2O$  content. However, IFSM evaluation and improvement has been limited by low availability of long-term  $N_2O$  emission data from open-lot systems.

Collaboration between model developers and researchers measuring emissions, both at the feedyard scale and under controlled laboratory conditions, will assist in improving the parameterization and evaluation of models for estimation of feedyard  $N_2O$  emissions. Valid process-based models will be useful tools to quantify  $N_2O$  fluxes from open-lot cattle systems, allowing feedyard operators to make management decisions that minimize  $N_2O$  volatilization from manure.



**Fig. 7.** Measured and Integrated Farm System Model (IFSM)-predicted  $N_2O$  emissions for an Idaho dairy from May 2010 to March 2011. Average measured emission rates were 17 g  $N_2O$  animal<sup>-1</sup> d<sup>-1</sup> and IFSM predicted rates were 14 g  $N_2O$  animal<sup>-1</sup> d<sup>-1</sup>, with RMSE of 13 g  $N_2O$  animal<sup>-1</sup> d<sup>-1</sup> and index of agreement of 80%. Source: Bonfacio et al. (2015).

# Mitigation of Feedyard N<sub>2</sub>O Emissions

## Increasing Cattle Productivity and Reducing Dietary Protein Intake

Feedyard N<sub>2</sub>O emission is related to animal productivity, where higher-producing cattle consume more feed and excrete greater quantities of manure N than lower-producing animals. However, as meat and milk production increases, N<sub>2</sub>O losses per unit of marketable product are reduced. There is less cumulative N<sub>2</sub>O production from high-producing cattle than from lower-producing cattle when expressed as emission intensity (N<sub>2</sub>O emitted per unit of animal product). Currently, increasing animal productivity is the most viable means of reducing feedyard N<sub>2</sub>O emission. Cattle productivity can be increased through improvements in genetics, feeding, health, and management. As an example, Hristov et al. (2013) estimated that total GHG emissions were 70% lower from cattle gaining 1.5 versus 1.0 kg body weight d<sup>-1</sup>.

Ionophores, anabolic implants (estrogens, testosterone, and progestins), antibiotics, and β-agonists (nonhormonal compounds that reduce fat metabolism) are often used at conventionally managed US feedyards to improve animal productivity and to prevent disease (Wileman et al., 2009). These treatments reduce beef cattle N<sub>2</sub>O emission intensity by increasing growth rate and feed efficiency, resulting in a shorter time-frame to reach target finishing weight. Anabolic implants increase final body weight while maintaining a similar body composition to unimplanted animals (Guiroy et al., 2002). Ionophores and antibiotics improve feed efficiency and animal health, whereas β-agonists fed during the final phases of finishing improve weight gain efficiency and increase lean tissue gain.

Other than improved animal efficiency, it remains unclear how growth-promoting treatments influence feedyard N<sub>2</sub>O emission. Stackhouse-Lawson et al. (2013) evaluated performance, carcass traits, and N<sub>2</sub>O emissions from steers receiving different growth-promoting agents. Overall, there was a 7 to 20% increase in N<sub>2</sub>O emissions from cattle that received zilpatrol hydrochloride, monensin, and tylosin phosphate compared with unsupplemented cattle; however, N<sub>2</sub>O emission intensity decreased when cattle received multiple growth-promoting compounds. Hristov et al. (2013) suggested that growth-promoting compounds may have little effect on feedyard emissions. However, improved animal health results in less cattle mortality and should decrease the feedyard N<sub>2</sub>O contribution per unit of saleable product.

Montes et al. (2013) reviewed methods for mitigating manure-derived N<sub>2</sub>O emissions and recommended dietary manipulation to improve the balance between nutrient inputs, production, and environmental impact. Diet composition directly affects N<sub>2</sub>O emission by influencing N excretion characteristics and manure microbial composition. The manure from dairy cattle fed diets containing a high proportion of forage tends to contain more fungi and emit less N<sub>2</sub>O than manure from cattle consuming a high protein/low fiber diet (Jost et al., 2013). Hünnerberg et al. (2014) conducted a life-cycle analysis and estimated that inclusion of WDGS at 40% of daily DM intake slightly increased N<sub>2</sub>O emissions from 21 to 22% of total CO<sub>2</sub>e in finishing cattle. Distillers' grains contain approximately 30 to 40% CP, as compared with approximately 10% CP for steam-flaked corn (NRC, 2000). The authors attributed higher predicted N<sub>2</sub>O emissions to excessive N consumption in relation to cattle requirements. The importance

of cattle diet on N excretion was recently reviewed by Waldrip et al. (2015a). Hales et al. (2013) reported that inclusion of 45% WDGS in cattle diets increased total N excretion by 47%. Both total N intake and diet composition are important considerations when assessing the N sustainability of cattle operations.

## Feedyard Pen Management to Mitigate N<sub>2</sub>O Emissions

### Managing Manure H<sub>2</sub>O Content

Nitrous oxide emissions can be reduced when gravimetric H<sub>2</sub>O content of the feedyard pen is maintained within 20 to 40% of manure DM; however, alternative pen designs that evenly distribute manure and H<sub>2</sub>O might be required for this practice to be effective (Miller and Berry, 2005). Aguilar et al. (2014) investigated Kansas feedyard N<sub>2</sub>O fluxes from areas designated as moist/muddy, dry and loose, dry and compacted, and wet/flooded. Moist/muddy areas were found to be "hot spots" for microbial activity, with median emission rates of 2.03 mg N<sub>2</sub>O m<sup>-2</sup> h<sup>-1</sup>, whereas N<sub>2</sub>O fluxes from dry or flooded areas were negligible (Table 1). Flooded conditions can reduce N<sub>2</sub>O emissions due to lowered redox potential (Johnson-Beebout et al., 2009) and restricted gas diffusion through the manure pack.

### Chemical Amendments

Urease inhibitors, nitrification inhibitors, and other chemical amendments have been evaluated for reducing N<sub>2</sub>O emissions from fertilized fields, but research on their use with CAFOs and grazing livestock has been relatively limited, and results have been highly variable (de Klein and Monaghan, 2011; Gerber et al., 2013; Subbarao et al., 2006). *N*-(*n*-butyl) thiophosphoric triamide (NBPT) is a urease inhibitor used on fertilized cropland. Urease inhibitors delay urea hydrolysis to NH<sub>4</sub><sup>+</sup>. This extends the time available for root uptake of urea and reduces the opportunity for nitrification of NH<sub>4</sub><sup>+</sup> to NO<sub>3</sub><sup>-</sup>, where N<sub>2</sub>O may be produced as an intermediate. The probable mechanism of NBPT inhibition is chelation with nickel in the active site of the urease enzyme (Blakeley and Zerner, 1984; Upadhyay, 2012). In cattle pens and manure, NBPT application can reduce NH<sub>3</sub> emissions (Parker et al., 2005, 2012; Shi et al., 2001). However, little is known regarding the effects of NBPT on feedyard N<sub>2</sub>O emissions. It is possible that NBPT use in feedyards could lead to "pollution swapping," where NBPT-treated manure with higher total N content would emit more N<sub>2</sub>O and less NH<sub>3</sub> than manure that did not receive NBPT application. Further research is warranted in this area.

Nitrification inhibitors directly block the nitrification pathway, limiting substrates for denitrification and reducing N<sub>2</sub>O losses from urine spots deposited by cattle (Morales et al., 2015). The most commonly used nitrification inhibitors for agricultural soils are 2-chloro-6-(trichloromethyl) pyridine (nitrapyrin), dicyandiamide (DCD), and 3,4-dimethylpyrazole phosphate (DMPP). It is proposed that DCD specifically inhibits NH<sub>3</sub> oxidation by blocking the conversion of NH<sub>3</sub> to NH<sub>2</sub>OH (Di et al., 2009), which is the first step in nitrification (Fig. 2). Morales et al. (2015) observed that DCD applied at a rate of 10 mg kg<sup>-1</sup> (DM) to urine-amended soils reduced N<sub>2</sub>O emissions by 23 to 67%. No definitive changes with DCD application were observed in the soil community composition of nitrifiers, potential denitrifiers, or organisms involved in alternative N-utilizing pathways. Other researchers have studied the efficacy of DCD and DMPP on mitigating N<sub>2</sub>O emission from urine spots in grazed pasture,

with mixed results (Luo et al., 2013; Smith et al., 2008; Zaman and Nguyen, 2012). Mazzetto et al. (2015) investigated DCD for mitigation of  $N_2O$  from cattle urine on an acidic Brazilian Oxisol (pH 5.0) and found no benefit from DCD application.

Very little research has evaluated DCD or other nitrification inhibitors on  $N_2O$  from open-lot feedyards. The feedyard manure microbial community composition, constant excretion of urine and feces onto the pen floor, and high manure OM content may challenge nitrification inhibitors used successfully on fertilized cropland. Sahrawat et al. (1987) reported that both DCD and nitrapyrin slowed nitrification rates in a silt loam but were ineffective in an organic Histosol. Calderon et al. (2005) reported that nitrapyrin reduced nitrification but had inconsistent effects on  $N_2O$  emission from manured soils. Nitrapyrin delayed nitrification in anhydrous  $NH_3$ -fertilized corn plots and temporarily reduced  $N_2O$  emissions in a study by Parkin and Hatfield (2010); however, cumulative annual  $N_2O$  losses were not significantly different from plots that did not receive nitrapyrin. Research is needed to evaluate long-term cost effectiveness, optimal application rate, and required application frequency of chemical inhibitors as well as their potential toxicity to cattle and feedyard personnel and overall practicality for mitigation of feedyard  $N_2O$  emissions.

#### Organic and Other Amendments

The addition of biochar or other organic C sources to feedyard pens may reduce  $N_2O$  emissions (Aguilar et al., 2014). One proposed mechanism of  $N_2O$  inhibition after organic residue application is via sorption and retention of  $N_2O$ ,  $NH_4^+$ , and/or  $NO_3^-$  (Aguilar et al., 2014; Yao et al., 2012). However, the effectiveness of organic amendments is influenced by manure  $H_2O$  content, amendment application rate, and environmental temperature (Aguilar et al., 2014). Aguilar et al. (2014) observed 57 to 73% reduction in  $N_2O$  emission when manure biochars or activated C was added to moist (35%  $H_2O$ ) beef cattle manure in a short-term laboratory study. However, the high amendment rate required to reduce GHG losses in this study (1.1–5.1  $kg\ m^{-2}$ ) may not be practical or cost-effective for commercial feedyards. In this study, temperatures above 40°C caused  $N_2O$  desorption from amendments, leading to larger  $N_2O$  fluxes at high temperatures. In short, organic amendments may be ineffective for mitigation of  $N_2O$  emissions during the dry, hot summers typical of Texas and other major beef-finishing regions in the United States.

Amendments used to control emissions of  $NH_3$  and other compounds in CAFOs, including zeolite, lignite, bedding or other organic material, aluminum sulfate, tannins, acidifying compounds, and biochar, have yet to be evaluated for open-lot feedyards. This is an area that requires significant research regarding environmental impact, practicality, cost-effectiveness, and animal and human health effects before making producer-oriented suggestions for their use on commercial feedyards.

## Conclusions and Areas for Further Research

A thorough literature review revealed that published per capita flux rates for open-lot cattle systems were highly variable and ranged from 0.002 to 4.3  $kg\ N_2O\ animal^{-1}\ yr^{-1}$  or 0.02 to 37  $g\ N_2O\ animal^{-1}\ d^{-1}$ . On an area basis, published emissions rates from feedyards ranged from 0 to 41  $mg\ N_2O\ m^{-2}\ h^{-1}$ . Much of this

variation was due to inconsistency in the measurement technique used and to distinct spatial and temporal irregularity in  $N_2O$  production and volatilization due to management practices, animal diet, and environmental conditions. Feedyard  $N_2O$  emissions appear to be largely controlled by environmental temperature and manure N and  $H_2O$  content; however, relationships were also identified between animal diet, pen management, and animal stocking density. During the preparation of this review, five primary knowledge gaps were identified where current understanding is weak and further research is required: (i) accurate measurement of  $N_2O$  emissions from feedyards using appropriate, and when available, standardized methodologies; (ii) improved understanding of the microbiology, chemistry, and physical structure of pen surface layers that lead to  $N_2O$  emissions; (iii) improved understanding of factors that influence feedyard  $N_2O$  emissions, including manure  $H_2O$  content, porosity, density, available N and C contents, environmental temperatures, and use of veterinary pharmaceuticals; (iv) development of cost-effective and practical mitigation strategies to decrease  $N_2O$  emissions from pen surfaces, manure stockpiles, composting windrows, and retention ponds; and (v) improved process-based models that can accurately predict feedyard  $N_2O$  emissions, evaluate mitigation strategies, and forecast future  $N_2O$  emission trends.

Given the impetus toward increasing regulation, feedyard managers, nutritionists, and researchers may play increasingly important roles in feedyard N management. Common feedyard management practices may need modification or refinement to balance production efficiency with environmental concerns. Large-scale micrometeorological measurement campaigns and small-scale chamber studies are needed to assess the overall magnitude of feedyard  $N_2O$  emissions and to determine key factors driving  $N_2O$  production and emission.

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